

Reduction of reaction mechanisms

Time-scale analysis

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Funded by
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جامعة خليفة
Khalifa University

Multiscale analysis in combustion

1938, Zeldovich and Frank-Kamenetskii (large activation energy, flame speeds)

1939, Frank-Kamenetskii (large activation energy, ignition limits)

1963, Friedlander and Keller (Damkohler number asymptotics)

1964, Blythe (activation energy asymptotics, “sudden freezing” in supersonic flow)

1970, Bush and Fendell (activation energy asymptotics, laminar flame speed)

1971, Linan and Williams (activation energy asymptotics, ignition time)

1971, Williams (activation energy asymptotics, laminar flame speed and diffusion flames)

1980s Peters, Williams, Rogg, Pitsch, Mauss, Sheshardi,
Linian, Sivashinsky, Buckmaster, De Goey

Rate-ratio asymptotics

Williams, Physica D 20:21-34, 1986

Buckmaster, Physica D 20:91-108, 1986

Williams, Proc. Combust. Inst., 30:1-19, 2005

Refs



Reduced mechanisms: rate-ratio asymptotics

1985

NUMERICAL AND ASYMPTOTIC ANALYSIS OF SYSTEMATICALLY REDUCED REACTION SCHEMES FOR HYDROCARBON FLAMES

N. Peters

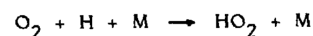
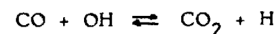
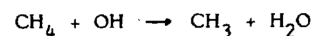
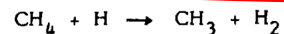
Institut für Allgemeine Mechanik
RWTH Aachen, West-Germany

Abstract

We intend to close the gap between the numerical description of hydrocarbon flames using a large number of elementary reactions and the asymptotic description using very few reaction steps. To proceed in this direction we first reduce the hydrocarbon chemistry to the smallest number of steps that still provides a realistic flame structure. For the example of methane oxidation these are the four reaction steps



Their rates are algebraically complicated expressions that contain kinetic data from 9 elementary reactions. If standard values for these data are used, the flame velocity of a stoichiometric methane-air flame is calculated as 45.6 cm/sec. The reaction rates may be simplified to include only data from the five most important reactions



where the rate expression of the fourth of these reactions is modified. This model is proposed for the asymptotic analysis of the flame structure.

N Peters, *Numerical Simulation of Combustion Phenomena: Proceedings of the Symposium Held at INRIA Sophia-Antipolis, France May 21–24, 1985*. Berlin, Heidelberg: Springer Berlin Heidelberg, 2005.

1. Non-physical reactions
2. Rates are linear combination of elementary reaction rates
3. Accompanied by a number of QSSAs



Reduced mechanisms: rate-ratio asymptotics

1987

COMBUSTION AND FLAME 68: 185–207 (1987)

185

The Asymptotic Structure of Stoichiometric Methane–Air Flames

N. PETERS

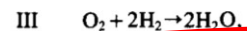
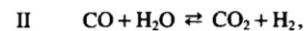
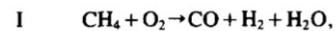
Institut für Allgemeine Mechanik, RWTH Aachen, West-Germany

and

F. A. WILLIAMS

Department of Mechanical and Aerospace Engineering, Princeton University, Princeton, NJ 08544

A C₁-chain mechanism for methane flames is systematically reduced through steady-state and partial-equilibrium assumptions to the three-step mechanism

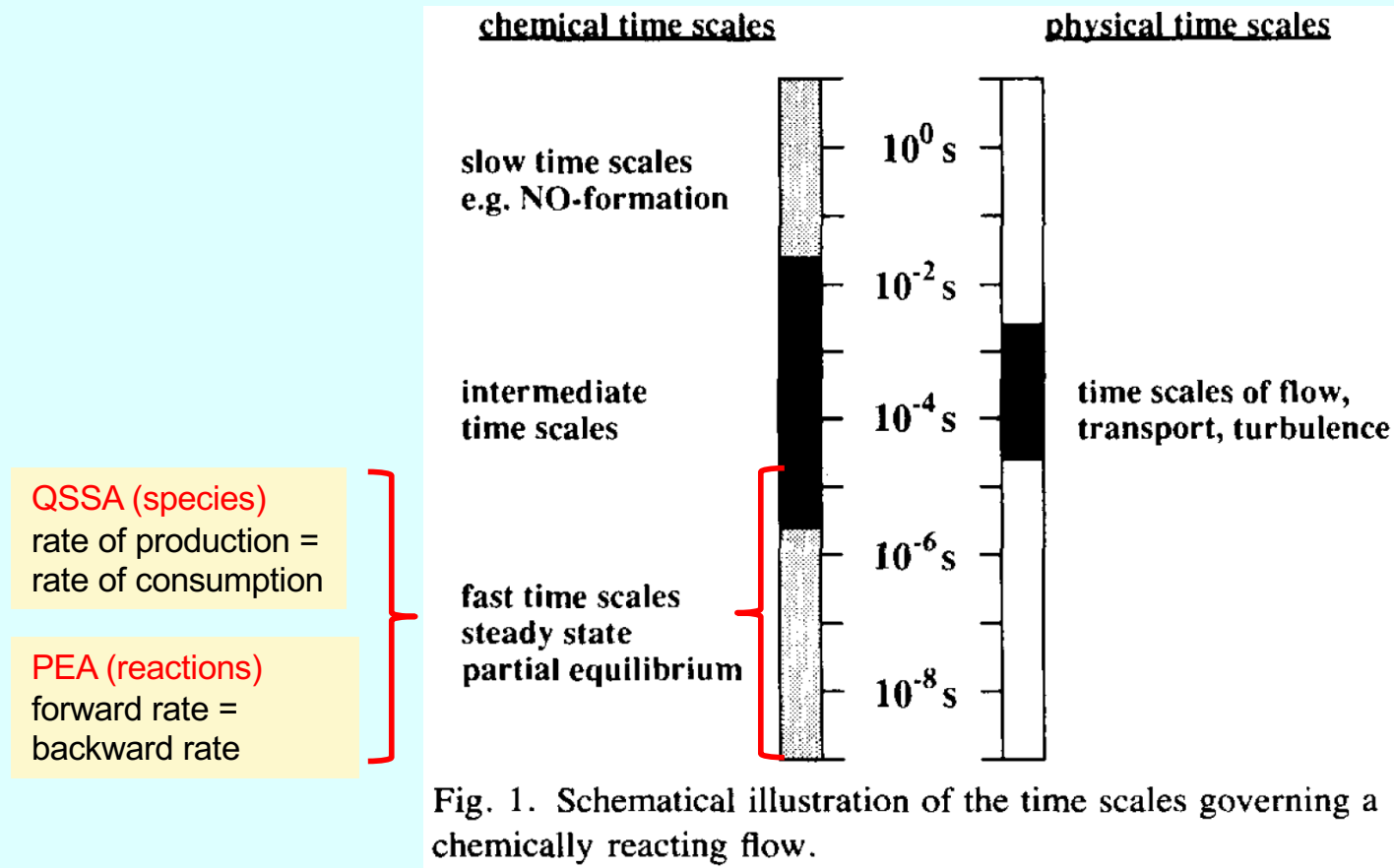


with rates that still contain the kinetic information of the elementary mechanism.

1. Non-physical reactions
2. Rates are linear combination of elementary reaction rates
3. Accompanied by a number of QSSAs



Multiscale character of reacting processes



Maas & Pope, C&F 1992

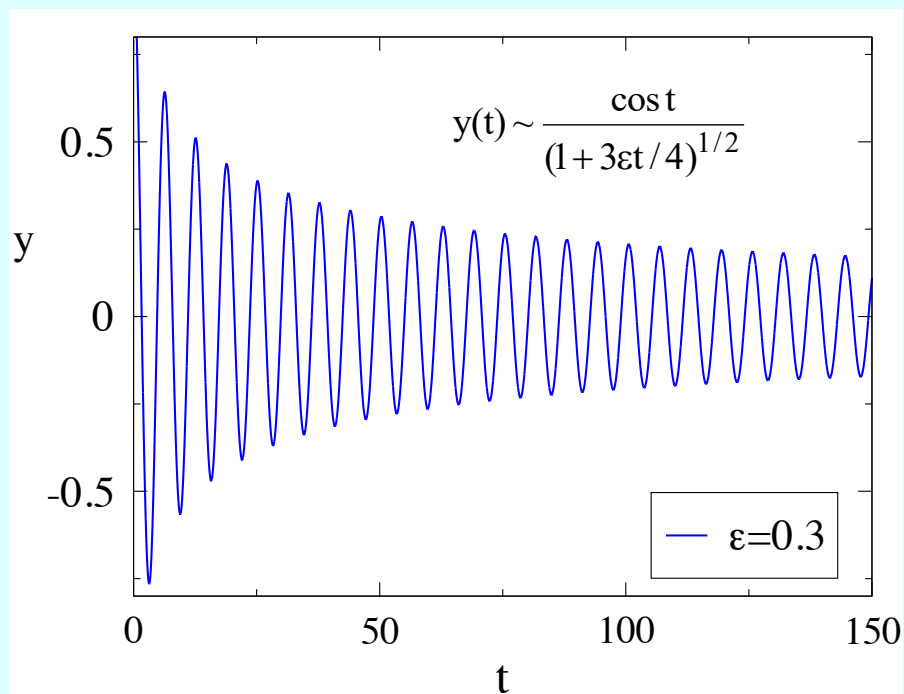
Outline

1. Multiscale (fast/slow) dynamics
2. Time scales
3. Traditional reduction tools and their limitations
4. Alternative view for model reduction
5. Algorithmic model reduction; CSP
6. Various extensions; ILDM, REDIM, ISAT
7. Applications



Fast / Slow systems

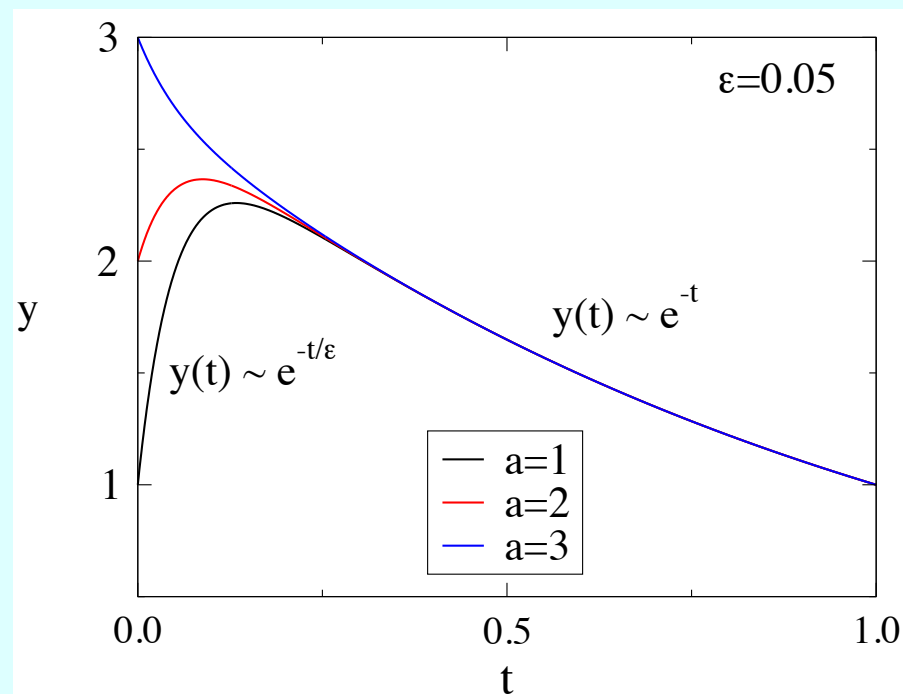
Multiple time scale problem



$$y'' + \epsilon(y')^3 + y = 0$$

$$y(0) = 1 \quad y'(0) = 0$$

Boundary layer problem

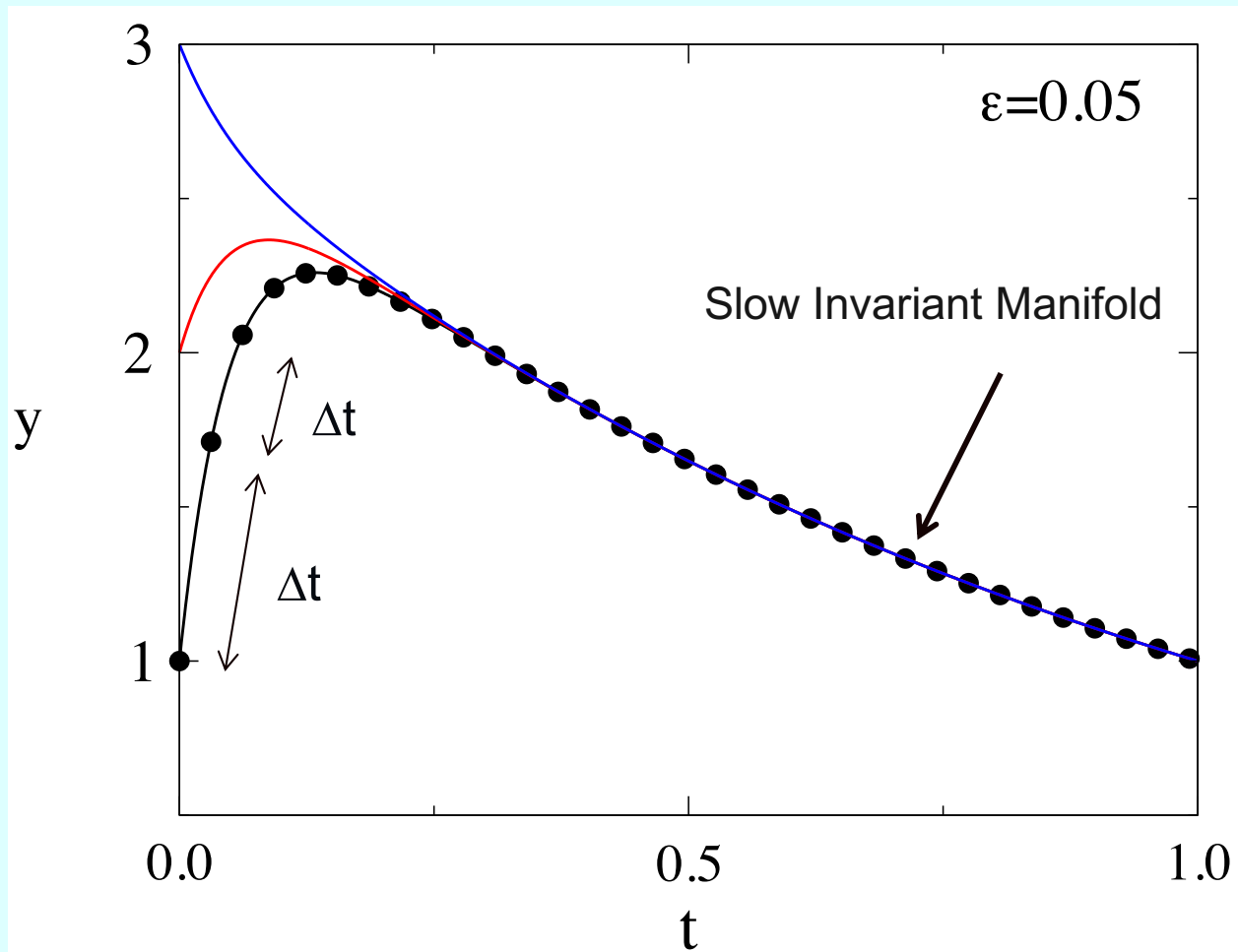


$$\epsilon y'' + (1 + \epsilon)y' + y = 0$$

$$y(0) = a \quad y(1) = 1$$

Bender and Orszag 1978

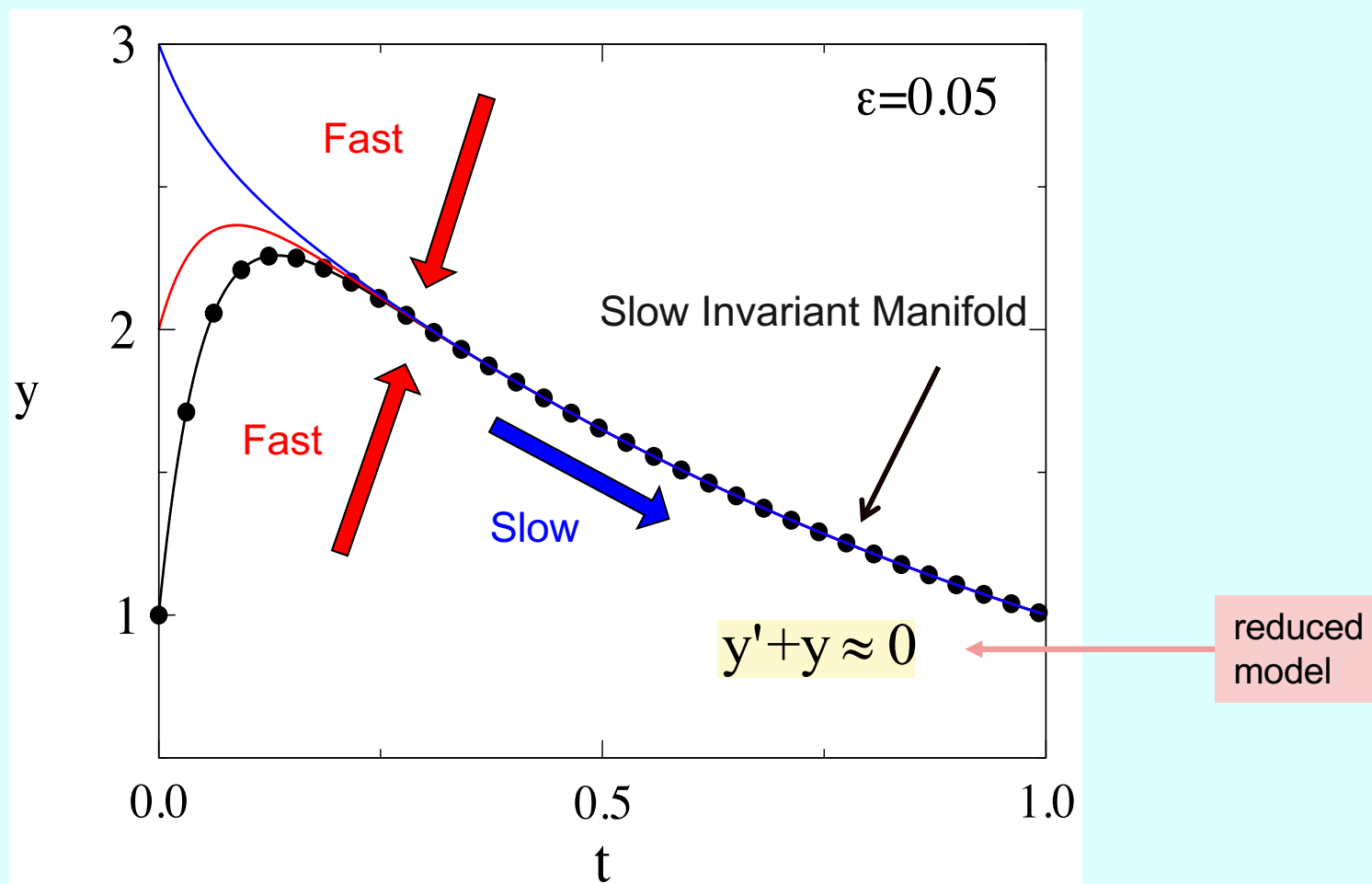
Boundary layer problem



$$\varepsilon y'' + (1 + \varepsilon)y' + y = 0$$
$$y(0) = a \quad y(1) = 1$$



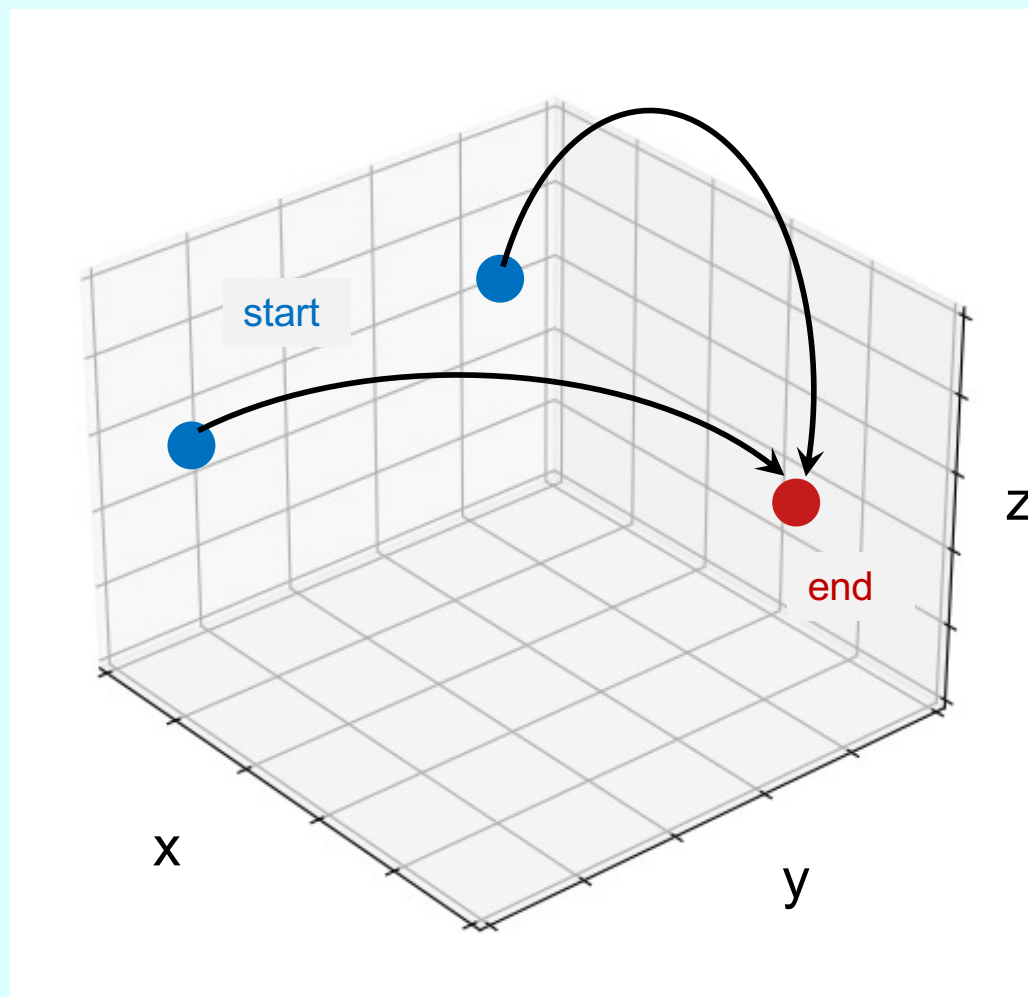
Fast and slow time scales



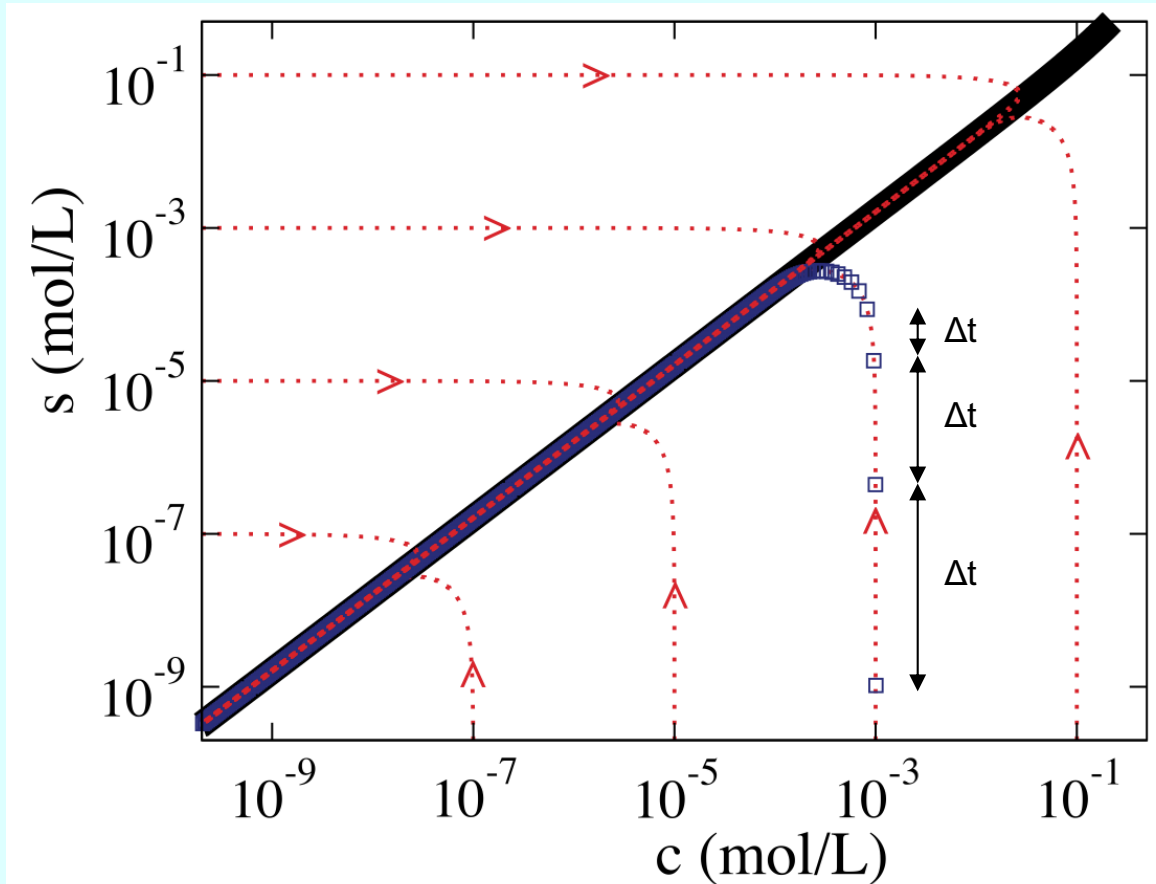
$$\varepsilon y'' + (1 + \varepsilon)y' + y = 0$$

$$y(0) = a \quad y(1) = 1$$

Distinct paths ?

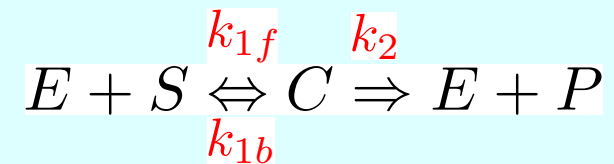


Distinct paths ?



$$k_{1f} = 1 \text{ L}(\text{mol s})^{-1} \quad k_{1b} = k_2 = 1 \text{ s}^{-1} \quad e_o = 1 \text{ mol L}^{-1}$$

Michaelis – Menten model



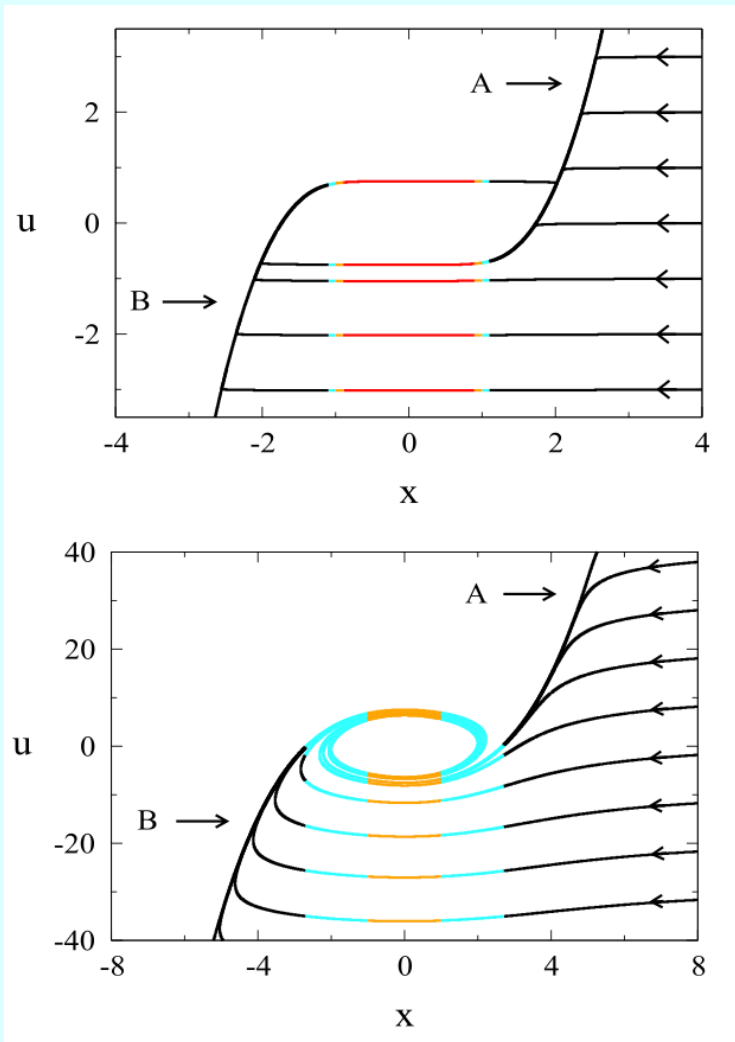
$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} = \mathbf{S}_1 R^1 + \mathbf{S}_2 R^2$$

$$\mathbf{S}_1 = \begin{bmatrix} -1 \\ 1 \end{bmatrix} \quad \mathbf{S}_2 = \begin{bmatrix} 0 \\ -1 \end{bmatrix}$$

$$R^1 = k_{1f}(e_o - c)s - k_{1b}c$$

$$R^2 = k_2 c$$

Distinct paths ?



$$\epsilon = 10^{-2}$$

Van der Pol model

$$\frac{dx}{dt} = \frac{1}{\epsilon}(u - q(x))$$

$$\frac{du}{dt} = -x$$

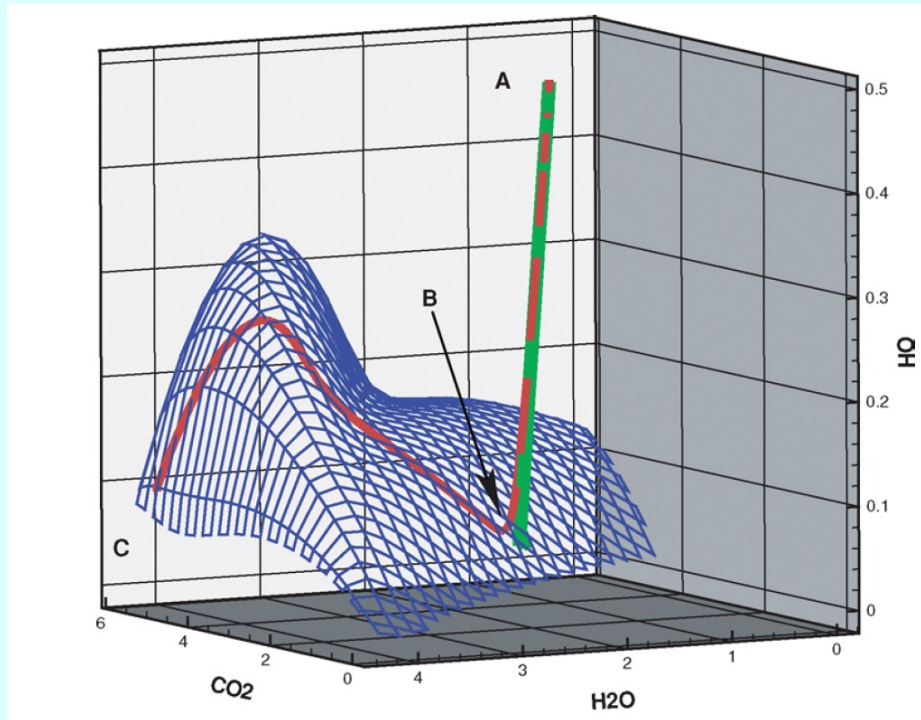
$$q(x) = \frac{x^3}{3} - x$$

$$\epsilon = 10^1$$

B. Van der Pol, *Radio Review* 1 (1920)



Distinct paths ?

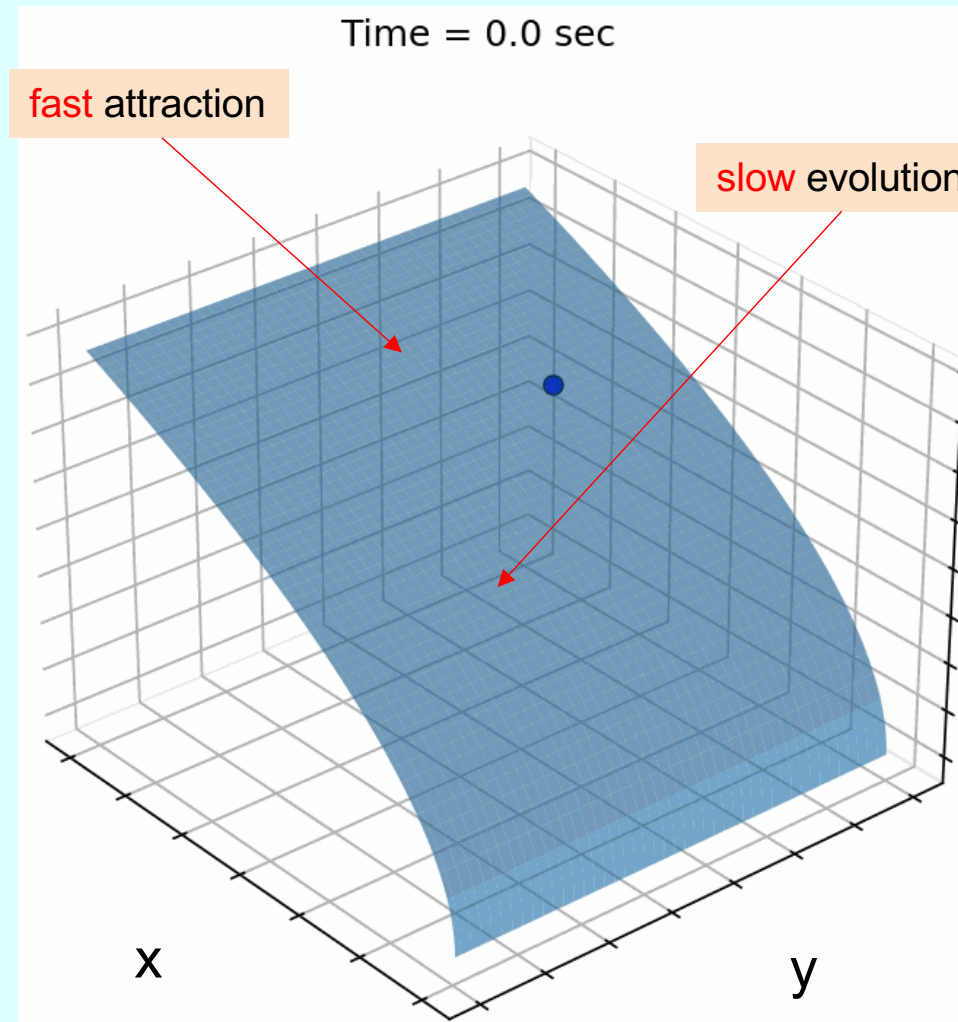


Homogeneous syngas/air system onto the CO₂ - H₂O - OH space

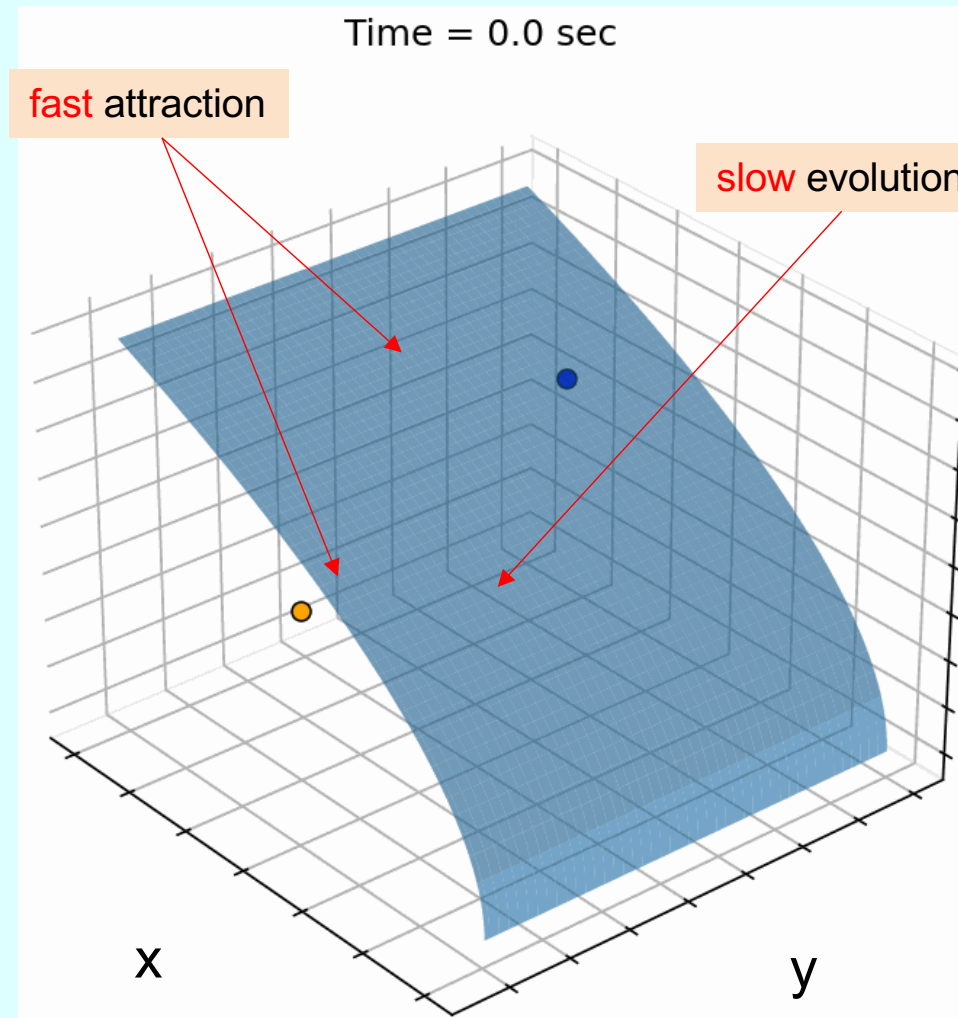
Bykov & Maas, CTM 2007



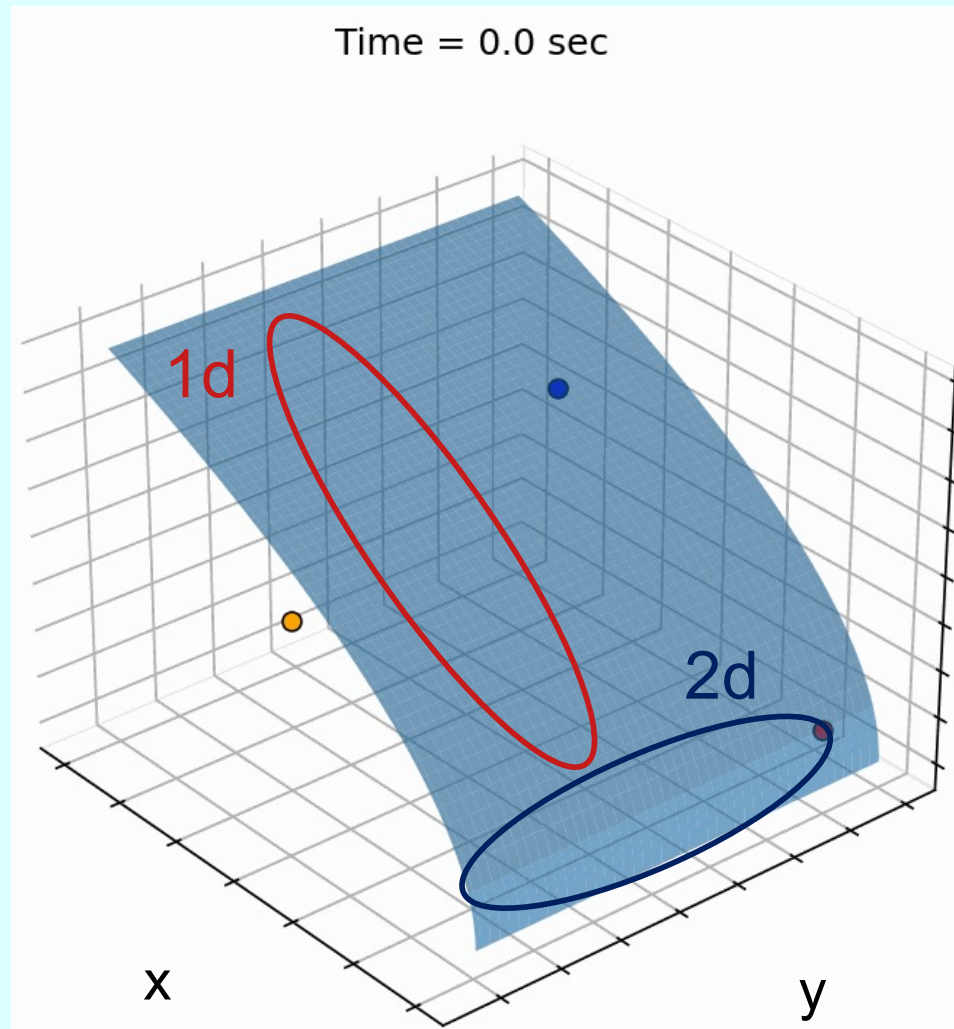
Fast-Slow transitions



Fast-Slow transitions



Fast-Slow transitions



Find:

- (i) the constraints
- (ii) the SIMPLE model that governs the slow evolution



Multi-scale analysis



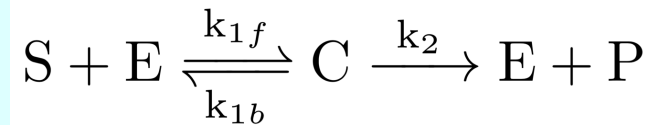
Henri Poincaré

Celestial mechanics

"On curves defined by differential equations"
(1881–1882)

Methods of: Asymptotic expansions
Singular perturbations

Very tedious procedures
(most dreadful graduate course !!)



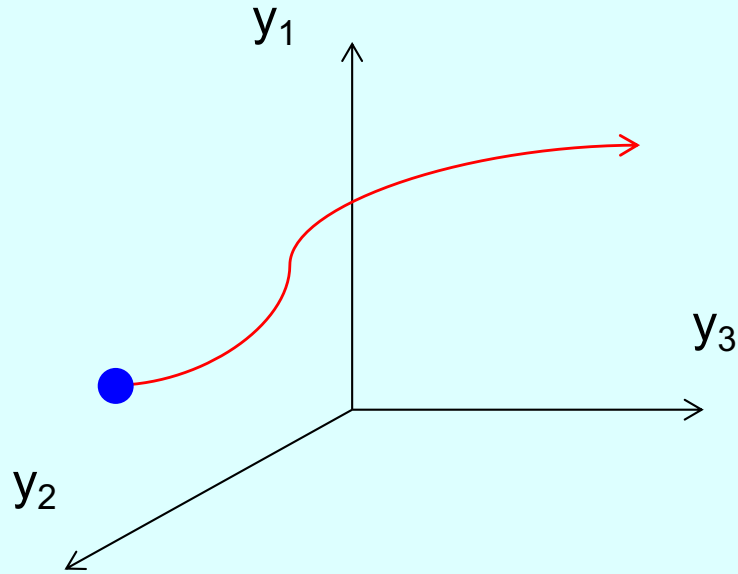
Michaelis – Menten model

- [1] V. Henri, Lois générales de l'action des diastases, Librairie Scientifique A. Hermann, 1903.
- [2] A. J. Brown, Enzyme action, Journal of the Chemical Society, Transactions 81 (1902) 373–388.
- [3] L. Michaelis, M. L. Menten, Die kinetik der invertinwirkung, Biochem Z. 49 (1913) 333–369.

Most analyzed multi-scale model (2D)
1st global reduced model in 2018



Phase space



$\mathbf{g}(\mathbf{y})$: chemical kinetics source term:

y_i : concentration of i -th species

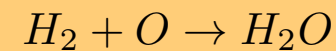
R^k : rate of k -th reaction

$\mathbf{S}_k$: stoichiometric vector of k -th reaction

$$\frac{d}{dt} \begin{bmatrix} y_1 \\ y_2 \\ y_3 \end{bmatrix} = \begin{bmatrix} g_1(y_1, y_2, y_3) \\ g_2(y_1, y_2, y_3) \\ g_3(y_1, y_2, y_3) \end{bmatrix}$$

$$\frac{d\mathbf{y}}{dt} = \mathbf{g}(\mathbf{y})$$

$$\mathbf{g}(\mathbf{y}) = \mathbf{S}_1 R^1 + \mathbf{S}_2 R^2 + \dots + \mathbf{S}_K R^K$$

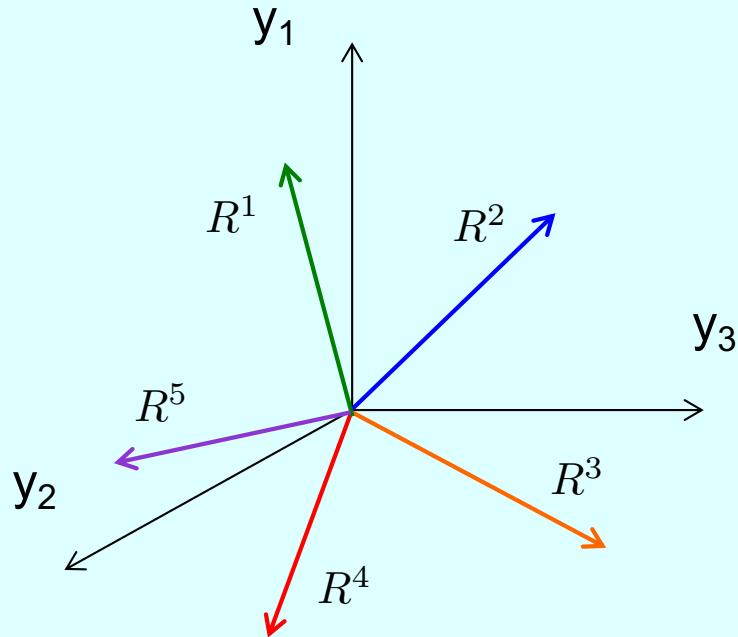


$$y_1 = [H_2], \quad y_2 = [O], \quad y_3 = [H_2O] \quad \mathbf{S} = \begin{bmatrix} -1 \\ -1 \\ 1 \end{bmatrix}$$

$$R = k[H_2][O]$$



Stoichiometric vectors – Reaction rates



$$\frac{dy}{dt} = g(y)$$

$$g(y) = S_1 R^1 + S_2 R^2 + \dots + S_K R^K$$

The i -th stoichiometric vector S_i points to the direction in which the i -th reaction tends to move the system

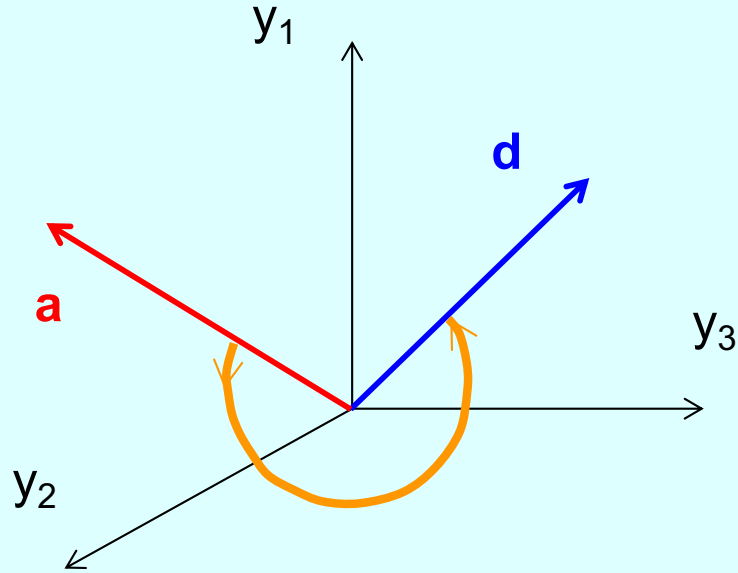
Which reactions drive the system or determine the characteristic time scale?

Which part of phase space is accessible by the trajectories?

Does the influence of the i -th reaction depends on the magnitude of its rate R^i ?

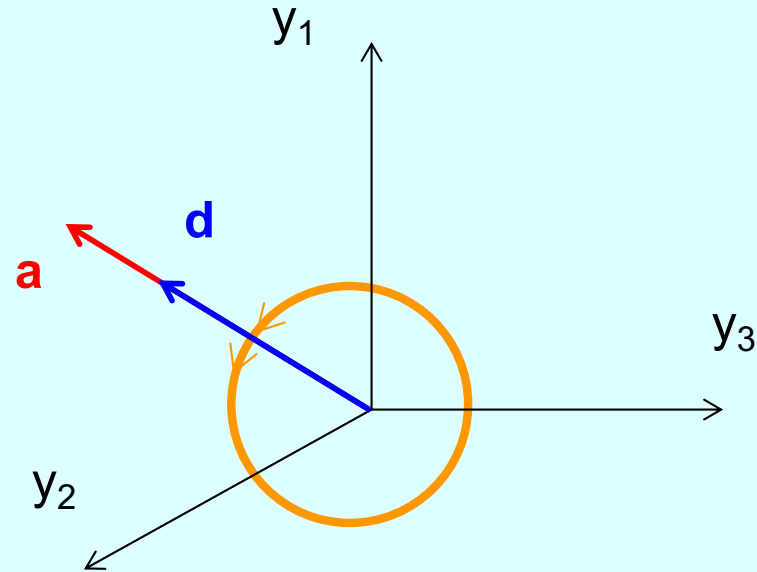


Eigenvalues and Eigenvectors



$$\begin{bmatrix} C_1^1 & C_2^1 & C_3^1 \\ C_1^2 & C_2^2 & C_3^2 \\ C_1^3 & C_2^3 & C_3^3 \end{bmatrix} \begin{bmatrix} a_1 \\ a_2 \\ a_3 \end{bmatrix} = \begin{bmatrix} d_1 \\ d_2 \\ d_3 \end{bmatrix}$$

$$\mathbf{C}\mathbf{a} = \mathbf{d}$$

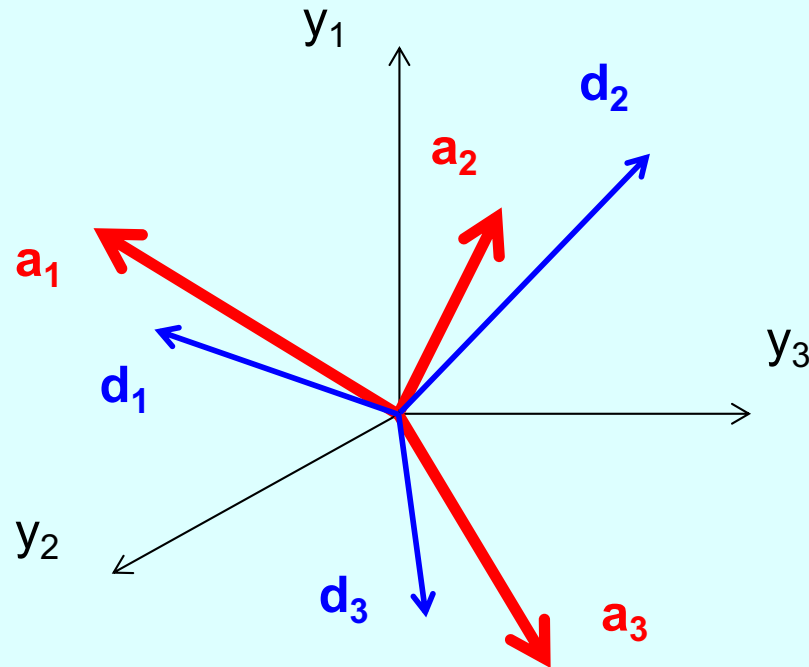


$$\begin{bmatrix} C_1^1 & C_2^1 & C_3^1 \\ C_1^2 & C_2^2 & C_3^2 \\ C_1^3 & C_2^3 & C_3^3 \end{bmatrix} \begin{bmatrix} a_1 \\ a_2 \\ a_3 \end{bmatrix} = \begin{bmatrix} d_1 \\ d_2 \\ d_3 \end{bmatrix} = \lambda \begin{bmatrix} a_1 \\ a_2 \\ a_3 \end{bmatrix}$$

$$\mathbf{C}\mathbf{a} = \mathbf{d} = \lambda\mathbf{a}$$



Eigenvalues and Eigenvectors



In general, given a $N \times N$ matrix there are N eigenvalues and N linearly independent eigenvectors

$$\mathbf{C}\mathbf{a}_n = \mathbf{d}_n = \lambda_n \mathbf{a}_n$$

Right (column) eigenvectors: $\mathbf{a} = [\mathbf{a}_1 \quad \mathbf{a}_2 \quad \dots \quad \mathbf{a}_N]$ $\mathbf{C}\mathbf{a}_n = \lambda_n \mathbf{a}_n$

Left (row) eigenvectors: $\mathbf{b} = \begin{bmatrix} \mathbf{b}^1 \\ \mathbf{b}^2 \\ \vdots \\ \mathbf{b}^N \end{bmatrix}$ $\mathbf{b}^n \mathbf{C} = \lambda_n \mathbf{b}^n$ $\mathbf{b}^i \cdot \mathbf{a}_j = \delta_j^i$



Time scales

$$\frac{dy}{dt} = g(y)$$

$$y = y_o + \delta y$$

$$\frac{d\delta y}{dt} = g(y_o) + J(y_o)\delta y$$

$$\delta y(0) = 0$$

$$J(y_o) = \nabla g(y_o)$$

$$\delta y = a_1 \left(e^{\lambda_1 t} - 1 \right) \frac{b^1 \cdot g(y_o)}{\lambda_1} + a_2 \left(e^{\lambda_2 t} - 1 \right) \frac{b^2 \cdot g(y_o)}{\lambda_2} + \dots$$

$$\tau_i = |\lambda_i|^{-1}$$

$$\begin{aligned} \lambda_i < 0 & \quad \tau_i : \text{dissipative} \\ \lambda_i > 0 & \quad \tau_i : \text{explosive} \end{aligned}$$

When $\frac{b^m \cdot g(y_o)}{\lambda_m} \approx 0 \quad m=1, \dots, M$

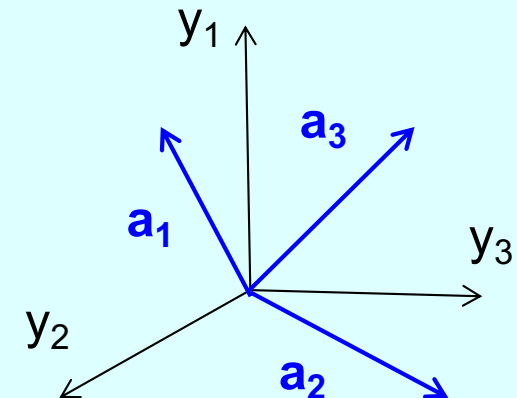
Fast time scales: $\tau_m = |\lambda_m|^{-1} \quad m=1, \dots, M$

Characteristic time scale: $\tau_{ch} = |\lambda_{M+1}|^{-1}$

a_n Right eigenvector

b^n Left eigenvector

$$|\lambda_1| > |\lambda_2| > \dots > |\lambda_N|$$



Fast / slow directions

$$\frac{dy}{dt} = g(y)$$

$$y = y_o + \delta y$$

$$\frac{d\delta y}{dt} = g(y_o) + J(y_o)\delta y$$

$$\delta y(0) = 0$$

$$J(y_o) = \nabla g(y_o)$$

$$\delta y = a_1 \left(e^{\lambda_1 t} - 1 \right) \frac{b^1 \cdot g(y_o)}{\lambda_1} + a_2 \left(e^{\lambda_2 t} - 1 \right) \frac{b^2 \cdot g(y_o)}{\lambda_2} + \dots$$

a_n Right eigenvector

b^n Left eigenvector

$$|\lambda_1| > |\lambda_2| > \dots > |\lambda_N|$$

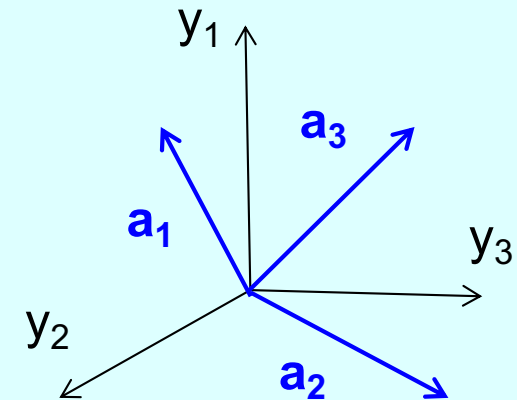
Given the orthogonality condition: $I = a_1 b^1 + a_2 b^2 + \dots + a_N b^N$

it follows that the vector field can be decomposed as:

$$g(y_o) = a_1 (b^1 \cdot g(y_o)) + a_2 (b^2 \cdot g(y_o)) + \dots + a_N (b^N \cdot g(y_o))$$

When $\tau_m \ll \tau_{ch}$, i.e.: $\frac{b^m \cdot g(y_o)}{\lambda_m} \approx 0 \quad m = 1, \dots, M$

The solution does not move along $a_m \quad m = 1, \dots, M$



FAST directions



Fast / slow directions

$$\frac{dy}{dt} = \mathbf{g}(\mathbf{y})$$

$$\mathbf{y} = \mathbf{y}_o + \delta \mathbf{y}$$

$$\frac{d\delta \mathbf{y}}{dt} = \mathbf{g}(\mathbf{y}_o) + \mathbf{J}(\mathbf{y}_o)\delta \mathbf{y}$$

$$\delta \mathbf{y}(0) = \mathbf{0}$$

$$\mathbf{J}(\mathbf{y}_o) = \nabla \mathbf{g}(\mathbf{y}_o)$$

When $\tau_m \ll \tau_{ch}$, i.e.: $\frac{\mathbf{b}^m \cdot \mathbf{g}(\mathbf{y}_o)}{\lambda_m} \approx 0 \quad m = 1, \dots, M$

The solution moves along $\mathbf{a}_m \quad m = M+1, \dots, N$

← SLOW directions

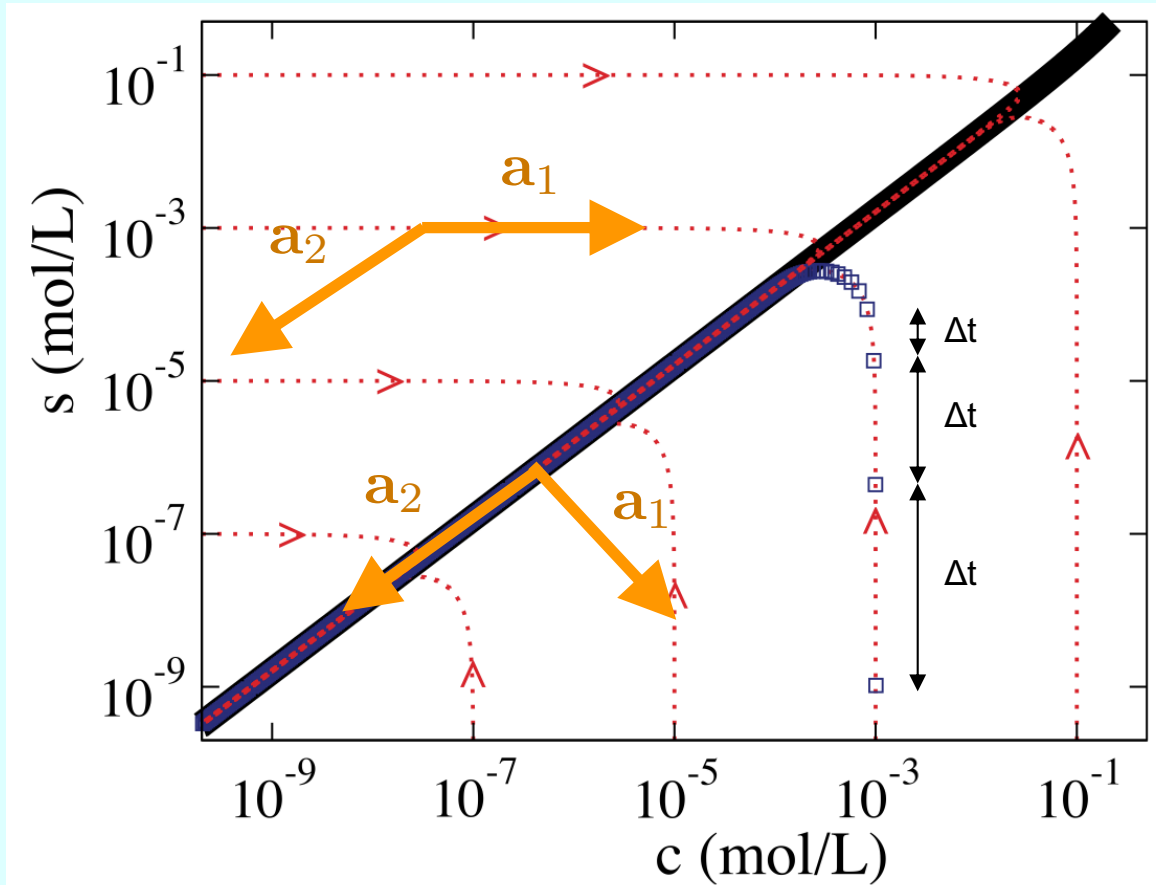
$$\delta \mathbf{y} \approx \mathbf{a}_{M+1} \left(e^{\lambda_{M+1} t} - 1 \right) \frac{\mathbf{b}^{M+1} \cdot \mathbf{g}(\mathbf{y}_o)}{\lambda_{M+1}} + \dots + \mathbf{a}_N \left(e^{\lambda_N t} - 1 \right) \frac{\mathbf{b}^N \cdot \mathbf{g}(\mathbf{y}_o)}{\lambda_N}$$

This is the solution of: $\frac{dy}{dt} = \left[\mathbf{a}_{M+1} \mathbf{b}^{M+1} + \dots + \mathbf{a}_N \mathbf{b}^N \right] \mathbf{g}(\mathbf{y})$

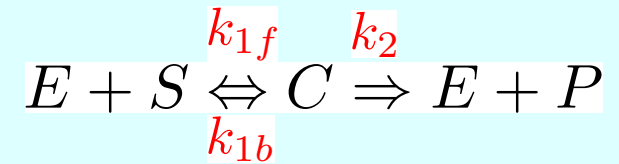
the components of $\mathbf{g}$ along the slow directions



Fast / slow dynamics



Michaelis – Menten model



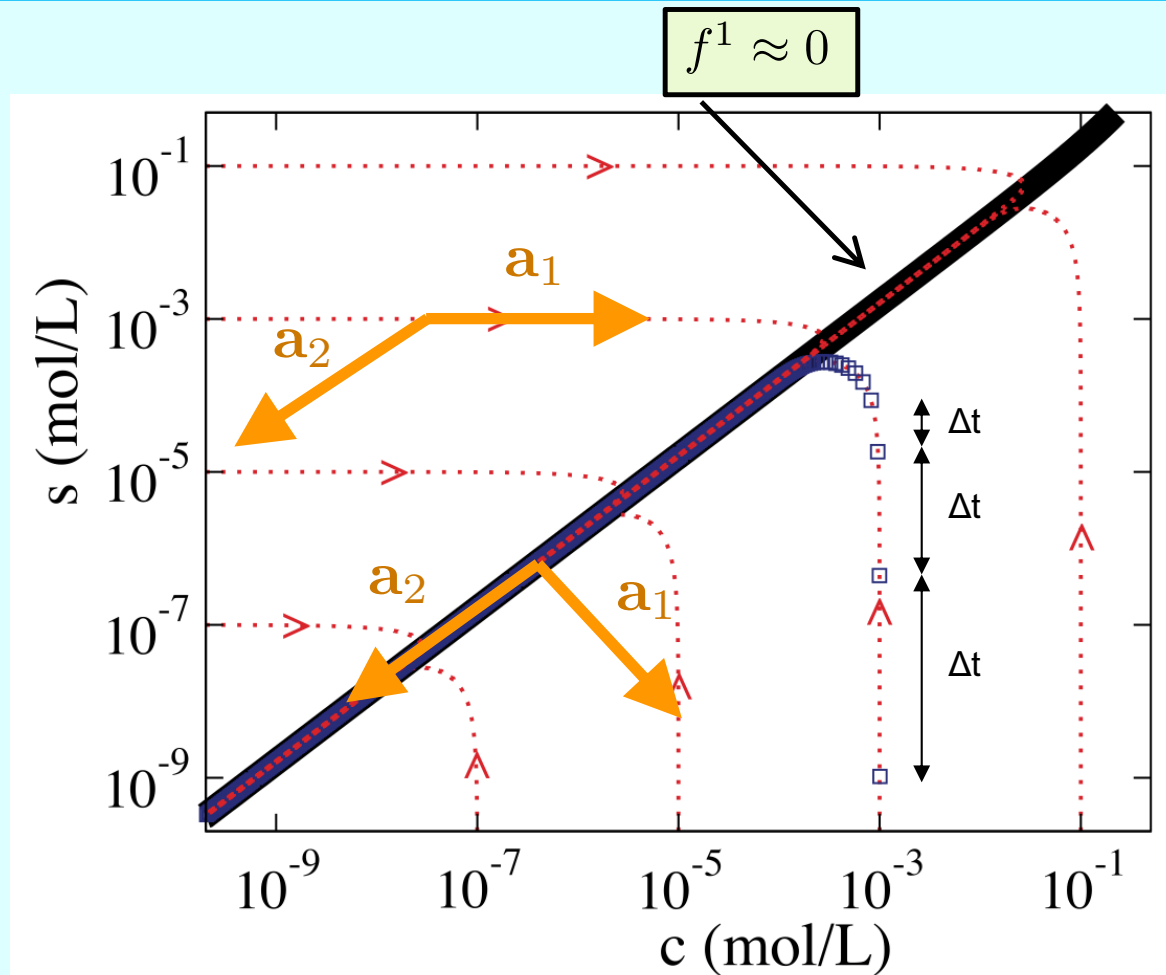
$$\begin{aligned} \frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} &= \mathbf{S}_1 R^1 + \mathbf{S}_2 R^2 = \mathbf{g} \\ &= \mathbf{a}_1 f^1 + \mathbf{a}_2 f^2 \\ &\quad \text{fast} \quad \text{slow} \end{aligned}$$

$$f^i = \mathbf{b}^i \cdot \mathbf{g}$$

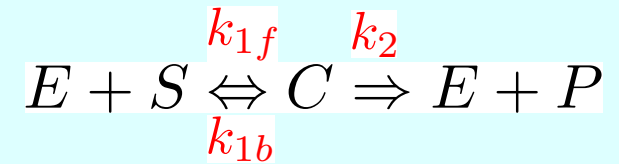
$$\mathbf{b}^i \cdot \mathbf{a}_j = \delta_j^i$$



Fast / slow dynamics



Michaelis – Menten model



$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} = \underset{\text{fast}}{a_1 f^1} + \underset{\text{slow}}{a_2 f^2}$$

Fast transient

$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} \approx a_1 f^1 \quad (f^1 \gg f^2)$$

Slow evolution

$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} \approx a_2 f^2 \quad f^1 \approx 0$$

Slow reduced model

Conventional methodologies: QSSA and PEA

$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} = \begin{bmatrix} -1 \\ 1 \end{bmatrix} R^1 + \begin{bmatrix} 0 \\ -1 \end{bmatrix} R^2 = \begin{bmatrix} g_s \\ g_c \end{bmatrix}$$

$$R^1 = k_{1f}(e_o - c)s - k_{1b}c \quad R^2 = k_2c$$

$QSSA_{c,R^1}$

$$c, R^1 : fast \rightarrow g_c \approx 0 \quad R^1 \approx R^2 \quad \frac{ds}{dt} \approx -R^2$$

$QSSA_{c,R^2}$

$$c, R^2 : fast \rightarrow g_c \approx 0 \quad R^2 \approx R^1 \quad \frac{ds}{dt} \approx -R^1$$

$QSSA_{s,R^1}$

$$s, R^1 : fast \rightarrow g_s \approx 0 \quad R^1 \approx 0 \quad \frac{dc}{dt} = -R^2$$

PEA_{R^1}

$$R^1 : fast \rightarrow \frac{dR^1}{dt} \approx 0 \quad R^1 \approx \frac{R^2}{1 + \nu} \quad \frac{ds}{dt} \approx -\frac{R^2}{1 + \nu}$$

$$\frac{dc}{dt} \approx -\frac{\nu R^2}{1 + \nu}$$

$$c = \frac{k_{1f}e_os}{(k_{1f}s + k_{1b} + k_2)}$$

$$s = \frac{k_{1b}c}{k_{1f}(e_o - c)}$$

$$\nu \rightarrow 0$$

$$QSSA_{c,R^1}$$

$$\nu \rightarrow \infty$$

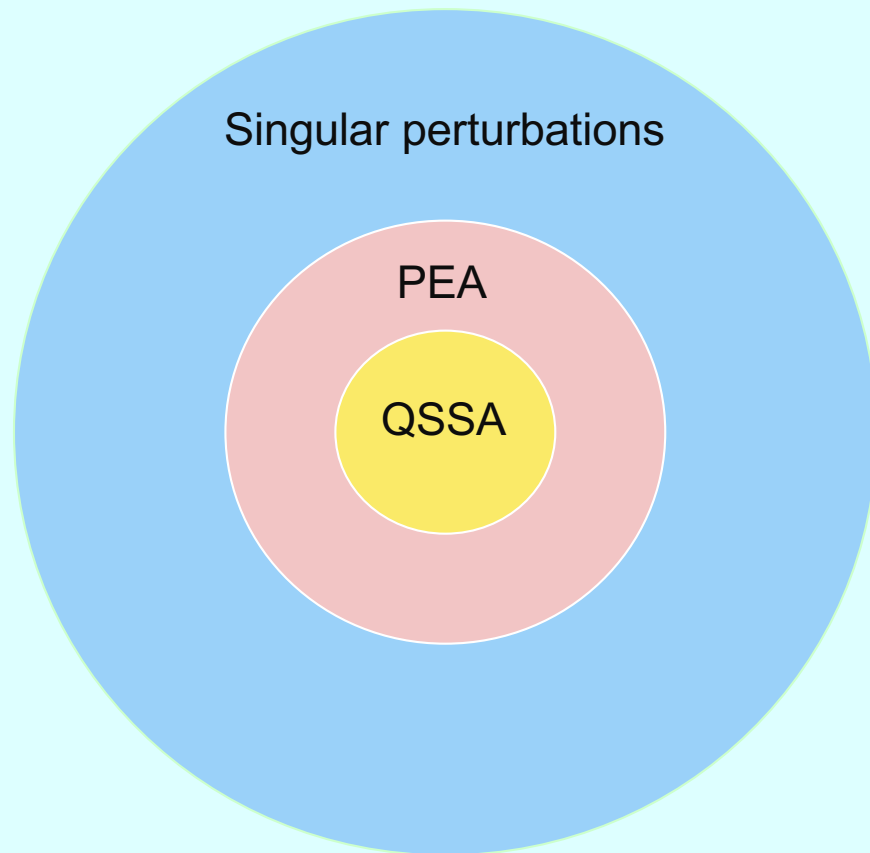
$$QSSA_{s,R^1}$$

$$\nu = \frac{e_o - c}{K_R + s}$$

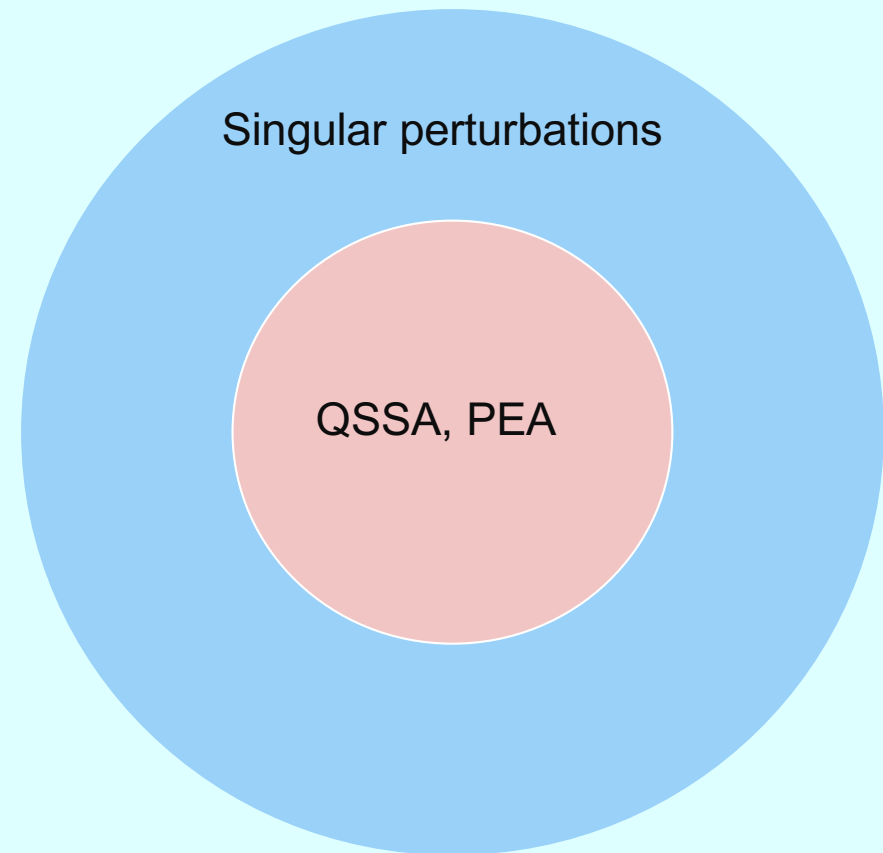
$$K_R = \frac{k_{1b}}{k_{1f}}$$

QSSA/PEA: stability and accuracy

Accuracy



Stability



Goussis, *CTM* 16 (2012)

Conventional methodologies: QSSA and PEA

$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} = \begin{bmatrix} -1 \\ 1 \end{bmatrix} R^1 + \begin{bmatrix} 0 \\ -1 \end{bmatrix} R^2 = \begin{bmatrix} g_s \\ g_c \end{bmatrix}$$

$$R^1 = k_{1f}(e_o - c)s - k_{1b}c \quad R^2 = k_2c$$

$QSSA_{c,R^1}$

$$c, R^1 : fast \rightarrow g_c \approx 0 \quad R^1 \approx R^2 \quad \frac{ds}{dt} \approx -R^2$$

$QSSA_{c,R^2}$

$$c, R^2 : fast \rightarrow g_c \approx 0 \quad R^2 \approx R^1 \quad \frac{ds}{dt} \approx -R^1$$

$QSSA_{s,R^1}$

$$s, R^1 : fast \rightarrow g_s \approx 0 \quad R^1 \approx 0 \quad \frac{dc}{dt} = -R^2$$

$$c = \frac{k_{1f}e_os}{(k_{1f}s + k_{1b} + k_2)}$$

$$s = \frac{k_{1b}c}{k_{1f}(e_o - c)}$$

PEA_{R^1}

$$R^1 : fast \rightarrow \frac{dR^1}{dt} \approx 0 \quad R^1 \approx \frac{R^2}{1 + \nu} \quad \frac{ds}{dt} \approx -\frac{R^2}{1 + \nu}$$

$$\frac{dc}{dt} \approx -\frac{\nu R^2}{1 + \nu}$$

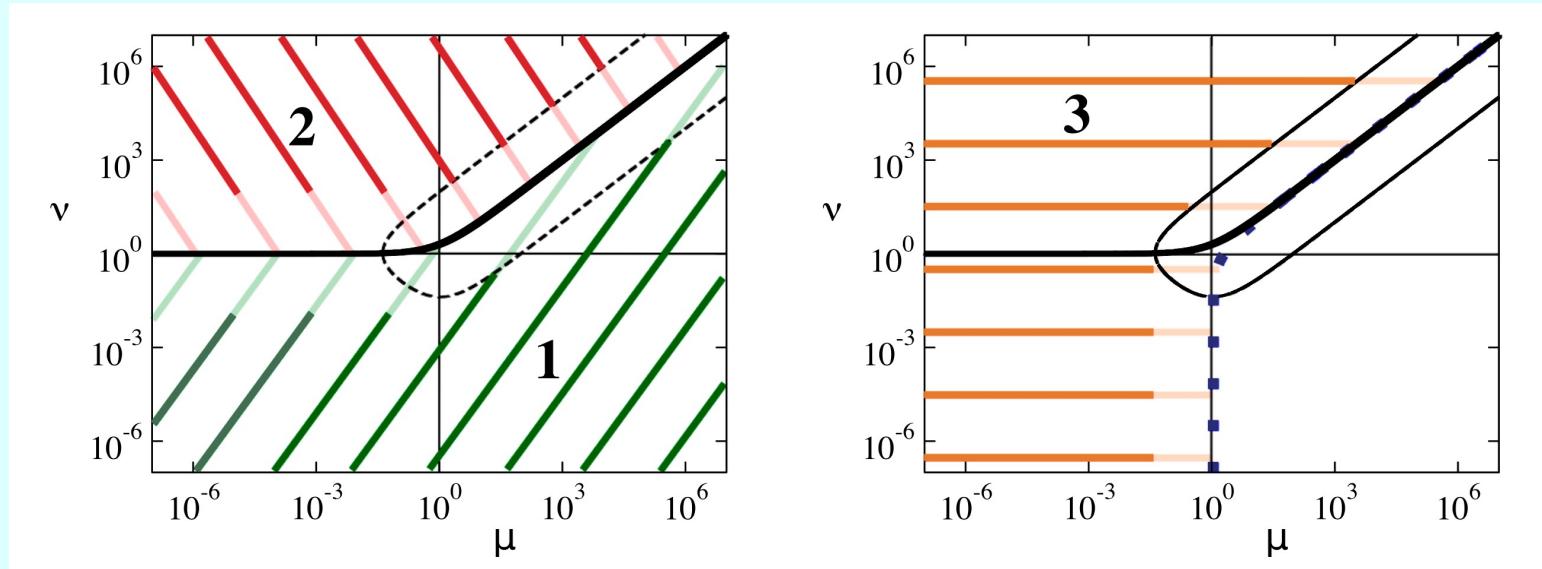
$$\nu \rightarrow 0 \\ QSSA_{c,R^1}$$

$$\nu \rightarrow \infty \\ QSSA_{s,R^1}$$

$$\nu = \frac{e_o - c}{K_R + s}$$

$$K_R = \frac{k_{1b}}{k_{1f}}$$

Conventional methodologies in the MM model



$$\nu = \frac{e_o - c}{K_R + s}$$

$$\mu = \frac{K}{K_R + s}$$

$$K_R = \frac{k_{1b}}{k_{1f}}$$

$$K = \frac{k_2}{k_{1f}}$$

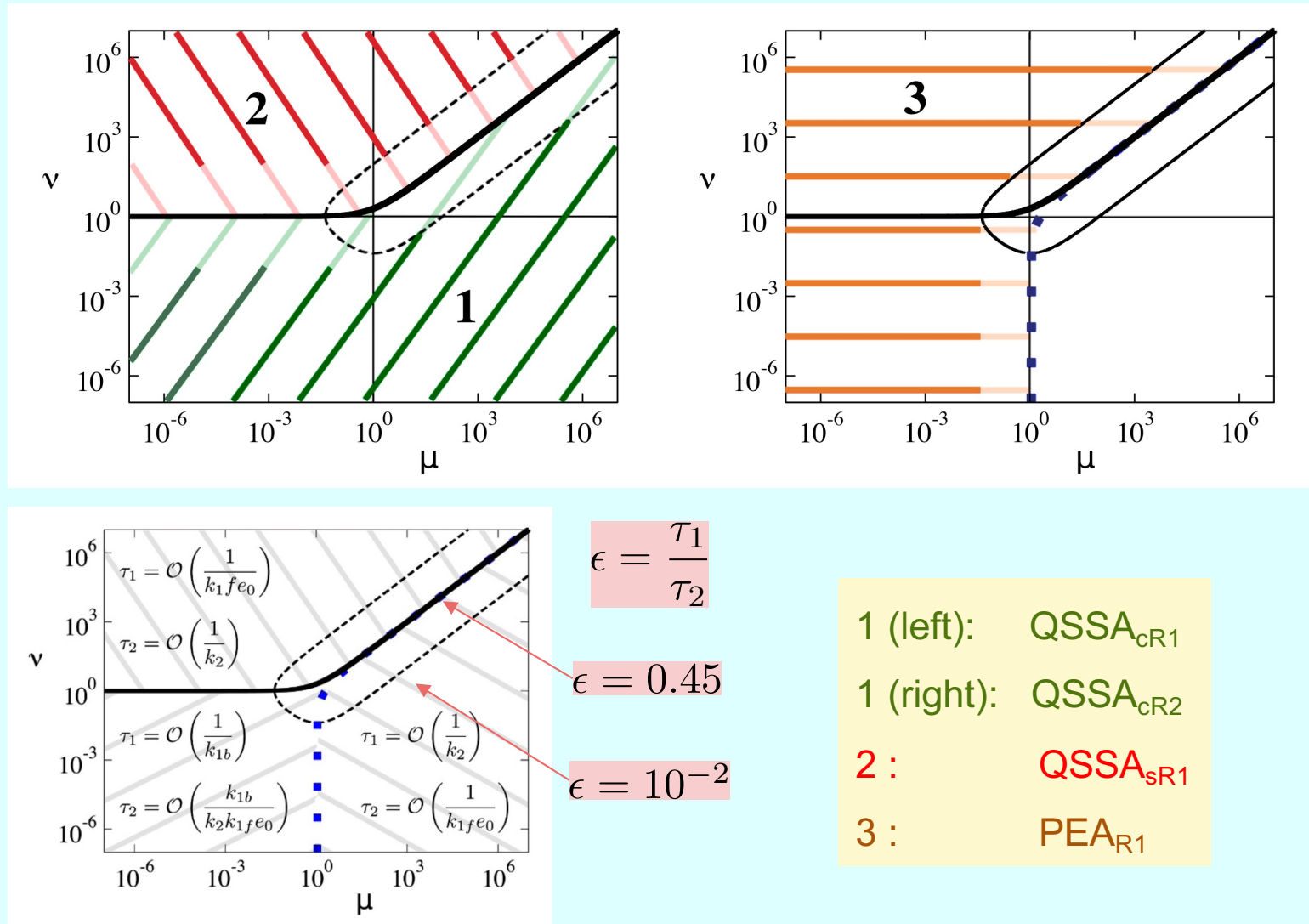
1 (left): $QSSA_{cR1}$

1 (right): $QSSA_{cR2}$

2 : $QSSA_{sR1}$

3 : PEA_{R1}

Conventional methodologies in the MM model



Obstacles in conventional methodologies

Identification of:

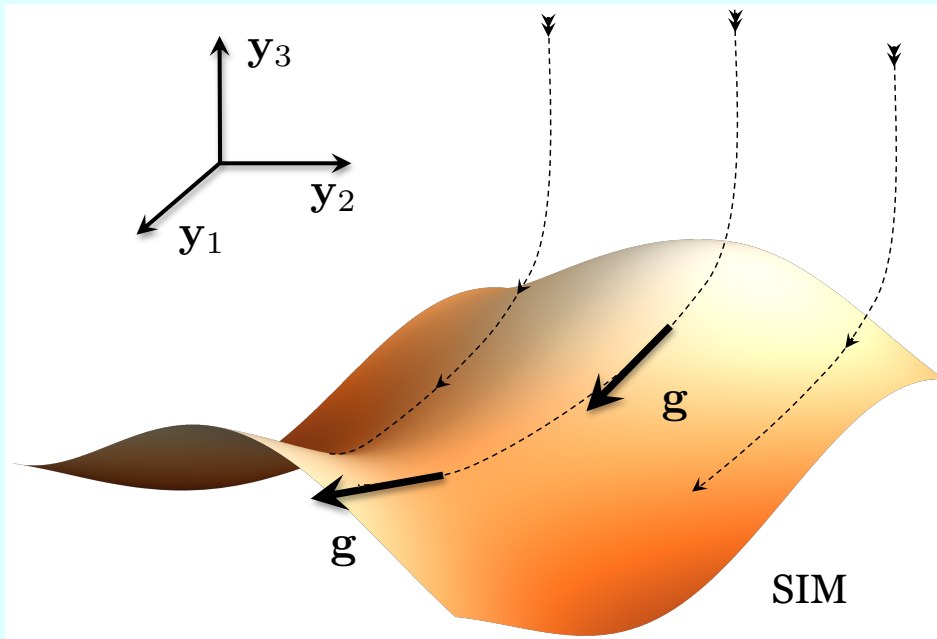
- i. Fast/slow/decoupled variables (e.g., species)
- ii. Fast/slow/unimportant processes (e.g., reactions)
- iii. Proper non-dimensional form of governing equations
- iv. Small parameter ε ; $\varepsilon \rightarrow 0$ yields the reduced model

Additional obstacles:

- Items in (i) and (ii) vary with time
- Number of items in (i) and (ii) vary with time
- Different sets of items in (i) and (ii) refer to different non-dimensional forms and ε .



Alternative view: Geometrical Singular Perturbations



Hirsch (1977), Fenichel (1979), Jones (1995)

Goals:

- Identify dynamical *structures* in phase space (SIM)
- Exploit their *properties*, such as fast/slow decomposition

Basic tools:

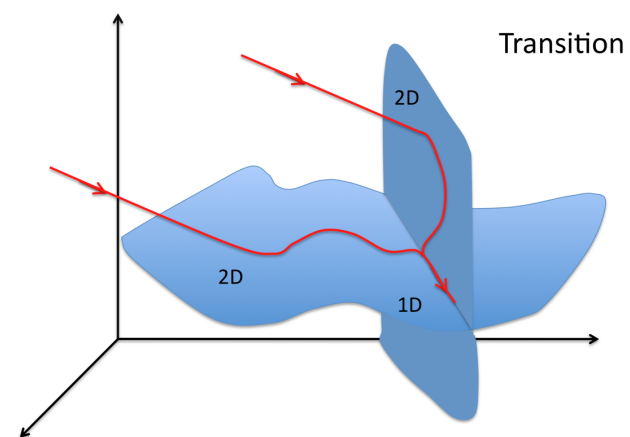
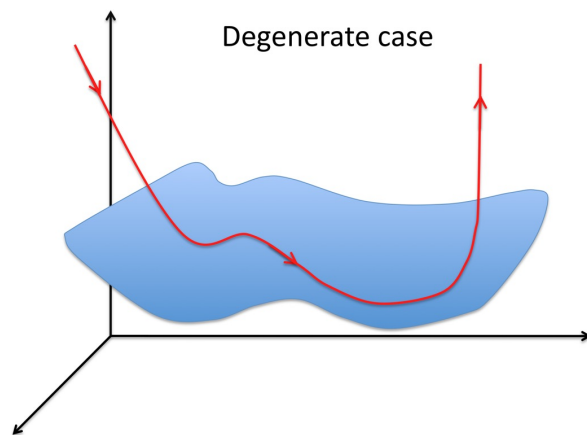
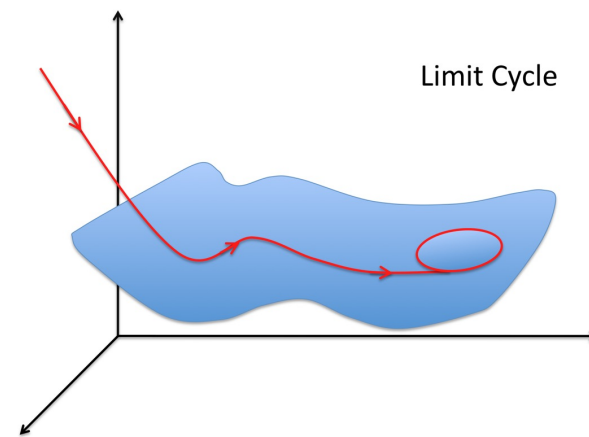
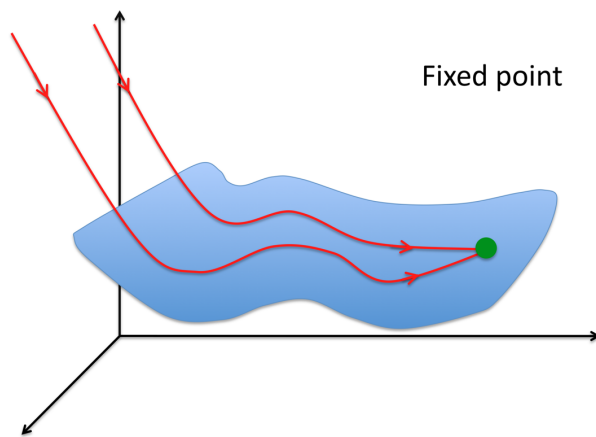
- The *tangent space* (the tangent bundle)
- Fast and slow *sub-domains* of tangent space

$$\frac{dy}{dt} = g(y)$$

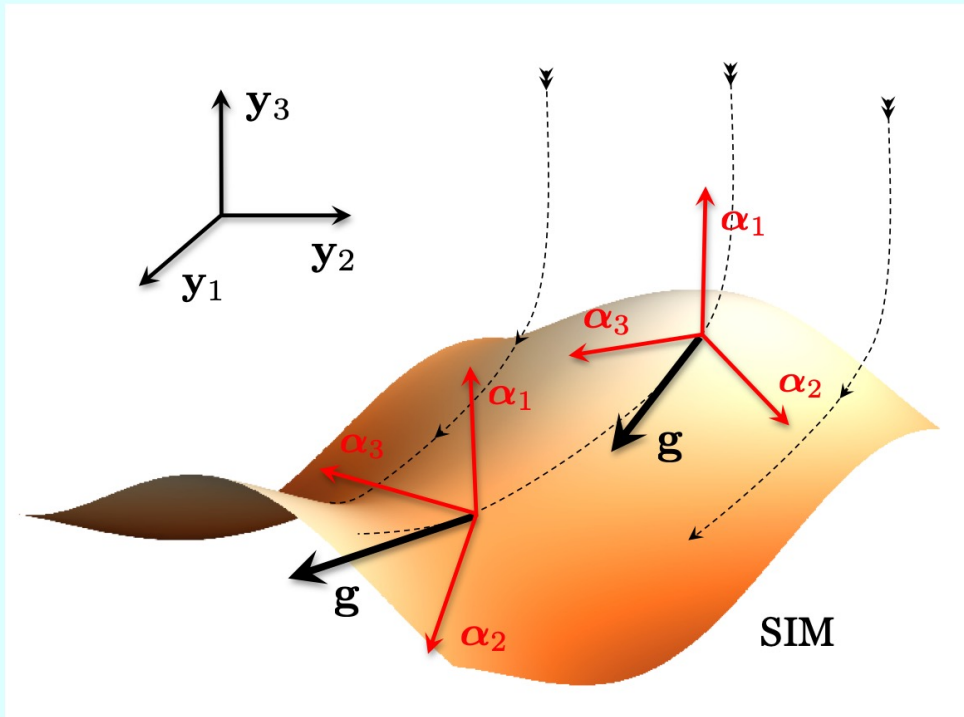
(Tasso Kaper 1999 - CKRT Jones 2006)



Structures in phase space



Alternative view: Computational Singular Perturbations (CSP)



$$\frac{dy}{dt} = g(y)$$

$$\frac{dy}{dt} = \underbrace{a_1 f^1}_{\text{fast}} + \underbrace{a_2 f^2 + a_3 f^3}_{\text{slow}}$$

Slow Invariant
Manifold (SIM)

$$f^1 \approx 0$$

Reduced (slow)
model

$$\frac{dy}{dt} \approx a_2 f^2 + a_3 f^3$$

Lam & Goussis, *PCI* (1989)



Alternative view of conventional methodologies: QSSA_{c,R1}

$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} = \begin{bmatrix} -1 \\ 1 \end{bmatrix} R^1 + \begin{bmatrix} 0 \\ -1 \end{bmatrix} R^2 = \begin{bmatrix} g_s \\ g_c \end{bmatrix}$$

$$R^1 = k_{1f}(e_o - c)s - k_{1b}c \quad R^2 = k_2c$$

QSSA_{c,R1}

$$c, R^1 : \text{fast} \rightarrow g_c \approx 0 \quad R^1 \approx R^2 \quad \frac{ds}{dt} \approx -R^2$$

$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} = \underset{\text{fast}}{\mathbf{a}_1} f^1 + \underset{\text{slow}}{\mathbf{a}_2} f^2 \quad f^i = \mathbf{b}^i \cdot \mathbf{g}$$

$$\mathbf{a}_1 = \begin{bmatrix} -1 \\ 1 \end{bmatrix} \quad \mathbf{a}_2 = \begin{bmatrix} 1 \\ 0 \end{bmatrix} \quad \mathbf{b}^1 = \begin{bmatrix} 0 & 1 \end{bmatrix} \quad \mathbf{b}^2 = \begin{bmatrix} 1 & 1 \end{bmatrix}$$

$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} = \begin{bmatrix} -1 \\ 1 \end{bmatrix} (\cancel{R^1 - R^2}) - \begin{bmatrix} 1 \\ 0 \end{bmatrix} R^2$$

$$f^1 \approx 0$$

$$R^1 - R^2 \approx 0$$

$$\frac{ds}{dt} \approx -R^2$$

Alternative view of conventional methodologies

$$\frac{d}{dt} \begin{bmatrix} s \\ c \end{bmatrix} = \begin{bmatrix} -1 \\ 1 \end{bmatrix} R^1 + \begin{bmatrix} 0 \\ -1 \end{bmatrix} R^2 = \begin{bmatrix} g_s \\ g_c \end{bmatrix}$$

$$R^1 = k_{1f}(e_o - c)s - k_{1b}c \quad R^2 = k_2c$$

$QSSA_{c,R^1}$

$$\mathbf{a}_1 = \begin{bmatrix} -1 \\ 1 \end{bmatrix} \quad \mathbf{a}_2 = \begin{bmatrix} 1 \\ 0 \end{bmatrix} \quad \mathbf{b}^1 = \begin{bmatrix} 0 & 1 \end{bmatrix} \quad \mathbf{b}^2 = \begin{bmatrix} 1 & 1 \end{bmatrix}$$

$QSSA_{c,R^2}$

$$\mathbf{a}_1 = \begin{bmatrix} 0 \\ -1 \end{bmatrix} \quad \mathbf{a}_2 = \begin{bmatrix} 1 \\ 0 \end{bmatrix} \quad \mathbf{b}^1 = \begin{bmatrix} 0 & -1 \end{bmatrix} \quad \mathbf{b}^2 = \begin{bmatrix} 1 & 0 \end{bmatrix}$$

$QSSA_{s,R^1}$

$$\mathbf{a}_1 = \begin{bmatrix} -1 \\ 1 \end{bmatrix} \quad \mathbf{a}_2 = \begin{bmatrix} 0 \\ 1 \end{bmatrix} \quad \mathbf{b}^1 = \begin{bmatrix} -1 & 0 \end{bmatrix} \quad \mathbf{b}^2 = \begin{bmatrix} 1 & 1 \end{bmatrix}$$

PEA_{R^1}

$$\mathbf{a}_1 = \begin{bmatrix} -1 \\ 1 \end{bmatrix} \quad \mathbf{a}_2 = \begin{bmatrix} 1 \\ \nu \end{bmatrix} \quad \mathbf{b}^1 = \frac{1}{1+\nu} \begin{bmatrix} -\nu & 1 \end{bmatrix} \quad \mathbf{b}^2 = \frac{1}{1+\nu} \begin{bmatrix} 1 & 1 \end{bmatrix}$$

$\nu \rightarrow 0$
 $QSSA_{c,R^1}$

$\nu \rightarrow \infty$
 $QSSA_{s,R^1}$

CSP algorithm

$$\mathbf{y} = \begin{bmatrix} y^1 \\ \vdots \\ y^N \end{bmatrix}$$

$$\frac{d\mathbf{y}}{dt} = \mathbf{S}_1 R^1 + \mathbf{S}_2 R^2 + \cdots + \mathbf{S}_K R^K$$

$$= \mathbf{a}_1 f^1 + \cdots + \mathbf{a}_M f^M + \mathbf{a}_{M+1} f^{M+1} + \cdots + \mathbf{a}_N f^N$$

$$f^i = \mathbf{b}^i \cdot \mathbf{g}$$

$$\underbrace{\tau_1 \quad \tau_M}_{\text{fast (dissipative)}} \ll \underbrace{\tau_{M+1} \quad \tau_N}_{\text{slow}}$$

$$\epsilon = \frac{\tau_{M+1}}{\tau_M} < 1$$

$$\mathbf{a}_f = [\mathbf{a}_1 \quad \cdots \quad \mathbf{a}_M]$$

$$\mathbf{a}_s = [\mathbf{a}_{M+1} \quad \cdots \quad \mathbf{a}_N]$$

$$\mathbf{b}^f = \begin{bmatrix} \mathbf{b}^1 \\ \vdots \\ \mathbf{b}_M \end{bmatrix} \quad \mathbf{b}^s = \begin{bmatrix} \mathbf{b}^{M+1} \\ \vdots \\ \mathbf{b}_N \end{bmatrix}$$

Accuracy improvement; $\mathbf{x} = \mathbf{x}_0 + \epsilon^1 \mathbf{x}_1 + \epsilon^2 \mathbf{x}_2 + \dots$

Fixed: $\mathbf{a}^f, \mathbf{b}^s$

Refined: $\mathbf{a}^s, \mathbf{b}^f$

Stability of reduced model

Refined: $\mathbf{a}^f, \mathbf{b}^s$

Fixed: $\mathbf{a}^s, \mathbf{b}^f$

Major drawback: $M=M(t)$

Leading order accuracy:
Left and right eigenvectors



CSP algorithm

$$\mathbf{y} = \begin{bmatrix} y^1 \\ \vdots \\ y^N \end{bmatrix}$$

$$\frac{d\mathbf{y}}{dt} = \mathbf{S}_1 R^1 + \mathbf{S}_2 R^2 + \cdots + \mathbf{S}_K R^K$$

$$= \mathbf{a}_1 f^1 + \cdots + \mathbf{a}_M f^M + \mathbf{a}_{M+1} f^{M+1} + \cdots + \mathbf{a}_N f^N$$

$$f^i = \mathbf{b}^i \cdot \mathbf{g}$$

$$\underbrace{\tau_1 \quad \tau_M}_{\text{fast}} \ll \underbrace{\tau_{M+1} \quad \tau_N}_{\text{slow}}$$

$$\epsilon = \frac{\tau_{M+1}}{\tau_M} < 1$$

Reduced model:

$$f^1 = f^2 = \cdots f^M = 0$$

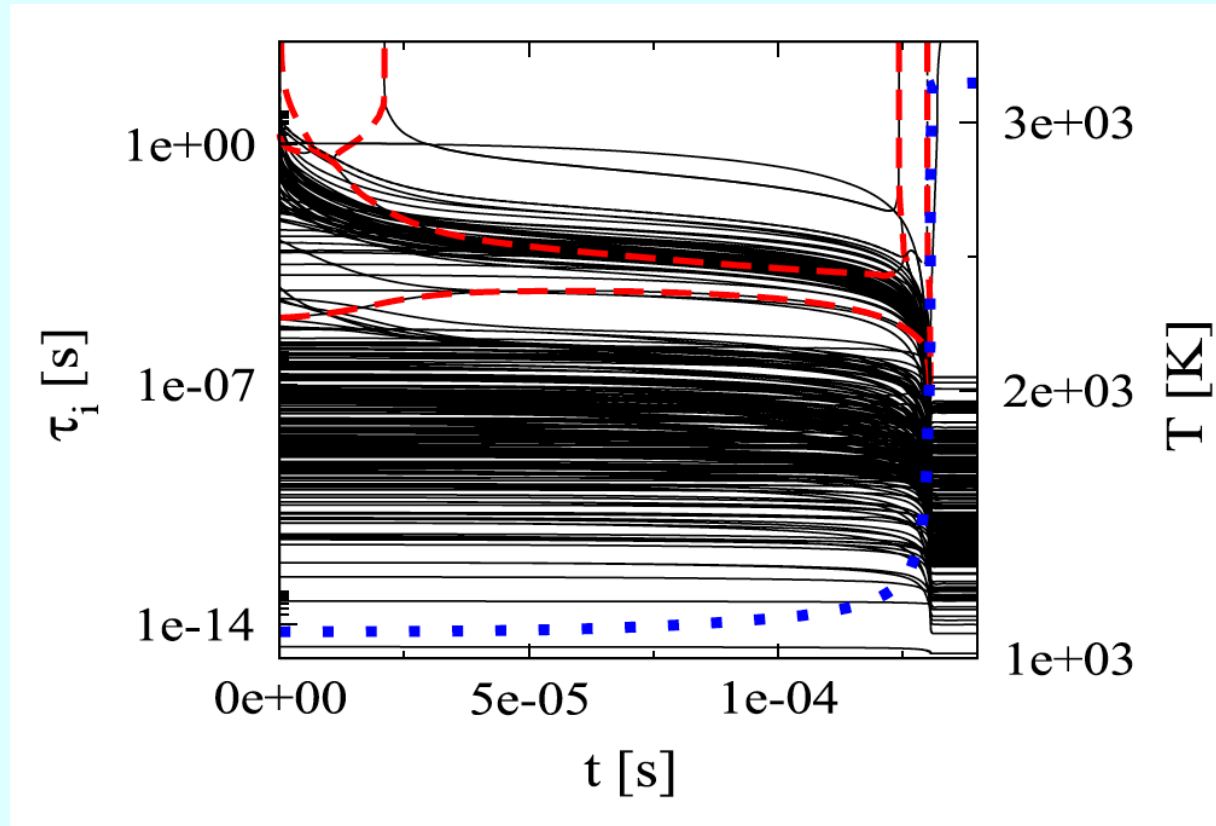
$$\tau_{char} = \tau_{M+1}$$

$$\frac{d\mathbf{y}}{dt} = \mathbf{a}_{M+1} f^{M+1} + \cdots + \mathbf{a}_N f^N$$

Major drawback: $M=M(t)$



Number of fast time scales homogeneous DME/air; $T(0)=1100\text{K}$



$N=253$ species
 $K=1542$ reversible reactions

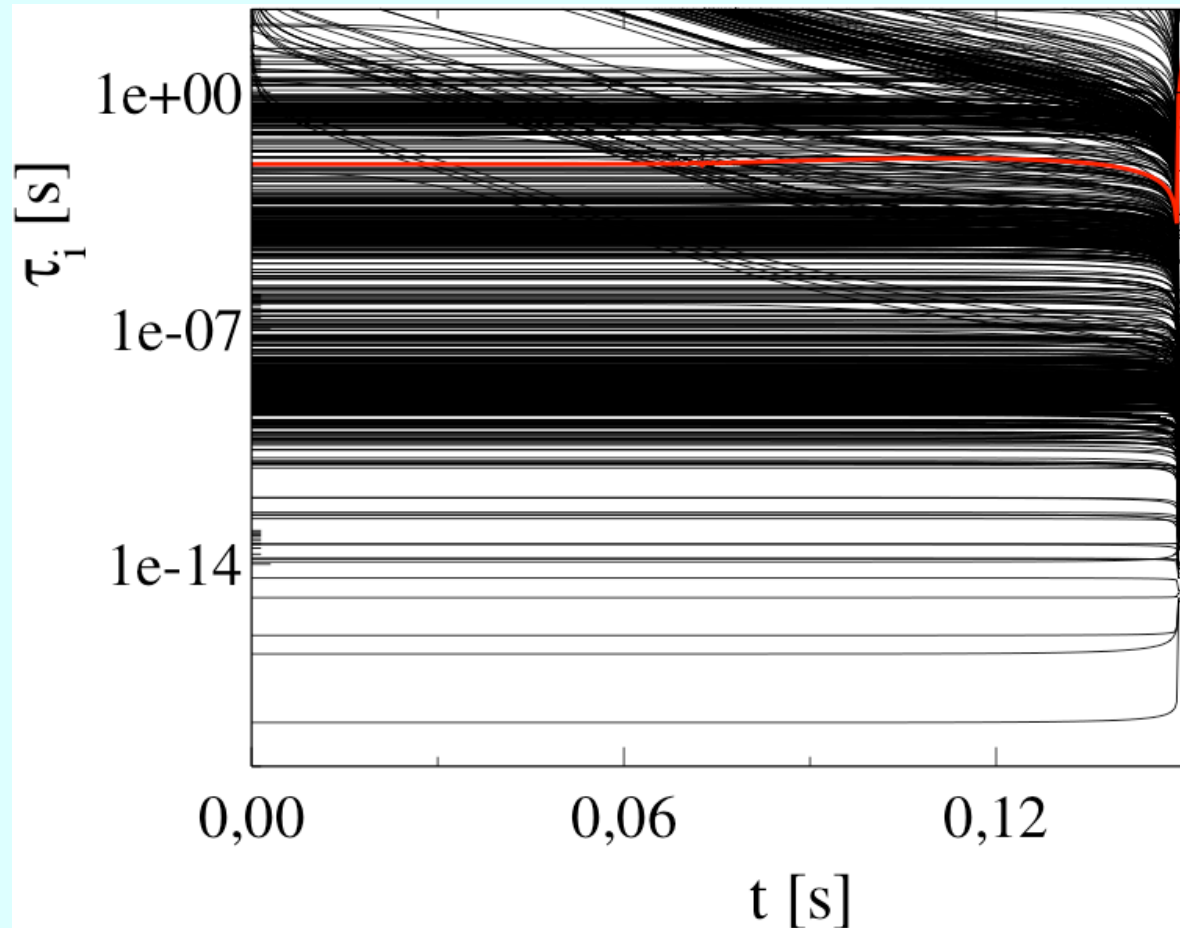
Explosive/Dissipative

Temperature

Tigas et al C&F 2015



Number of fast time scales
homogeneous n-hexane/air; $T(0)=600\text{K}$



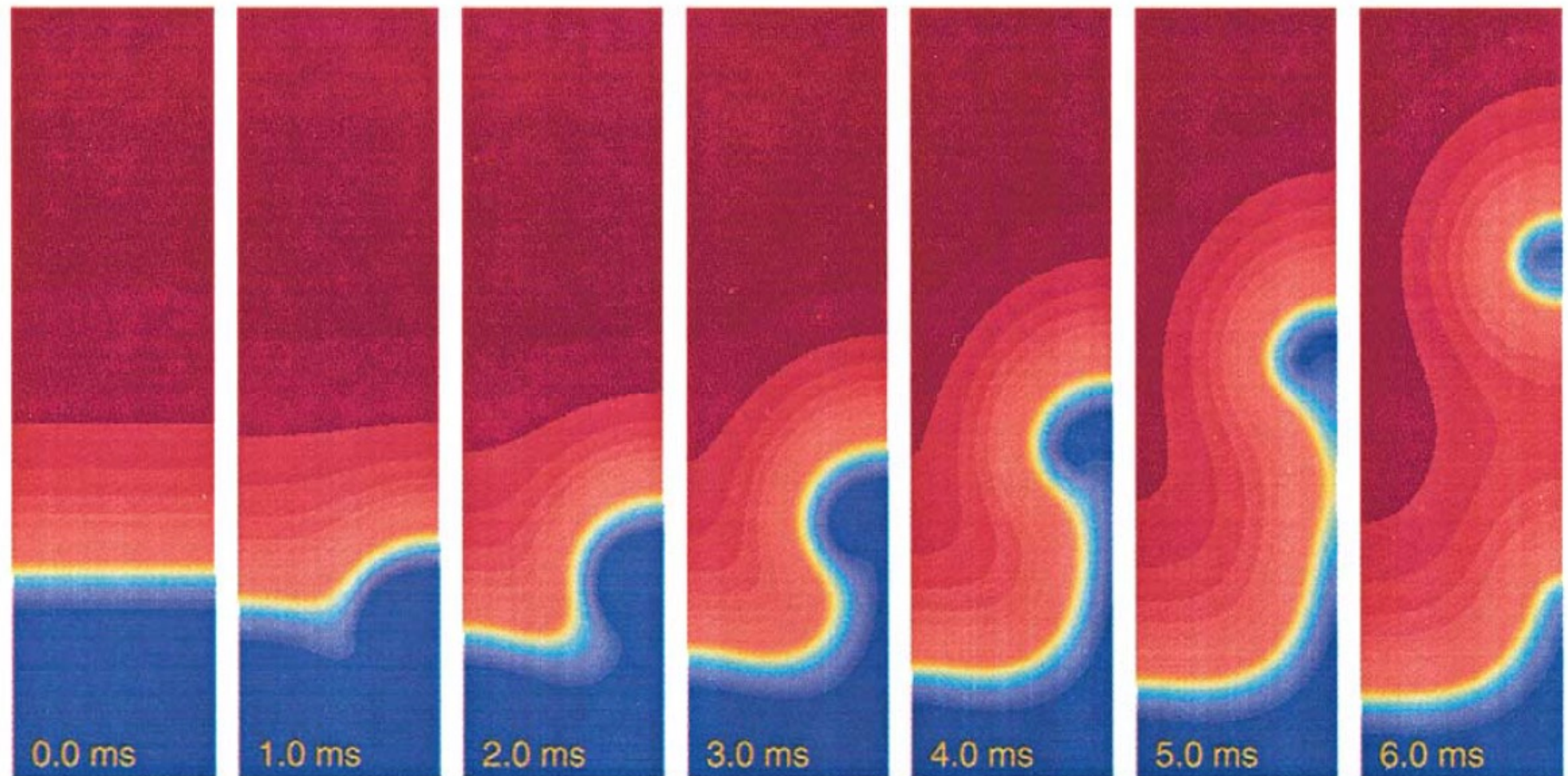
N=1118 species
K=4808 reversible reactions

Explosive/Dissipative

Tigas et al C&F 2018



Number of fast time scales transient CH₄/air flame-vortex interaction



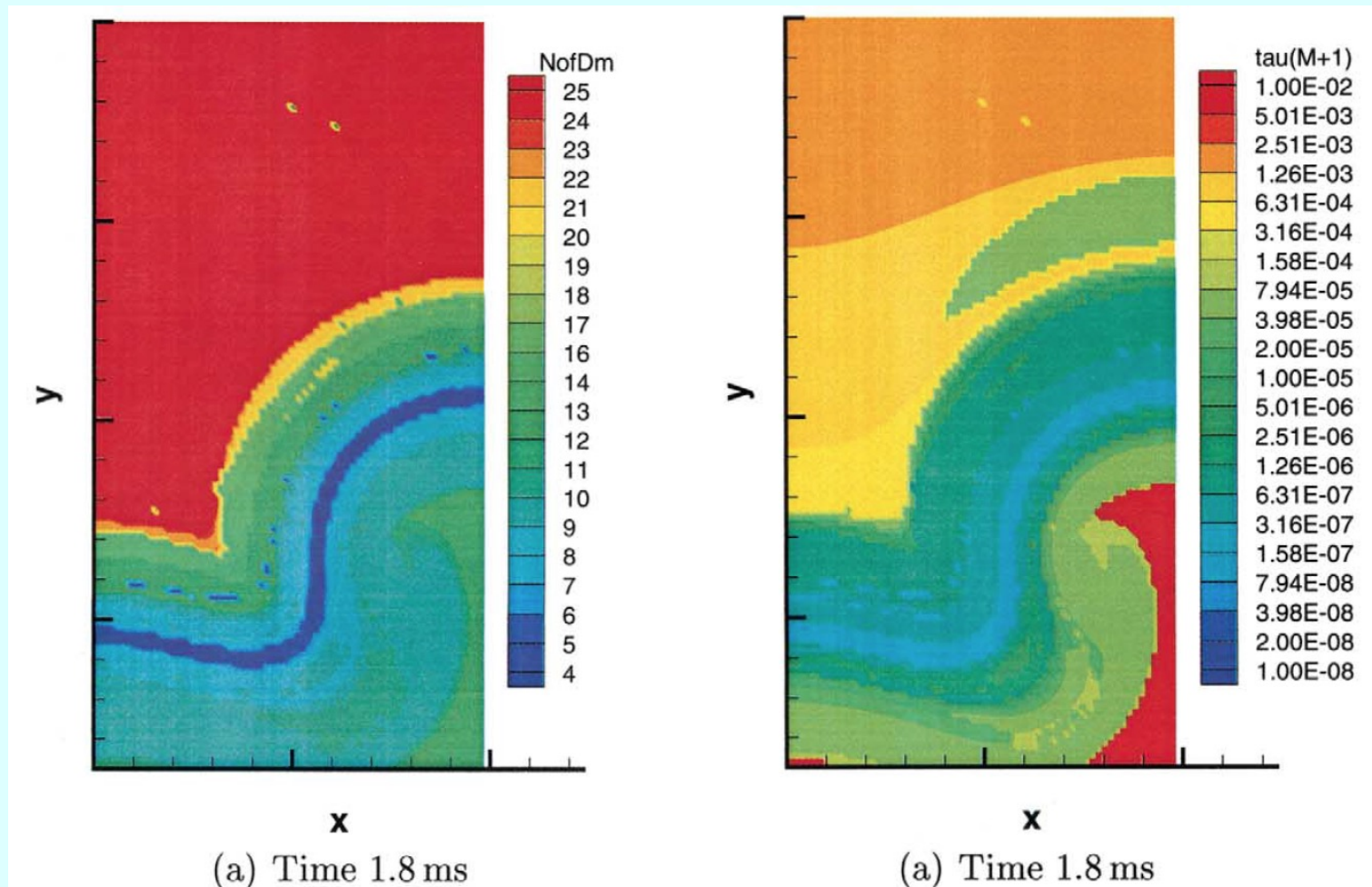
Temperature

Valorani et al, C&F 2003



Number of fast time scales transient CH₄/air flame-vortex interaction

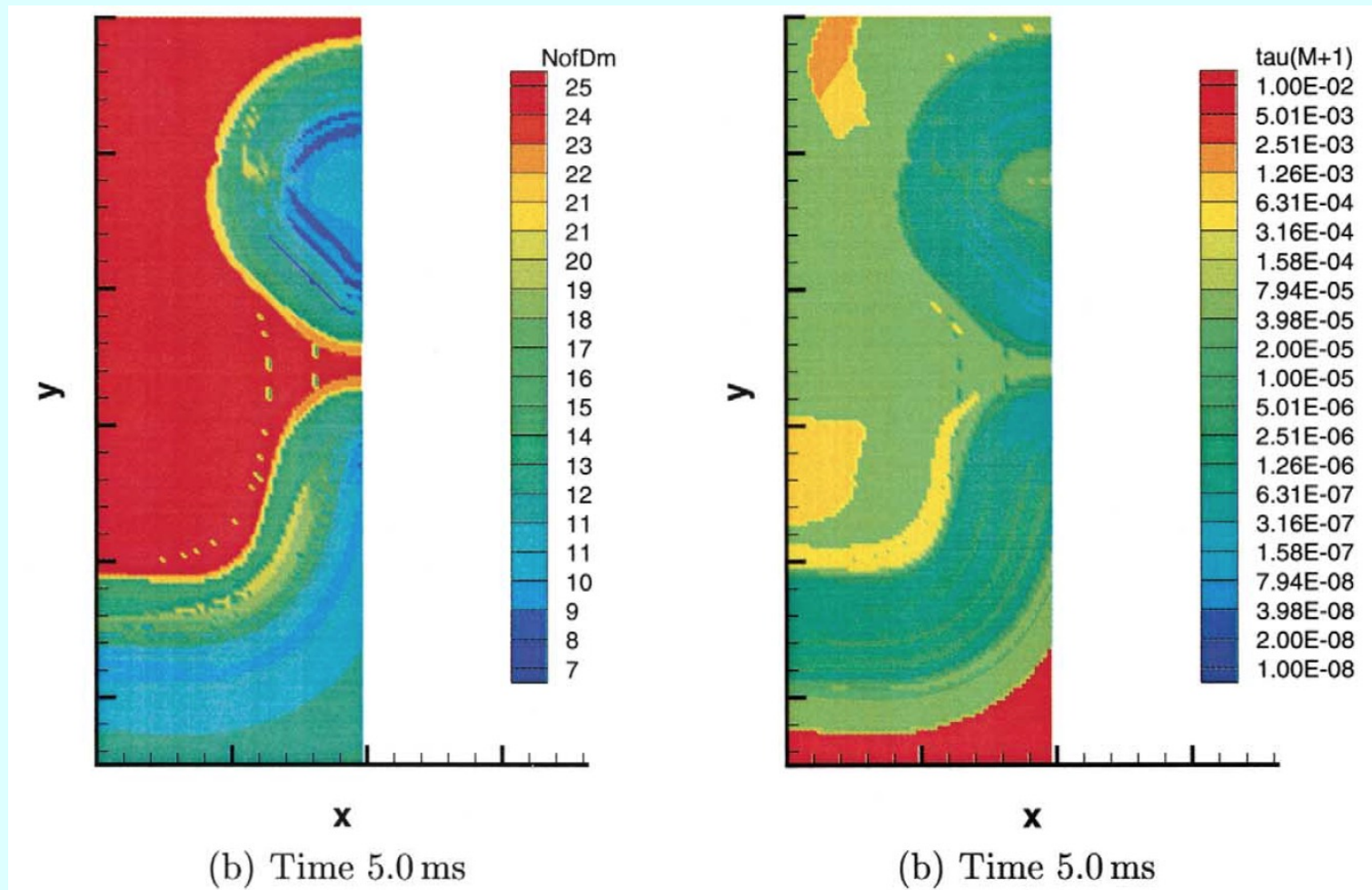
N=32
K=177



No. of fast time scales

Characteristic time scale

Number of fast time scales transient CH₄/air flame-vortex interaction



No. of fast time scales

Characteristic time scale

The issue of transport

$$\frac{dy}{dt} = \mathbf{S}_1 R^1 + \dots + \mathbf{S}_K R^K + \mathbf{L}(\mathbf{y})$$

$$= \mathbf{a}_1 f^1 + \dots + \mathbf{a}_M f^M + \mathbf{a}_{M+1} f^{M+1} + \dots + \mathbf{a}_N f^N$$

$$\underbrace{\tau_1 \quad \dots \quad \tau_M}_{\text{fast}} \ll \underbrace{\tau_{M+1} \quad \dots \quad \tau_N}_{\text{slow}}$$

$$f^i = \mathbf{b}^i \cdot (\mathbf{S}_1 R^1 + \dots + \mathbf{S}_K R^K + \mathbf{L}(\mathbf{y}))$$

$$\left. \begin{array}{ll} \tau_i & \text{chemical time scales} \\ \tau_{trans} & \text{transport time scale} \end{array} \right\} \tau_M \ll \tau_{M+1} \quad \tau_M \ll \tau_{trans}$$

$$f^i = \mathbf{b}^i \cdot (\mathbf{S}_1 R^1 + \dots + \mathbf{S}_K R^K) + \cancel{\mathbf{b}^i \mathbf{L}(\mathbf{y})} \quad i=1, M$$

Reduced model:

$$f^1 = f^2 = \dots = f^M = 0$$

transport independent

$$\frac{dy}{dt} = \mathbf{a}_{M+1} f^{M+1} + \dots + \mathbf{a}_N f^N$$

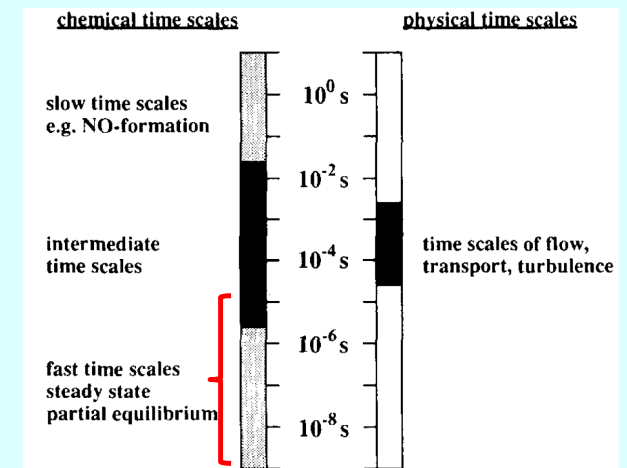


Fig. 1. Schematic illustration of the time scales governing a chemically reacting flow.



Various extensions of CSP; ILDM, REDIM, ISAT

1. Computational Singular Perturbation (CSP)
2. Intrinsic Low-Dimensional Manifolds (ILDM)
3. Reaction Diffusion Manifolds (REDIM)
4. In Situ Adaptive Tabulation (ISAT)

CSP: *Proc. Combust. Inst.*, 22:931-941 (1988), *Int. J. Chem. Kinet.*, 26:461-486 (1994)

ILDM: *Proc. Combust. Inst.*, 24:103-112 (1992), *Proc. Combust. Inst.*, 25:1349-1356 (1994)

REDIM: *Proc. Combust. Inst.*, 31:465-472 (2007), *Proc. Combust. Inst.*, 34:197-203 (2013)

ISAT: *Combust. Theory Model.*, 1:41-63 (1997), *J. Comp. Phys.*, 228:361-386 (2009)



ILDM

Full model:

$$\frac{dy}{dt} = \mathbf{S}_1 R^1 + \dots + \mathbf{S}_K R^K + \mathbf{L}(\mathbf{y})$$

$$= \mathbf{a}_1 f^1 + \dots + \mathbf{a}_M f^M + \mathbf{a}_{M+1} f^{M+1} + \dots + \mathbf{a}_N f^N$$

$$f^i = \mathbf{b}^i \cdot (\mathbf{S}_1 R^1 + \dots + \mathbf{S}_K R^K + \mathbf{L}(\mathbf{y}))$$

Reduced model:

$$f^1 = f^2 = \dots f^M = 0 \quad (\mathbf{b}^i \cdot \mathbf{L} = 0)$$

$$\frac{dy}{dt} = \mathbf{a}_{M+1} f^{M+1} + \dots + \mathbf{a}_N f^N$$

1. Use right and left eigenvectors for $\mathbf{a}_i$ and $\mathbf{b}^i$
2. Parametrize variables $\mathbf{y}_f = \mathbf{h}(\mathbf{y}_s)$
3. Tabulate $\dot{f}^i=0$ ($i=1,M$) relations
4. Gov. equations for the N-M slow variables only

Major drawback: M=fixed

Speed up by $O(10)$

Maas and Pope, Proc. CI, 1994

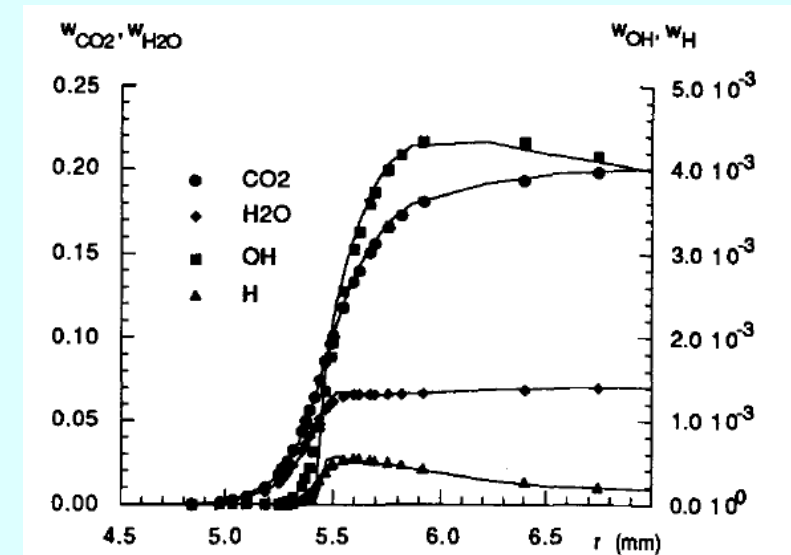


FIG. 4. Calculated structure of a syngas-air flame (points: reduced mechanism, lines: detailed mechanism).

REDIM

Full model:

$$\frac{dy}{dt} = \mathbf{S}_1 R^1 + \dots + \mathbf{S}_K R^K + \mathbf{L}(y)$$

$$= \mathbf{a}_1 f^1 + \dots + \mathbf{a}_M f^M + \mathbf{a}_{M+1} f^{M+1} + \dots + \mathbf{a}_N f^N$$

$$f^i = \mathbf{b}^i \cdot (\mathbf{S}_1 R^1 + \dots + \mathbf{S}_K R^K + \mathbf{L}(y))$$

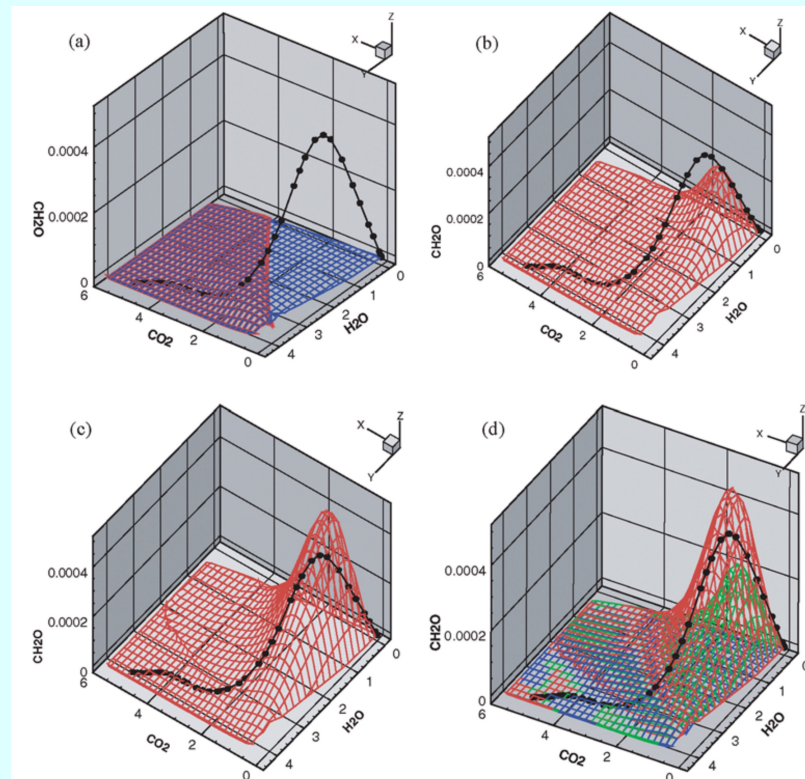
Reduced model:

$$f^1 = f^2 = \dots f^M = 0 \quad (\mathbf{b}^i \cdot \mathbf{L} = 0)$$

$$\frac{dy}{dt} = \mathbf{a}_{M+1} f^{M+1} + \dots + \mathbf{a}_N f^N$$

Issues to address:

1. $\mathbf{b}^i \cdot \mathbf{L} \neq 0$
2. Low temp domain



syngas/air

Figure 11. Two-dimensional relaxation, projection onto minor species. Manifolds: (a) extended ILDM; (b) after 10 iterations; (c) after 700 iterations; (d) combination of (a-c); black solid line is the detailed stationary solution.

Bykov and Maas
CTM 2007

ISAT

Main obstacle
in tabulation:

$D=10$ degrees of freedom
10 nodes in each direction

$10^D D = 10^{11}$ storage nodes

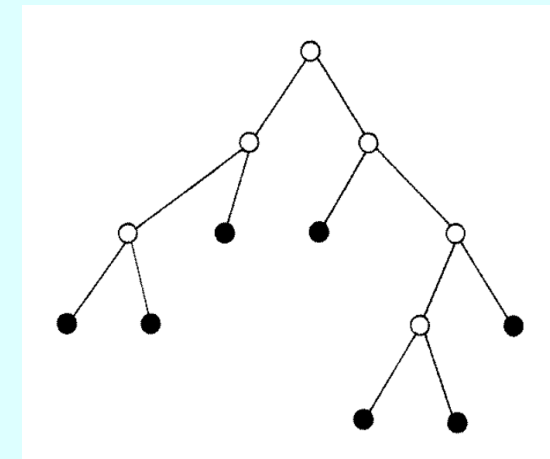
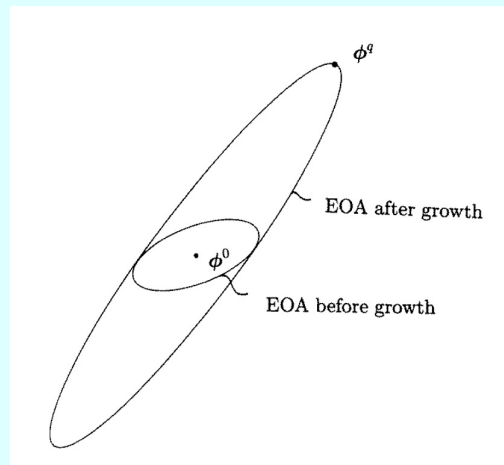
1 interpolation: $2^D D = 10^4$ op.

realizable
vs
accessed
region

accessed region:
low-dim. manifold

Table on accessed region:
Build up during the reactive
flow calculation

Extrapolation when inside of EOA
Integration when outside of EOA

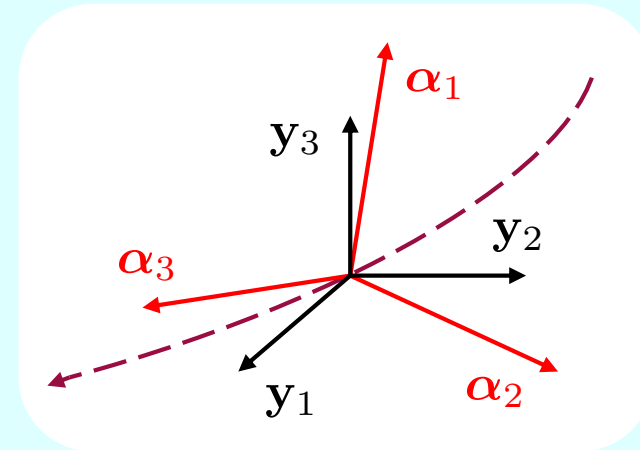
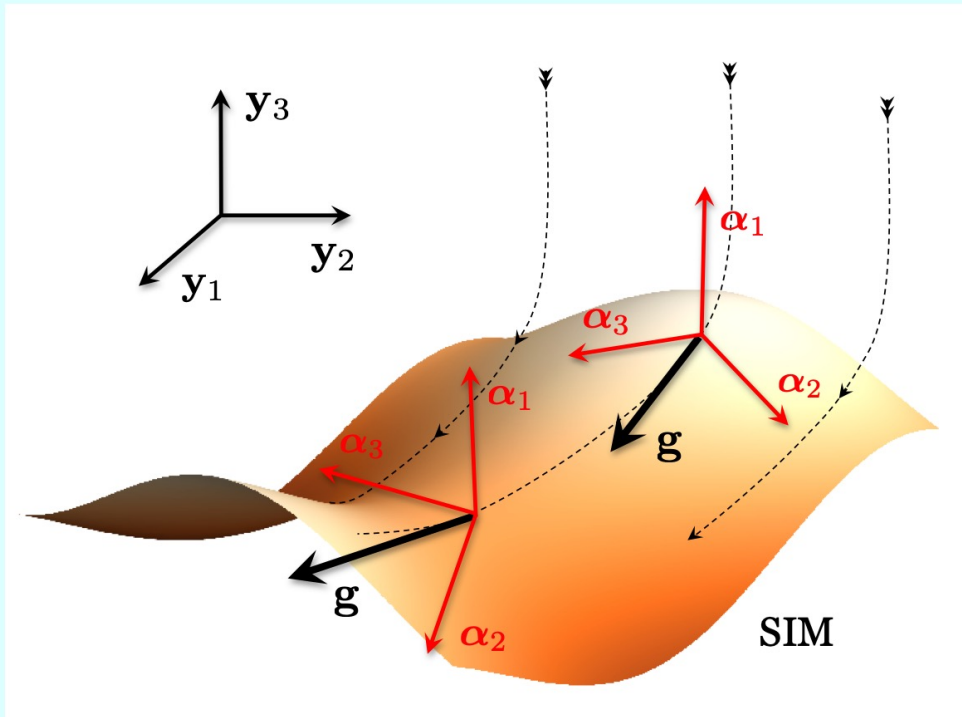


Speed up by $O(10^3)$

Pope, CST 1997

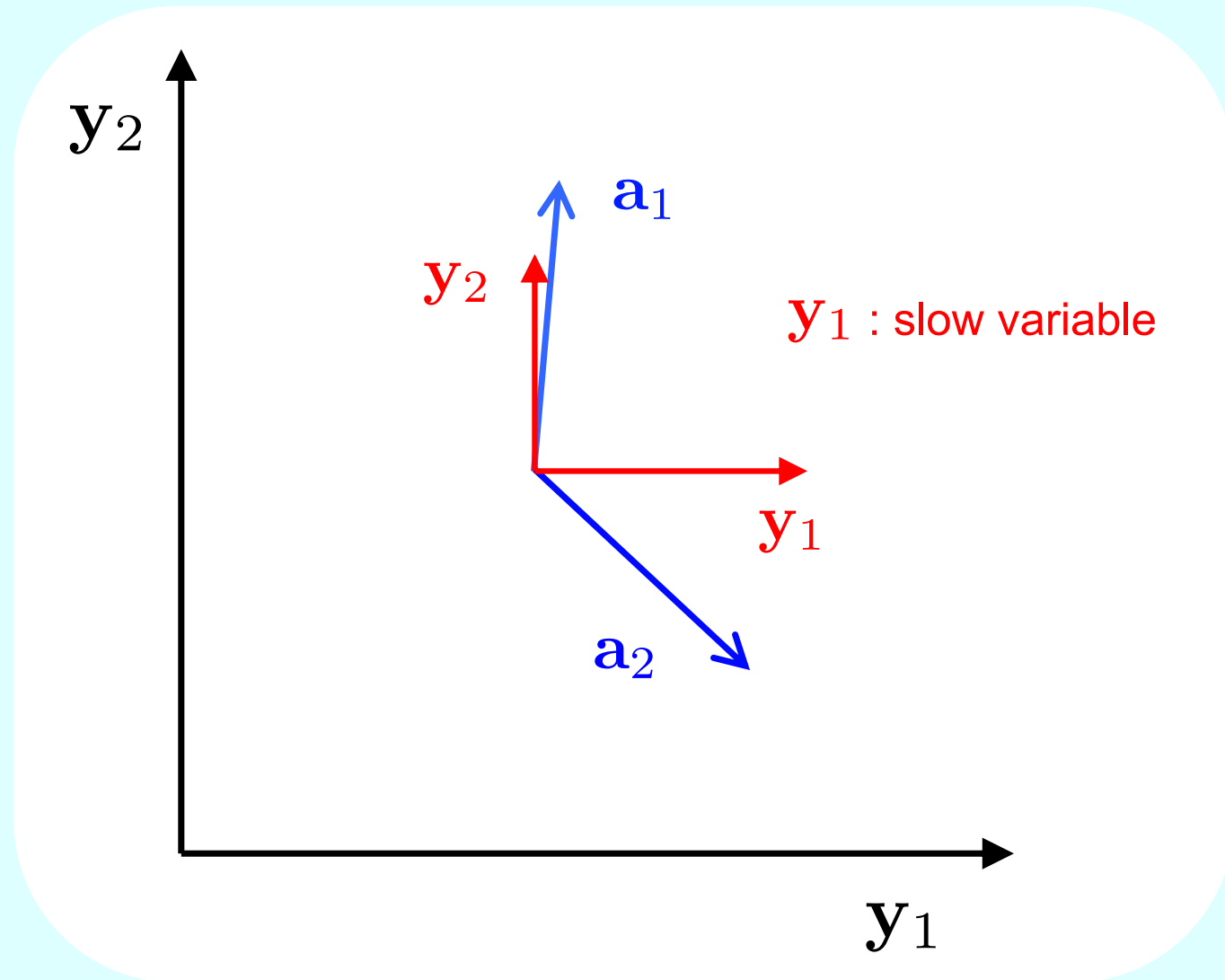


CSP Diagnostics

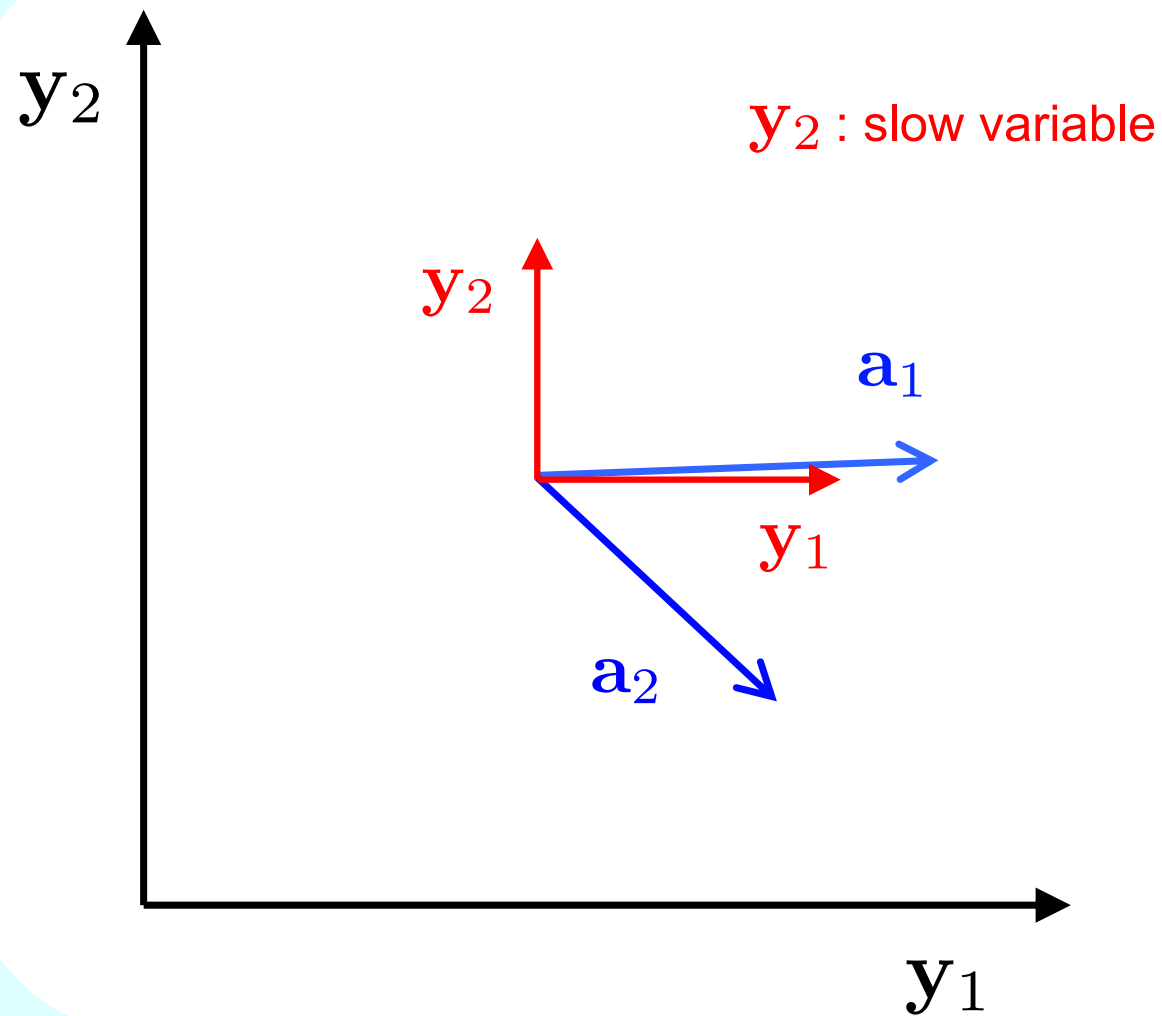


Which variables (species) relate the most to the evolution of the system ?

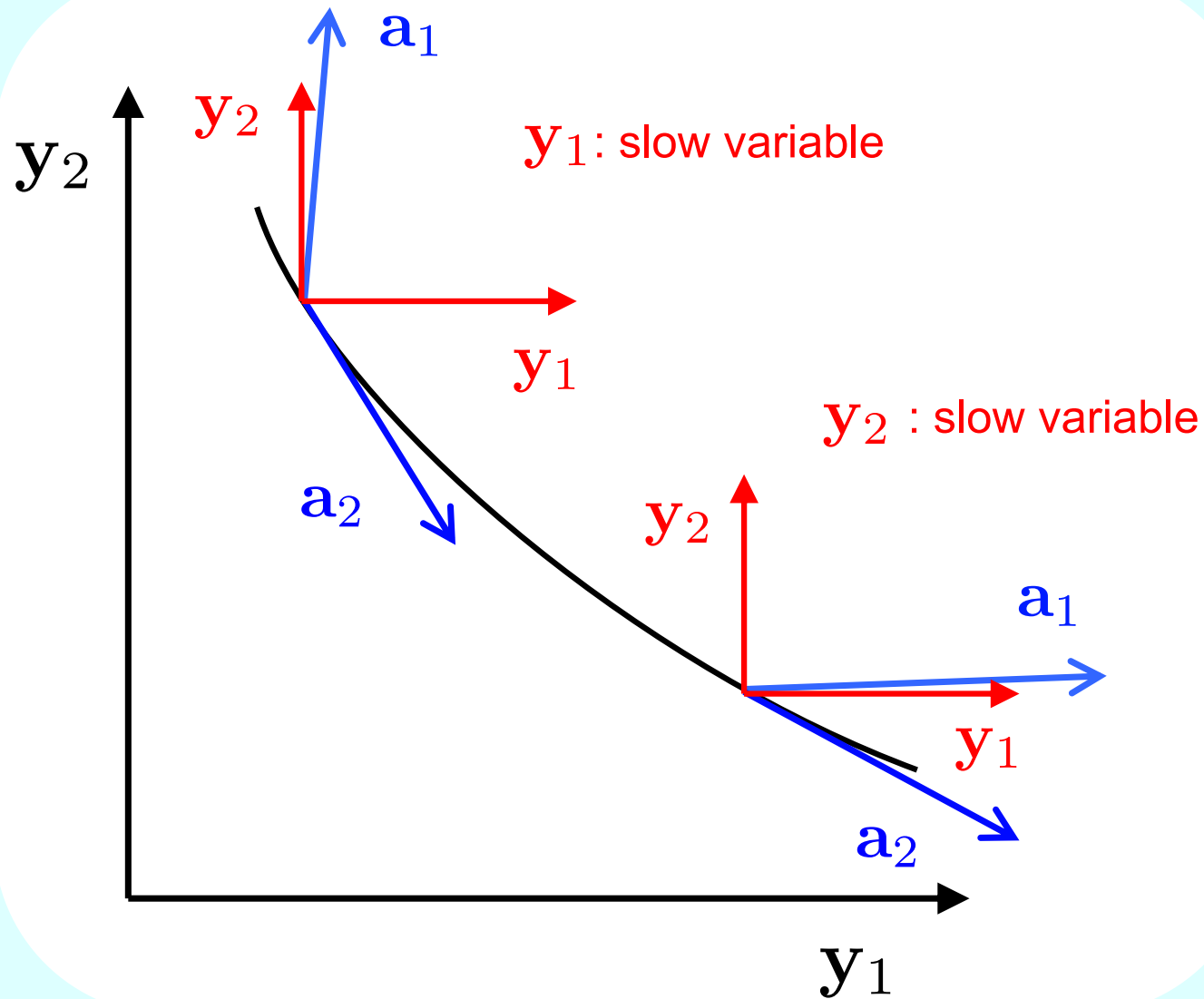
CSP Diagnostics: fast – slow variables



CSP Diagnostics: fast – slow variables



CSP Diagnostics: fast – slow variables



$$\mathbf{a}_1 = \begin{bmatrix} a_1^1 \\ a_1^2 \end{bmatrix}$$

$$\mathbf{b}^1 = \begin{bmatrix} b_1^1 & b_2^1 \end{bmatrix}$$

$$\mathbf{b}^1 \cdot \mathbf{a}_1 = 1$$

CSP Pointer

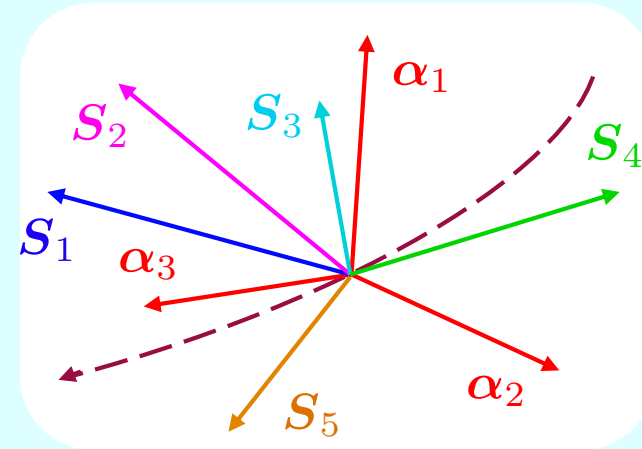
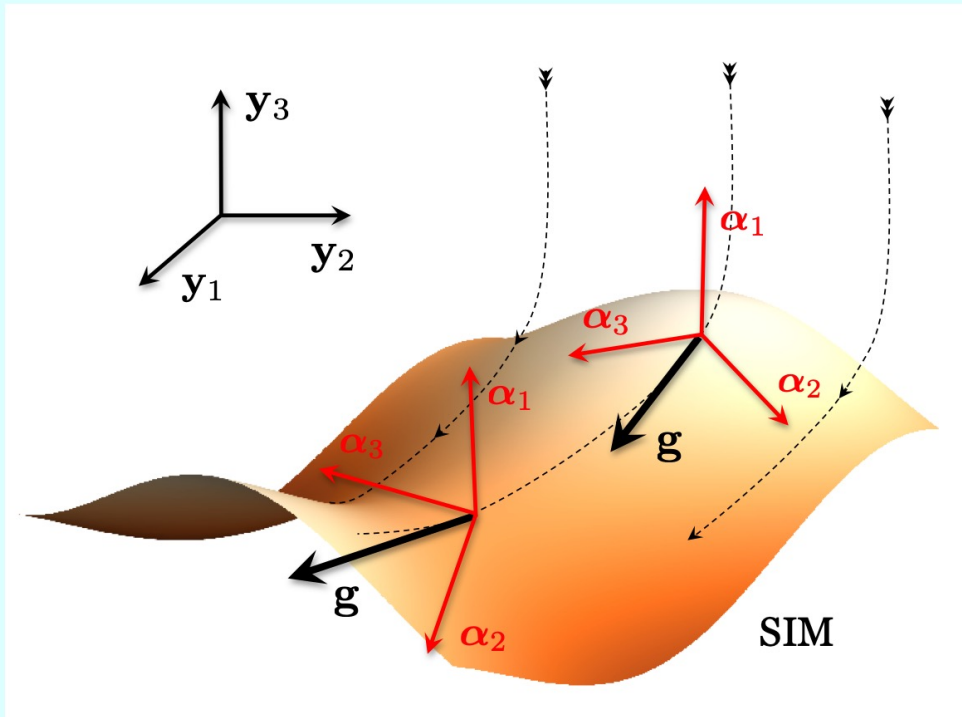
$$\mathbf{D}_1 = \begin{bmatrix} a_1^1 b_1^1 \\ a_1^2 b_2^1 \end{bmatrix}$$

$$a_1^1 b_1^1 + a_1^2 b_2^1 = 1$$

Manias et al, CNSNS 2025

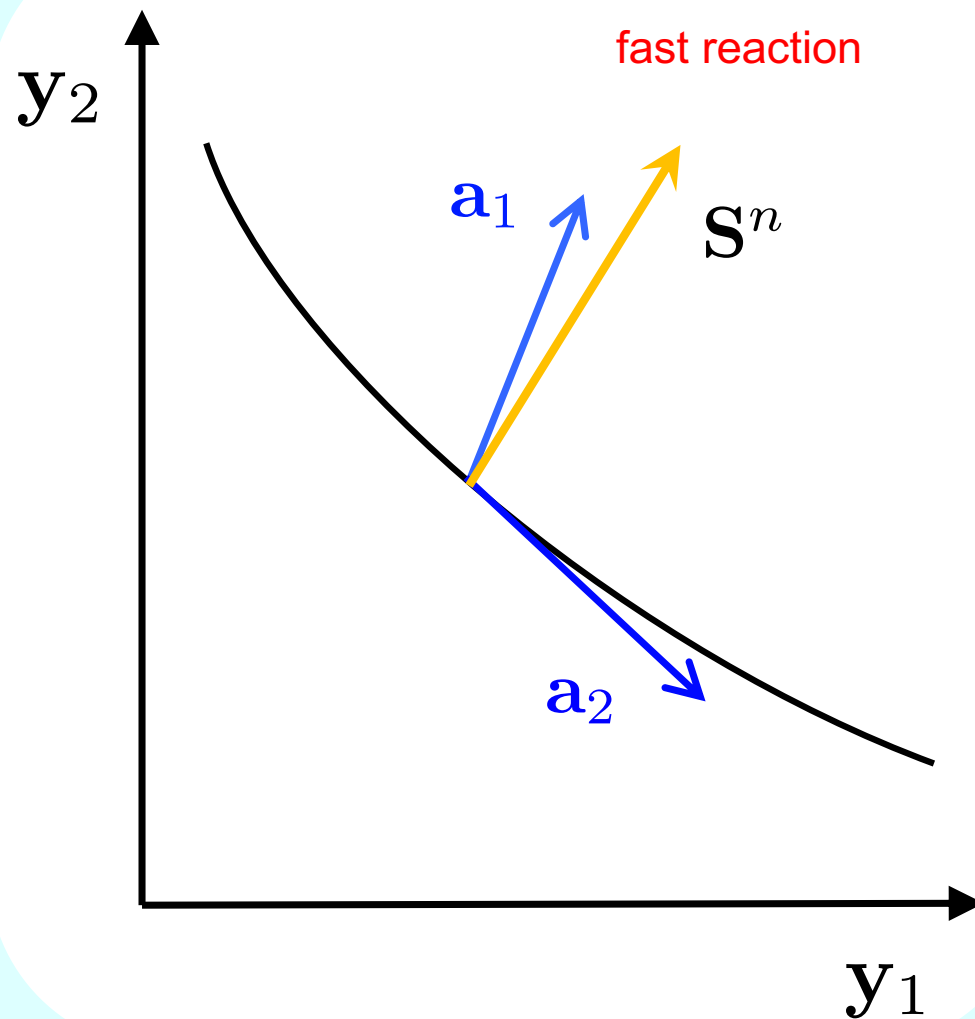


CSP Diagnostics



Which processes (reactions, transport) drive the system ?

CSP Diagnostics: fast – slow reactions



$$\frac{dy}{dt} = \mathbf{S}_1 R^1 + \cdots + \mathbf{S}_K R^K$$

$$\frac{dy}{dt} = \mathbf{a}_1 f^1 + \cdots + \mathbf{a}_N f^N$$

$$f^n = (\mathbf{b}^n \cdot \mathbf{S}_1) R^1 + \cdots + (\mathbf{b}^n \cdot \mathbf{S}_K) R^K$$

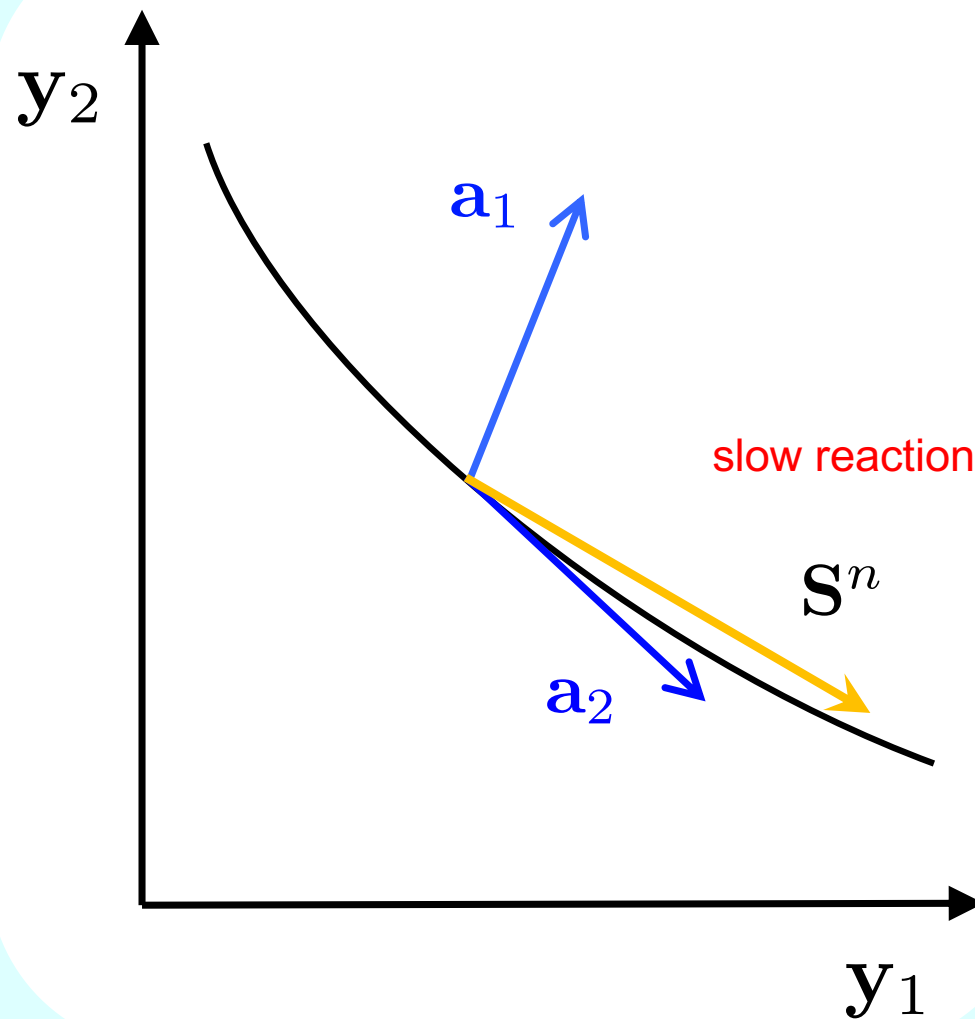
$$\mathbf{b}^i \cdot \mathbf{a}_j = \delta_j^i$$

$$\frac{dy}{dt} \approx \mathbf{a}_{M+1} f^{M+1} + \cdots + \mathbf{a}_N f^N$$

$$f^1, \cdots, f^M \approx 0$$



CSP Diagnostics: fast – slow reactions



$$\frac{dy}{dt} = \mathbf{S}_1 R^1 + \cdots + \mathbf{S}_K R^K$$

$$\frac{dy}{dt} = \mathbf{a}_1 f^1 + \cdots + \mathbf{a}_N f^N$$

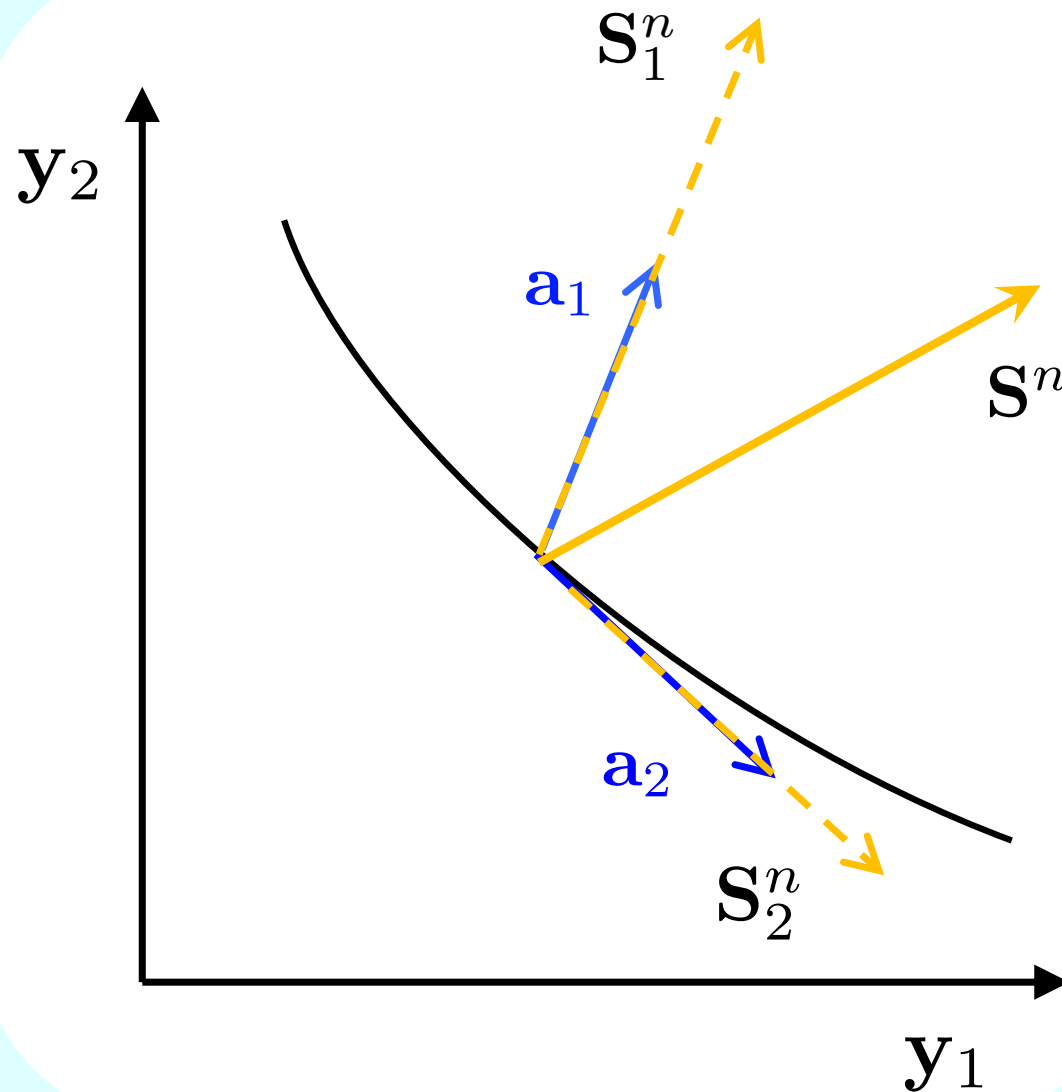
$$f^n = (\mathbf{b}^n \cdot \mathbf{S}_1) R^1 + \cdots + (\mathbf{b}^n \cdot \mathbf{S}_K) R^K$$

$$\mathbf{b}^i \cdot \mathbf{a}_j = \delta_j^i$$

$$\frac{dy}{dt} \approx \mathbf{a}_{M+1} f^{M+1} + \cdots + \mathbf{a}_N f^N$$

$$f^1, \dots, f^M \approx 0$$

CSP Diagnostics: fast – slow components of a reaction



$$S^n = S_1^n + S_2^n$$

On the manifold:

S_1^n has no influence

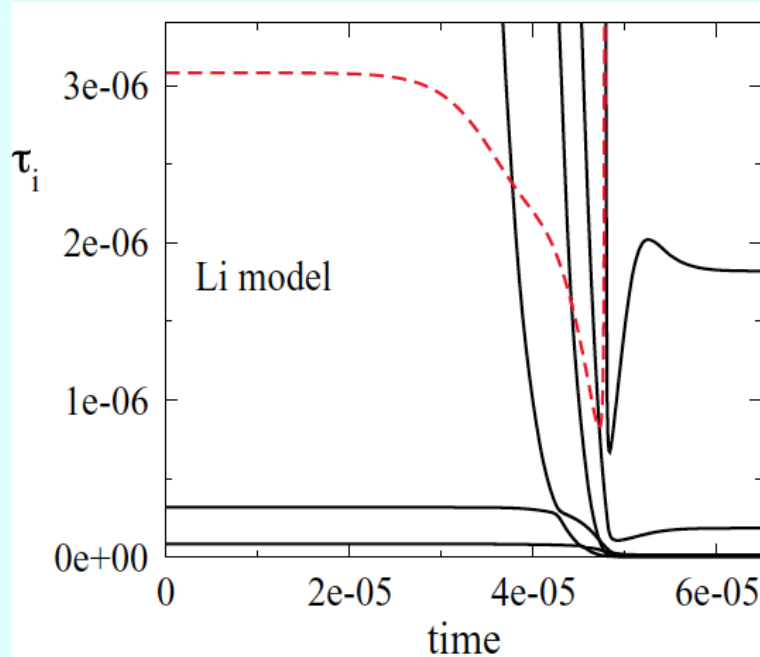
S_1^n } increase y_1
 S_2^n }

$S^n \rightarrow$ increases y_2

$S_2^n \rightarrow$ decreases y_2

Michalaki et al, J Math Biol, 2018

CSP diagnostics: explosive mode in H₂/air autoignition

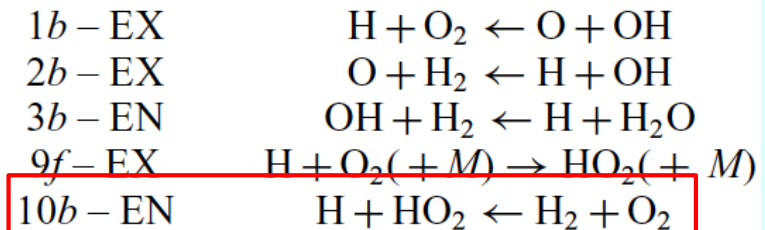
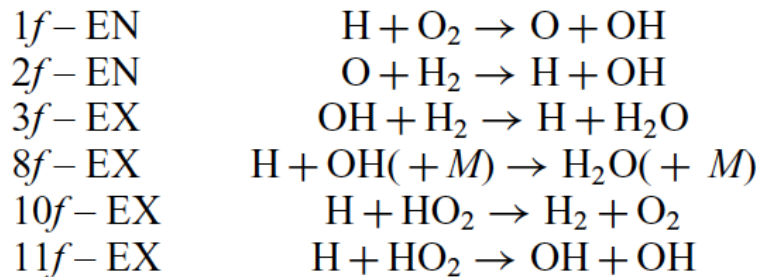


	$t = 0.0 \times 10^{-6} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.245 \times 10^5 \text{ s}^{-1}$	$t = 3.0 \times 10^{-6} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.245 \times 10^5 \text{ s}^{-1}$
TPI	1f: + 0.6273 2f: + 0.0739 3f: + 0.0183 9f: - 0.2792	1f: + 0.6273 2f: + 0.0739 3f: + 0.0183 9f: - 0.2792
API	10b: + 1.0000	1f: + 0.4987 10b: + 0.2100 2f: + 0.0543 3f: + 0.0138 9f: - 0.2220
Po	H: + 0.79 O: + 0.17 OH: + 0.04	H: + 0.79 O: + 0.17 OH: + 0.04

time scale

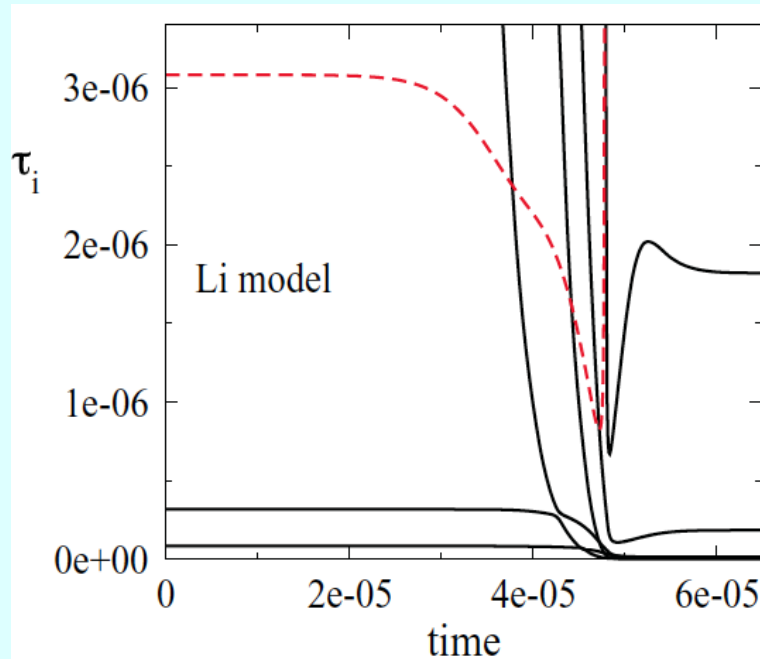
amplitude

variables



Diamantis et al, CTM 2015

CSP diagnostics: explosive mode in H₂/air autoignition

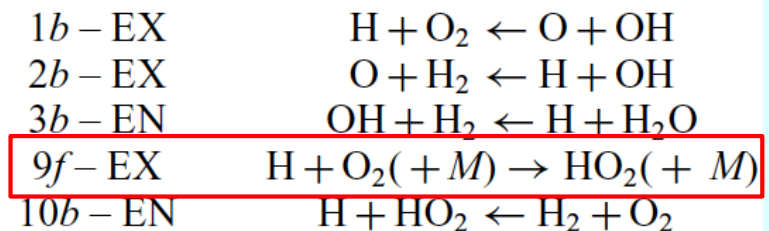
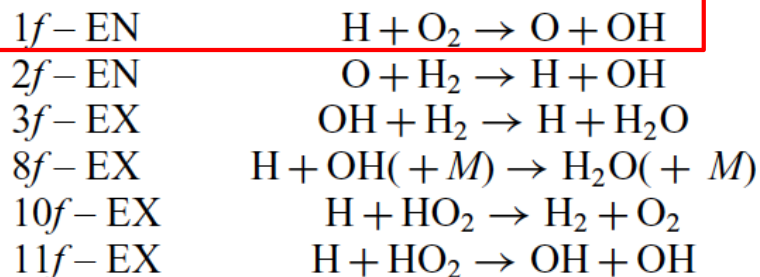


	$t = 0.0 \times 10^{-6} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.245 \times 10^5 \text{ s}^{-1}$	$t = 3.0 \times 10^{-6} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.245 \times 10^5 \text{ s}^{-1}$
TPI	1f: + 0.6273 2f: + 0.0739 3f: + 0.0183 9f: - 0.2792	1f: + 0.6273 2f: + 0.0739 3f: + 0.0183 9f: - 0.2792
API	10b: + 1.0000	1f: + 0.4987 10b: + 0.2100 2f: + 0.0543 3f: + 0.0138 9f: - 0.2220
Po	H: + 0.79 O: + 0.17 OH: + 0.04	H: + 0.79 O: + 0.17 OH: + 0.04

time scale

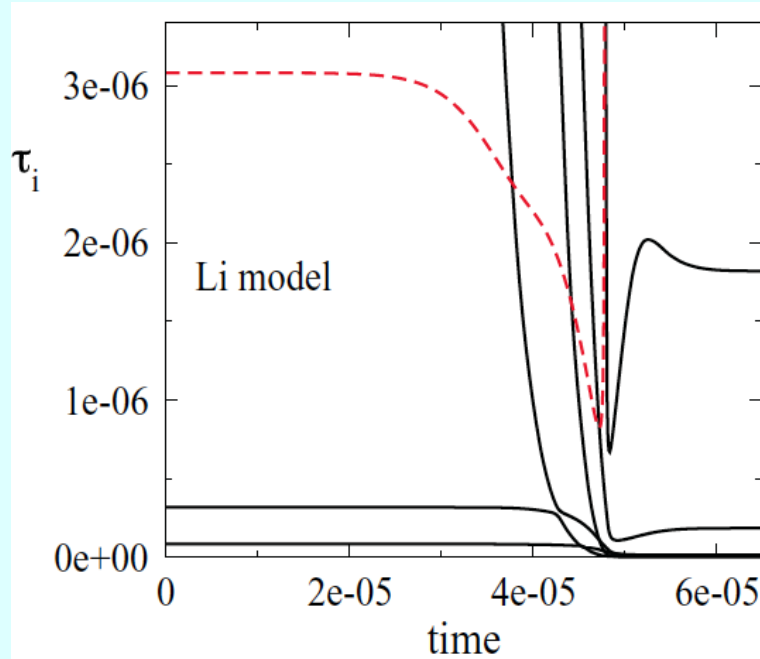
reactions

variables



Diamantis et al, CTM 2015

CSP diagnostics: explosive mode in H₂/air autoignition



	$t = 0.0 \times 10^{-6} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.245 \times 10^5 \text{ s}^{-1}$	$t = 3.0 \times 10^{-6} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.245 \times 10^5 \text{ s}^{-1}$
TPI	1f: + 0.6273 2f: + 0.0739 3f: + 0.0183 9f: - 0.2792	1f: + 0.6273 2f: + 0.0739 3f: + 0.0183 9f: - 0.2792
API	10b: + 1.0000	1f: + 0.4987 10b: + 0.2100 2f: + 0.0543 3f: + 0.0138 9f: - 0.2220
Po	H: + 0.79 O: + 0.17 OH: + 0.04	H: + 0.79 O: + 0.17 OH: + 0.04

time scale

reactions

variables

1f-EN
2f-EN
3f-EX
8f-EX
10f-EX
11f-EX

$\text{H} + \text{O}_2 \rightarrow \text{O} + \text{OH}$
 $\text{O} + \text{H}_2 \rightarrow \text{H} + \text{OH}$
 $\text{OH} + \text{H}_2 \rightarrow \text{H} + \text{H}_2\text{O}$
 $\text{H} + \text{OH}(+M) \rightarrow \text{H}_2\text{O}(+M)$
 $\text{H} + \text{HO}_2 \rightarrow \text{H}_2 + \text{O}_2$
 $\text{H} + \text{HO}_2 \rightarrow \text{OH} + \text{OH}$

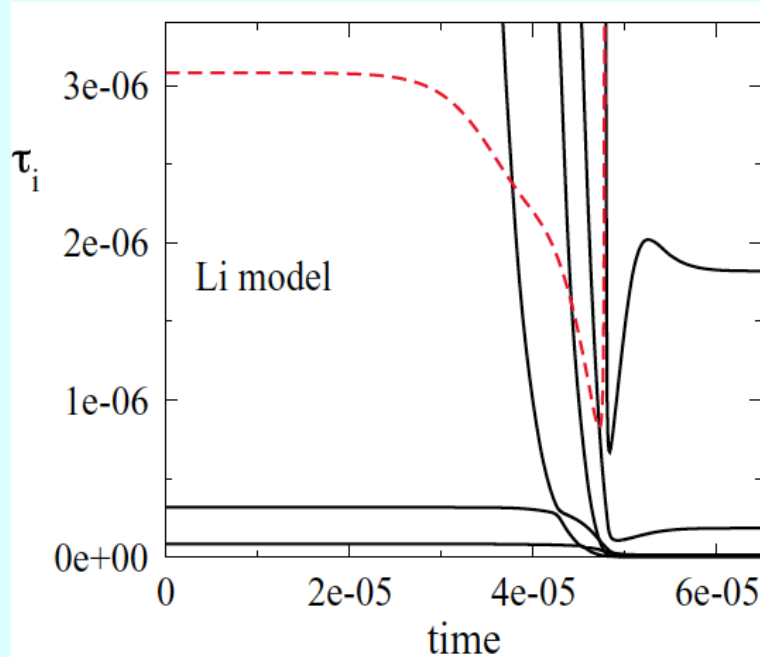
1b-EX
2b-EX
3b-EN
9f-EX
10b-EN

$\text{H} + \text{O}_2 \leftarrow \text{O} + \text{OH}$
 $\text{O} + \text{H}_2 \leftarrow \text{H} + \text{OH}$
 $\text{OH} + \text{H}_2 \leftarrow \text{H} + \text{H}_2\text{O}$
 $\text{H} + \text{O}_2(+M) \rightarrow \text{HO}_2(+M)$
 $\text{H} + \text{HO}_2 \leftarrow \text{H}_2 + \text{O}_2$

Diamantis et al, CTM 2015



CSP diagnostics: explosive mode in H₂/air autoignition



	$t = 0.0 \times 10^{-6} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.245 \times 10^5 \text{ s}^{-1}$
TPI	$1f: + 0.6273$ $2f: + 0.0739$ $3f: + 0.0183$ $9f: - 0.2792$
API	$10b: + 1.0000$
Po	$\text{H}: + 0.79$ $\text{O}: + 0.17$ $\text{OH}: + 0.04$

	$t = 3.0 \times 10^{-6} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.245 \times 10^5 \text{ s}^{-1}$
	$1f: + 0.6273$ $2f: + 0.0739$ $3f: + 0.0183$ $9f: - 0.2792$
	$1f: + 0.4987$ $10b: + 0.2100$ $2f: + 0.0543$ $3f: + 0.0138$ $9f: - 0.2220$
	$\text{H}: + 0.79$ $\text{O}: + 0.17$ $\text{OH}: + 0.04$

time scale

reactions

variables

$1f - \text{EN}$
 $2f - \text{EN}$
 $3f - \text{EX}$
 $8f - \text{EX}$
 $10f - \text{EX}$
 $11f - \text{EX}$

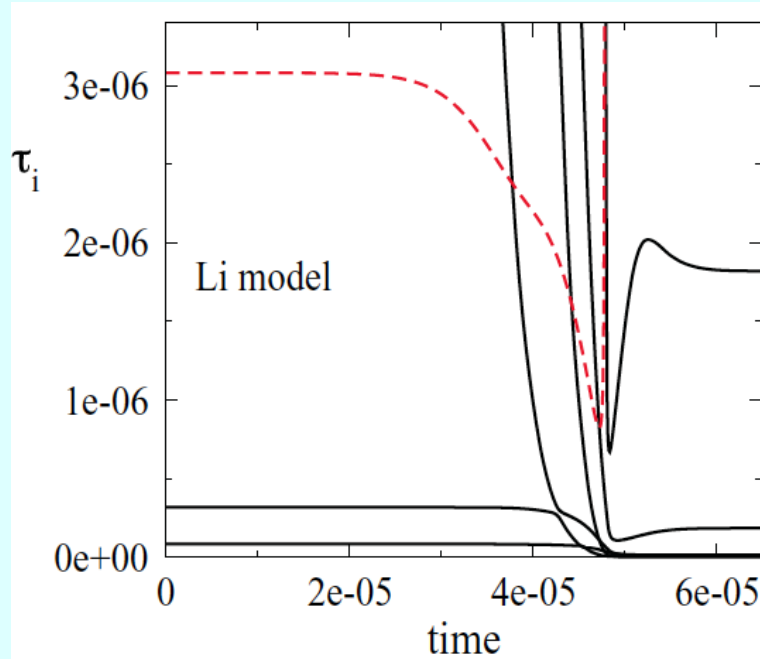
$\text{H} + \text{O}_2 \rightarrow \text{O} + \text{OH}$
 $\text{O} + \text{H}_2 \rightarrow \text{H} + \text{OH}$
 $\text{OH} + \text{H}_2 \rightarrow \text{H} + \text{H}_2\text{O}$
 $\text{H} + \text{OH}(+M) \rightarrow \text{H}_2\text{O}(+M)$
 $\text{H} + \text{HO}_2 \rightarrow \text{H}_2 + \text{O}_2$
 $\text{H} + \text{HO}_2 \rightarrow \text{OH} + \text{OH}$

$1b - \text{EX}$
 $2b - \text{EX}$
 $3b - \text{EN}$
 $9f - \text{EX}$
 $10b - \text{EN}$

$\text{H} + \text{O}_2 \leftarrow \text{O} + \text{OH}$
 $\text{O} + \text{H}_2 \leftarrow \text{H} + \text{OH}$
 $\text{OH} + \text{H}_2 \leftarrow \text{H} + \text{H}_2\text{O}$
 $\text{H} + \text{O}_2(+M) \rightarrow \text{HO}_2(+M)$
 $\text{H} + \text{HO}_2 \leftarrow \text{H}_2 + \text{O}_2$

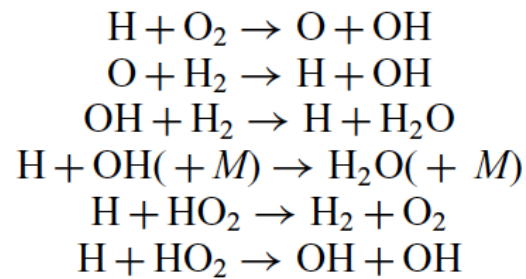
Diamantis et al, CTM 2015

CSP diagnostics: explosive mode in H₂/air autoignition

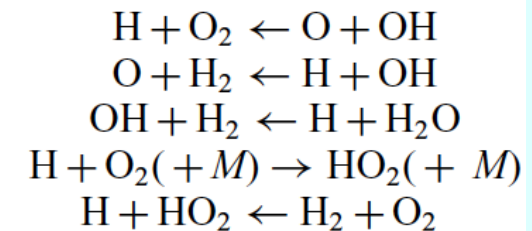


	$t = 2.000 \times 10^{-5} \text{ s}$ $T = 1100 \text{ K}$ $\lambda = 3.251 \times 10^5 \text{ s}^{-1}$	$t = 4.500 \times 10^{-5} \text{ s}$ $T = 1168 \text{ K}$ $\lambda = 7.014 \times 10^5 \text{ s}^{-1}$	$t = 4.782 \times 10^{-5} \text{ s}$ $T = 1593 \text{ K}$ $\lambda = 3.043 \times 10^5 \text{ s}^{-1}$
TPI	1f: + 0.6265 2f: + 0.0740 3f: + 0.0184 9f: - 0.2785	1f: + 0.5959 2f: + 0.1430 3f: + 0.0886 11f: + 0.0543 10f: - 0.0316 1b: - 0.0304 8f: - 0.0210	1f: + 0.2652 2f: + 0.1607 3f: + 0.0569 8f: + 0.0277 1b: - 0.1581 2b: - 0.1244 3b: - 0.0853
API	1f: + 0.6265 2f: + 0.0740 3f: + 0.0184 9f: - 0.2785	1f: + 0.5823 2f: + 0.1408 3f: + 0.0972 11f: + 0.0723 10f: - 0.0414 1b: - 0.0173 8f: - 0.0126	1f: + 0.3389 2f: + 0.1646 3f: + 0.1432 9f: + 0.0704 8f: + 0.0598 1b: - 0.0838 2b: - 0.0515 3b: - 0.0366
Po	H: + 0.79 O: + 0.17	H: + 0.65 T: + 0.19 O: + 0.13	T: + 3.50 O ₂ : -2.60 H ₂ O: -2.45

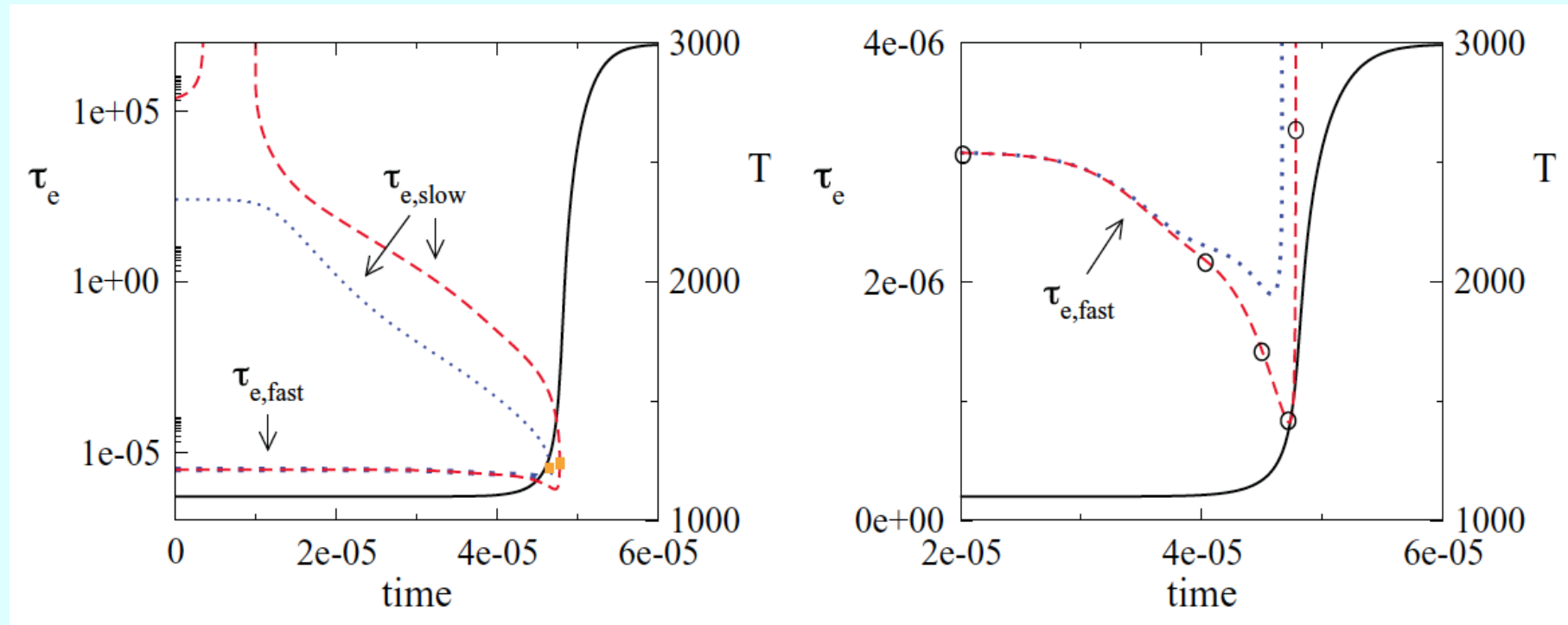
1f- EN
2f- EN
3f- EX
8f- EX
10f- EX
11f- EX



1b- EX
2b- EX
3b- EN
9f- EX
10b- EN



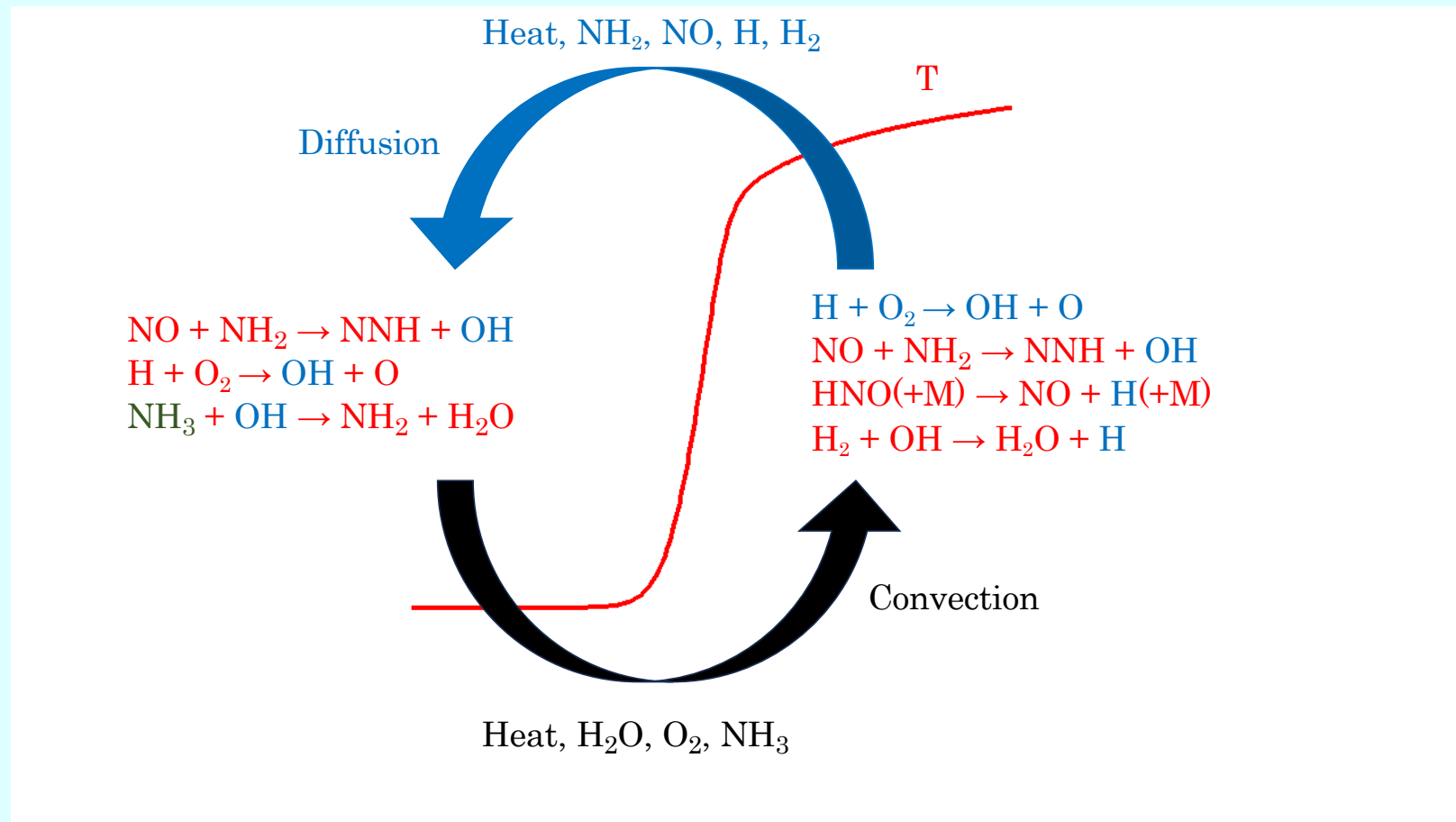
CSP diagnostics: explosive mode in H_2 /air autoignition



Chemical vs thermal runaway regime



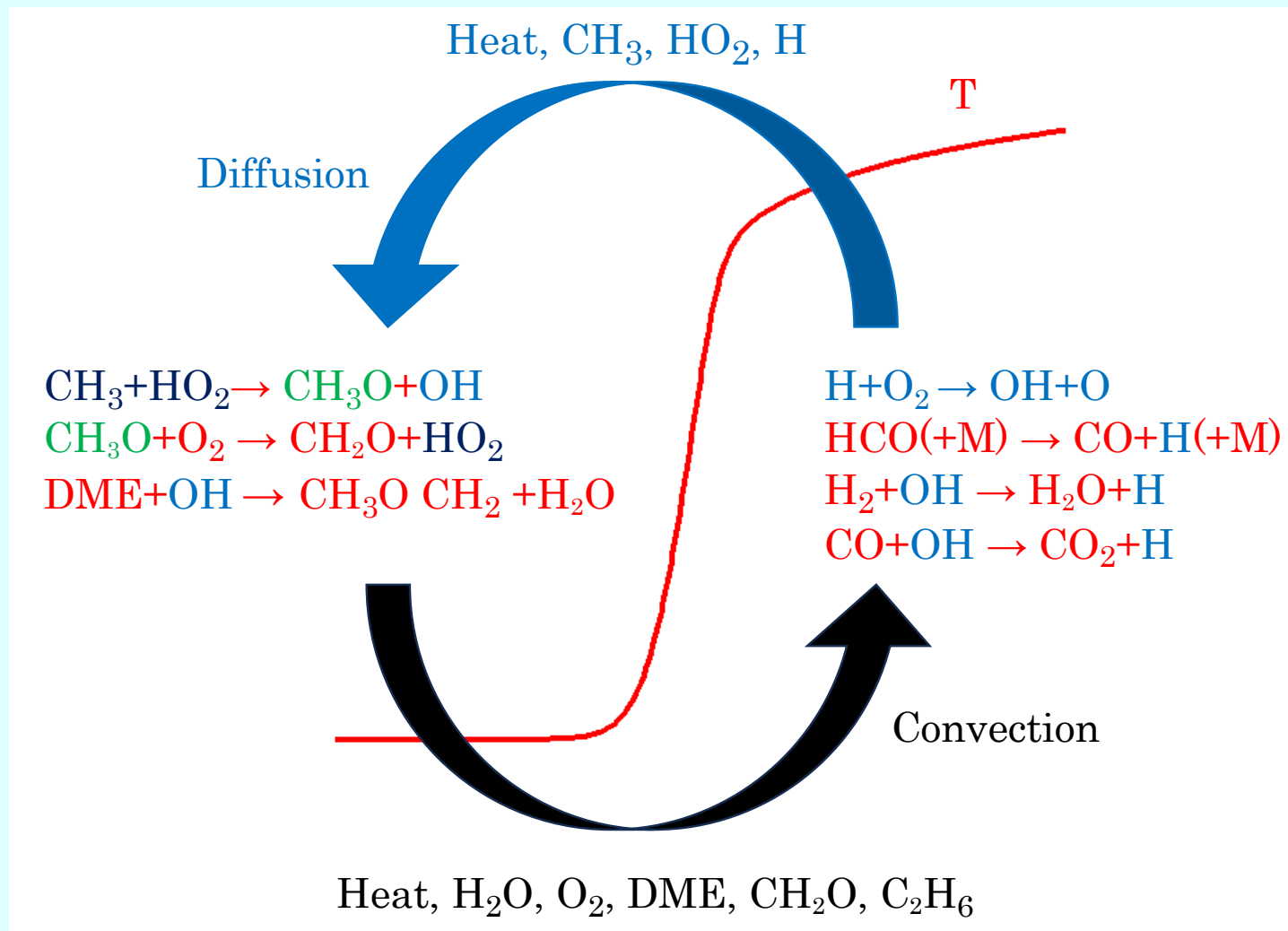
Propagation mechanism of NH₃/air premixed flame



Tingas et al, IJHE 2024



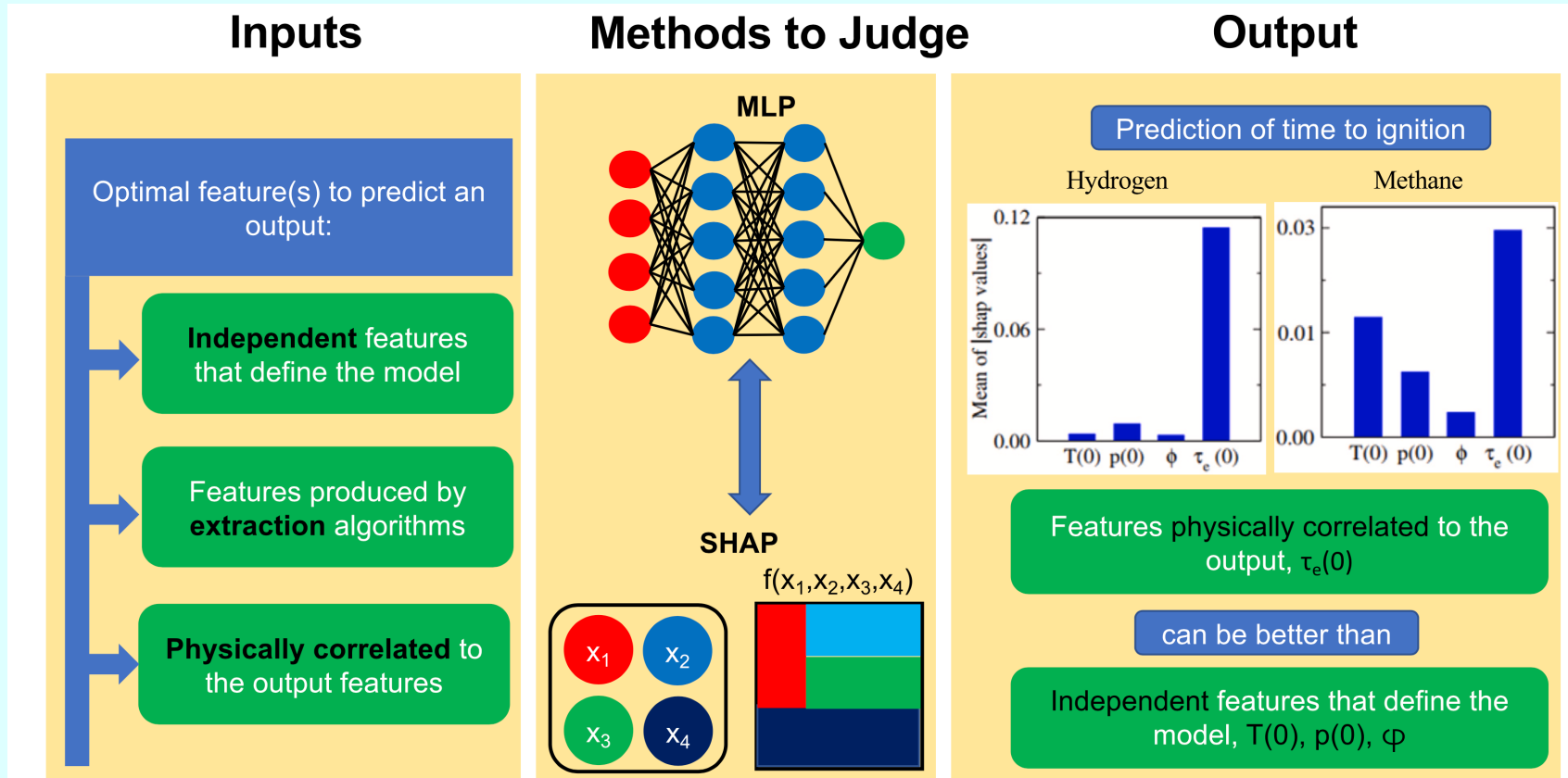
Propagation mechanism of DME₃/air premixed flame



Radaideh et al, submitted



Optimal input in Machine Learning algorithms



HIGHLIGHTS

- Inputs to ML algorithms must convey the physics of the process.
- Time scale(s) are frequently employed to characterize the dynamics.
- The characteristic time is shown an excellent input.

Radaideh et al, Energy and AI, 2023



Optimal input in Machine Learning algorithms

Homogeneous autoignition
of H₂/air CH₄/air mixtures

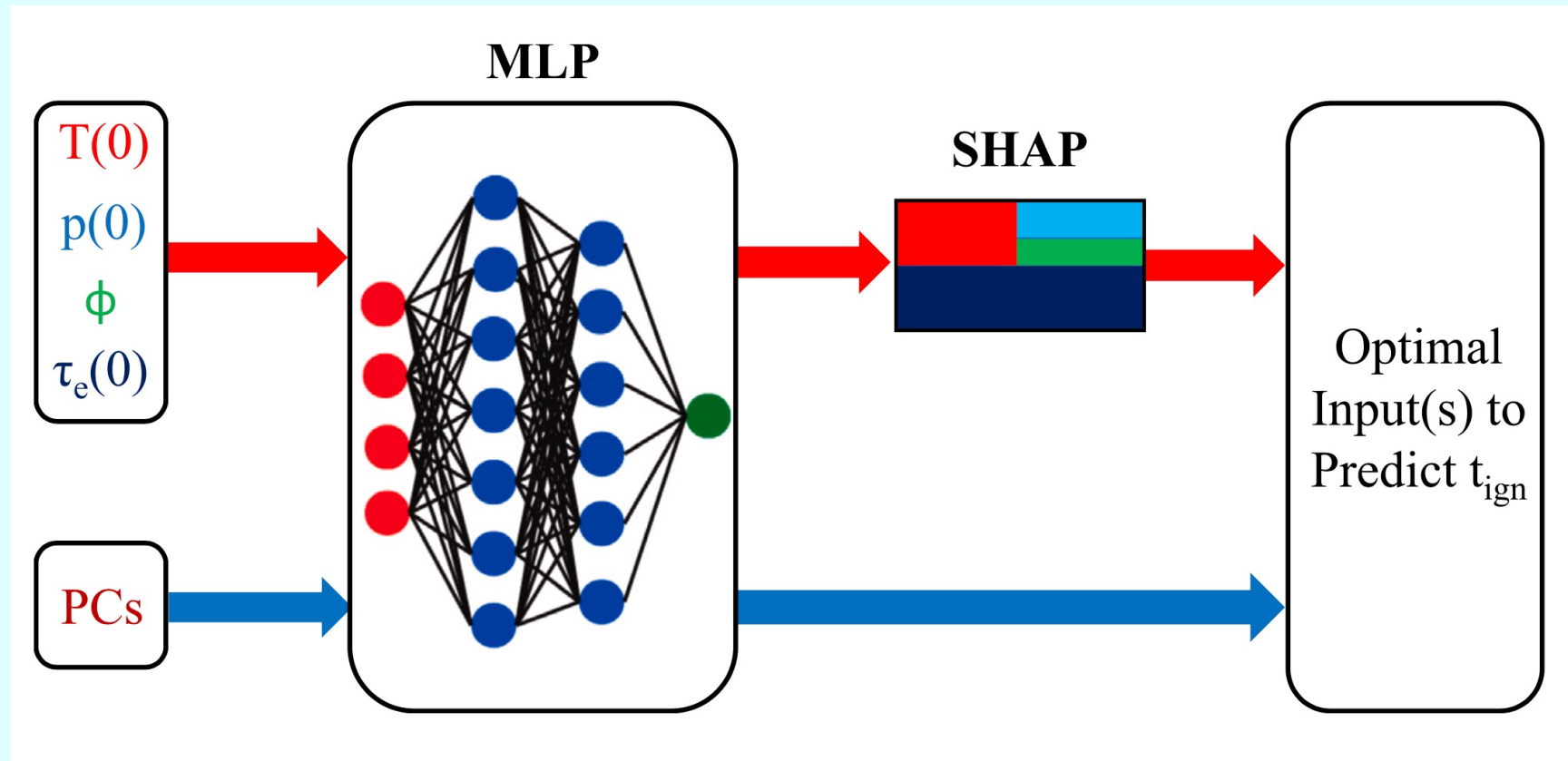
System fully specified when
T(0), p(0) and ϕ are known

Typical entries in the dataset for H₂/air mixtures.

$T(0)$ [K]	$p(0)$ [atm]	ϕ	$\tau_e(0)$ [s]	t_{ign} [s]
1100.00	1.00	0.50	4.94203973E-06	8.47149668E-05
1180.00	11.00	0.95	5.48137994E-06	3.50005334E-05
1260.00	7.00	1.70	5.23293601E-07	6.65079873E-06
:	:	:	:	:



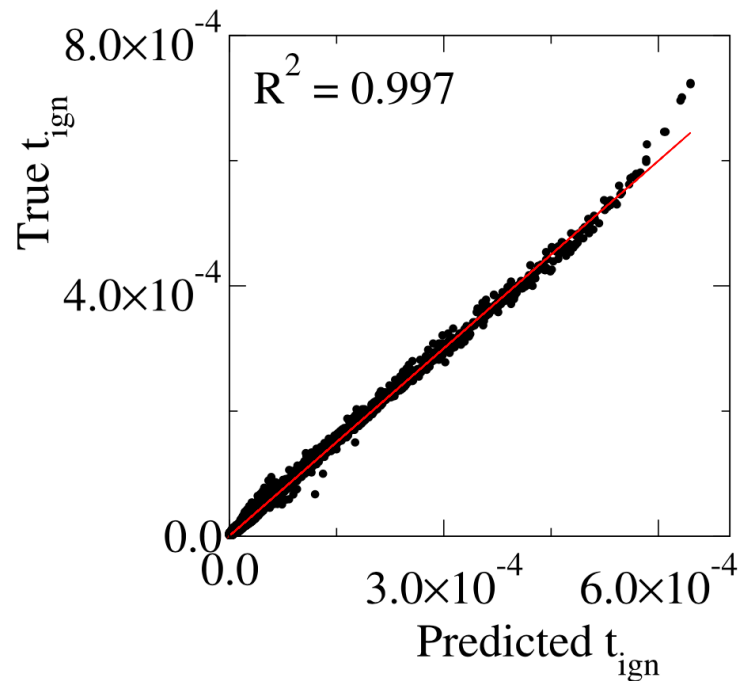
Optimal input in Machine Learning algorithms



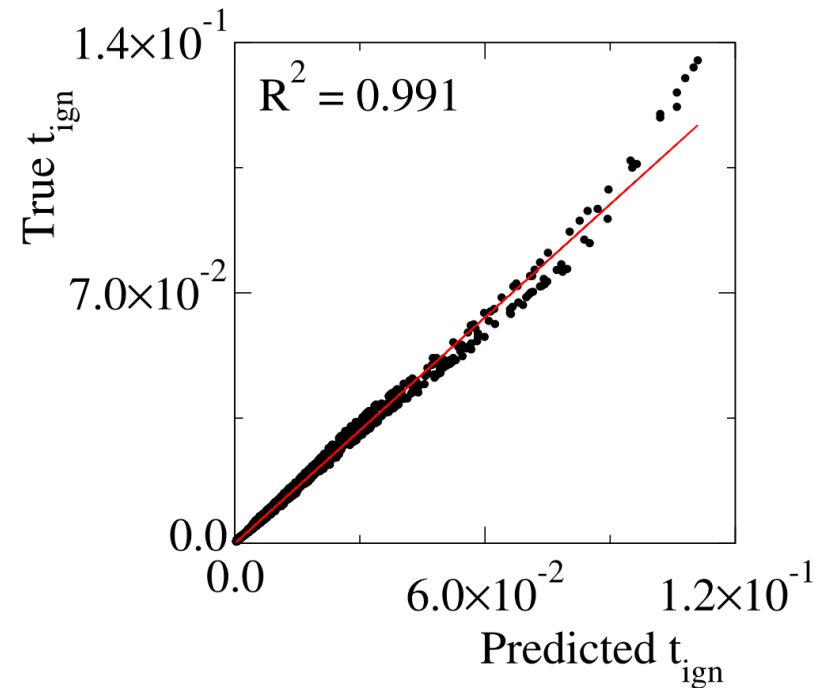
MLP: Multilayer Perceptron
SHAP: SHapley Additive exPlanations
PC: Principal Component

Optimal input in Machine Learning algorithms

H₂/air



CH₄/air

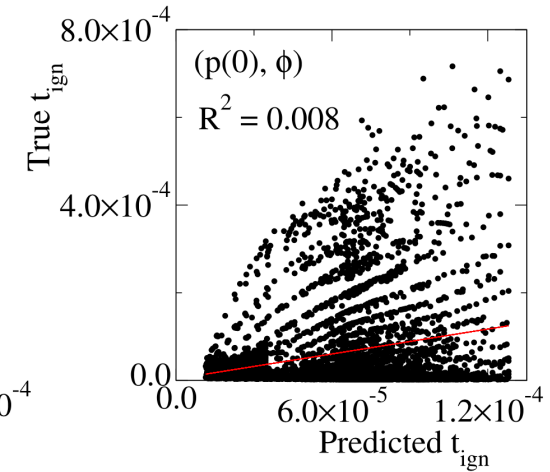
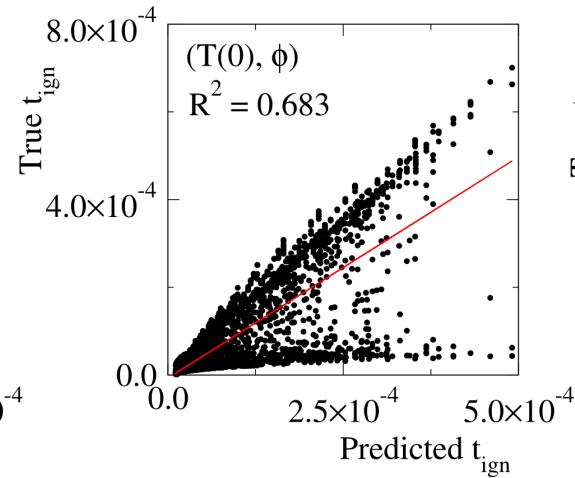
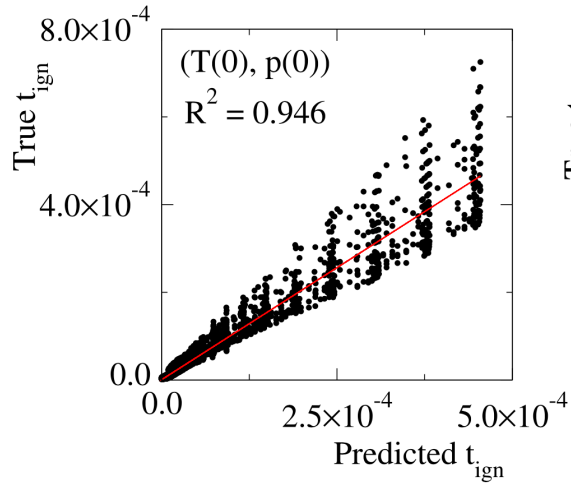


Input: $T(0)$, $p(0)$, ϕ

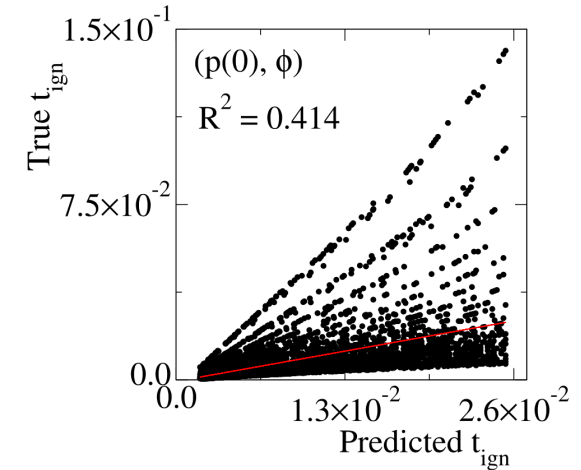
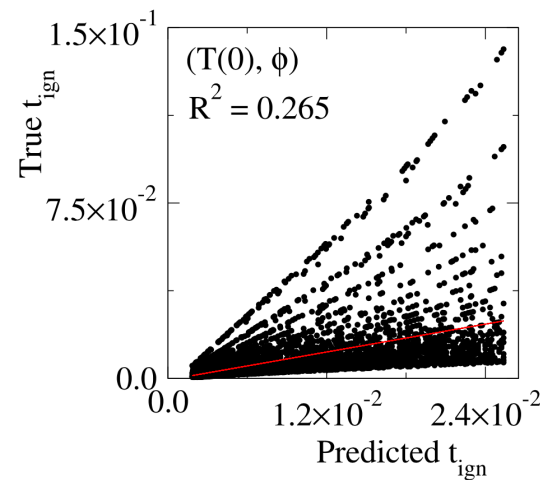
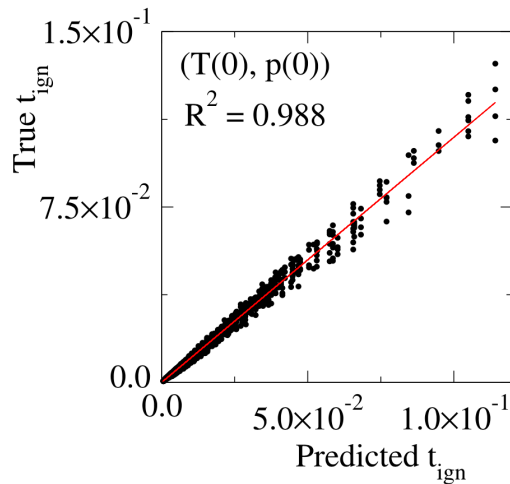


Optimal input in Machine Learning algorithms

H₂/air



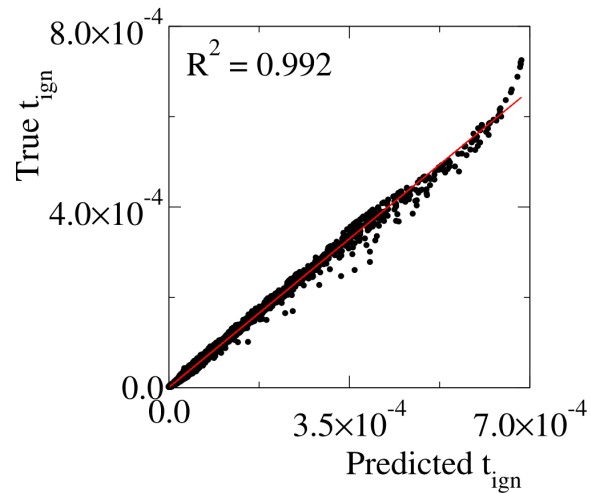
CH₄/air



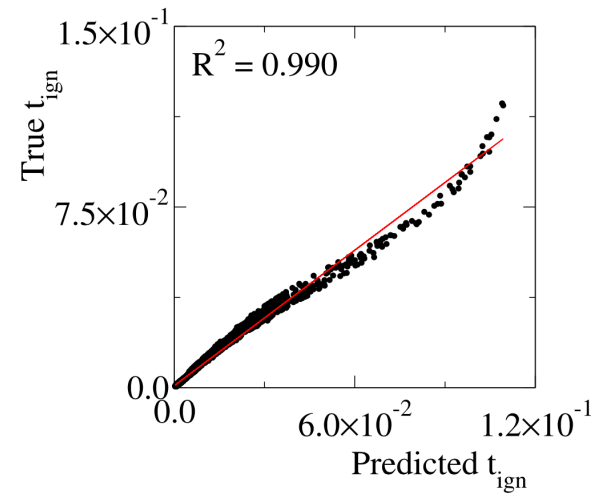
Optimal input in Machine Learning algorithms

3 PCs

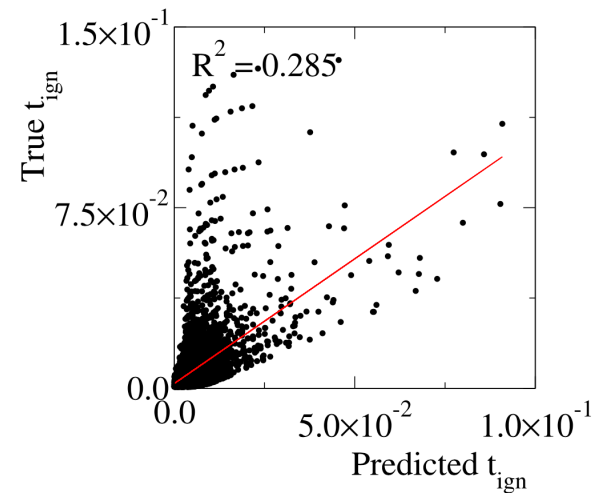
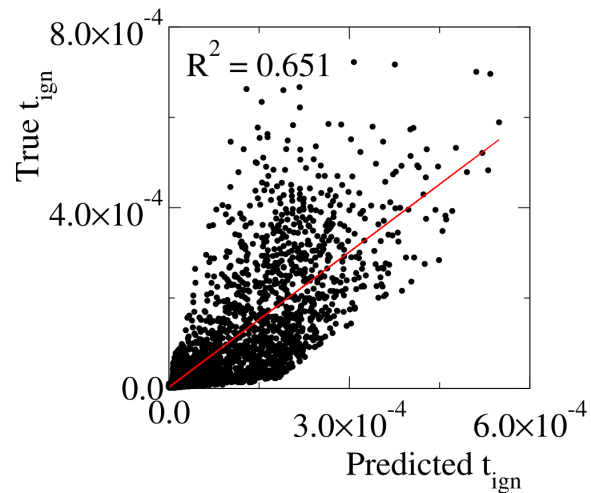
H₂/air



CH₄/air

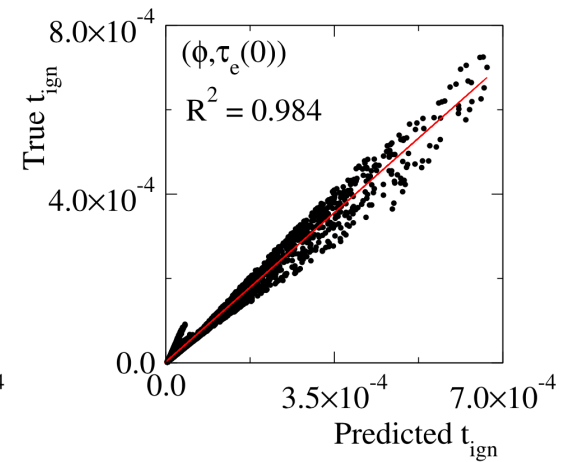
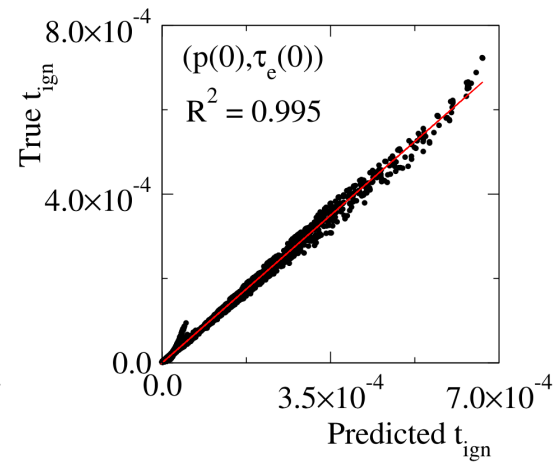
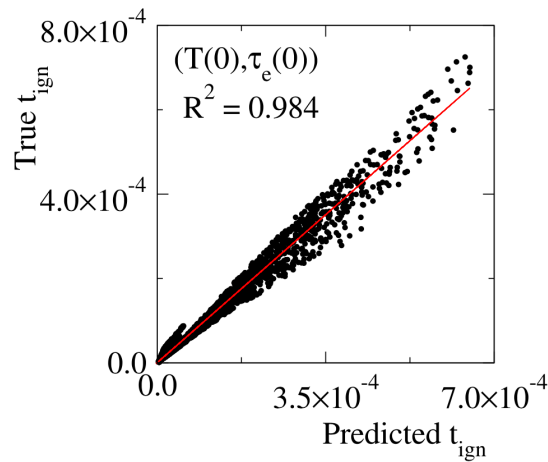


2 PCs

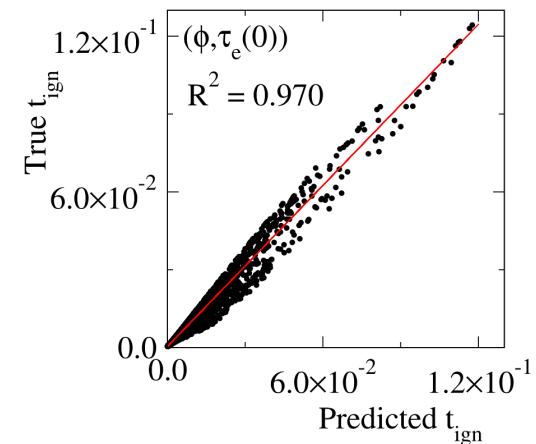
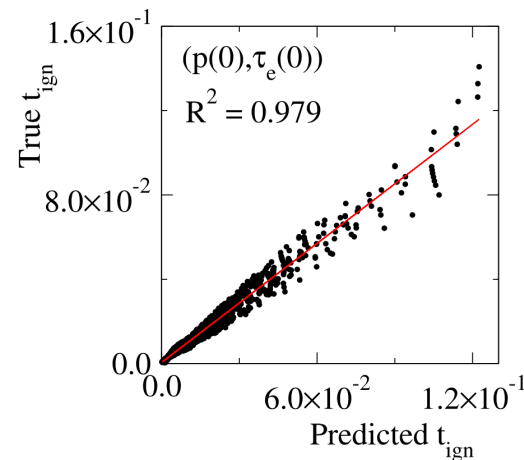
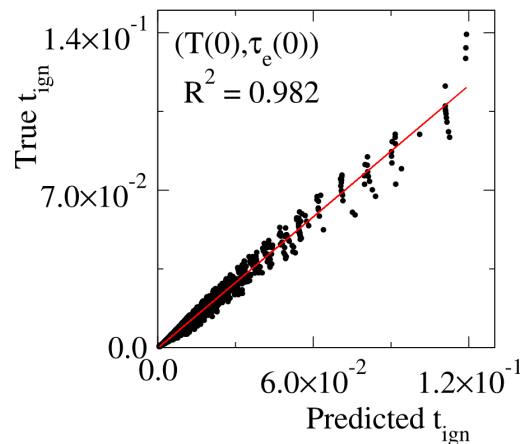


Optimal input in Machine Learning algorithms

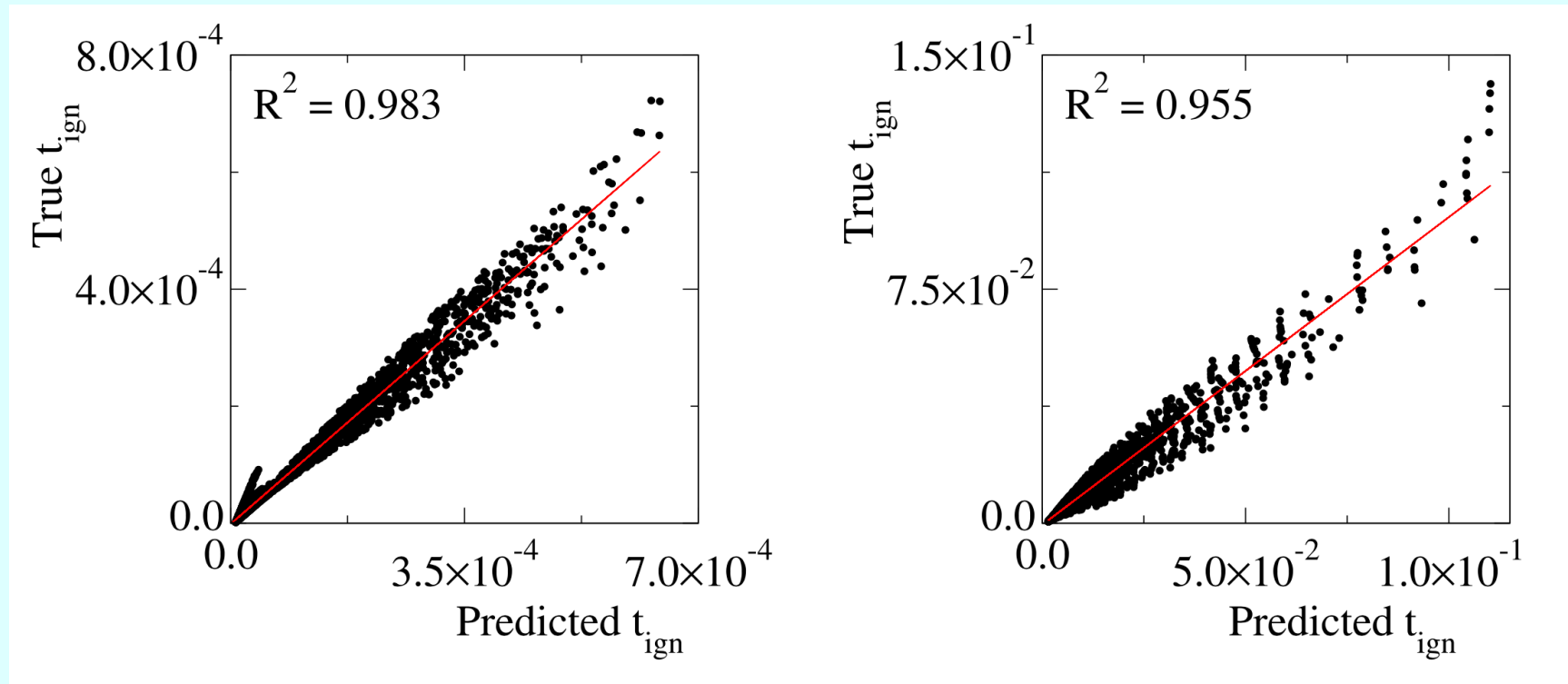
H₂/air



CH₄/air



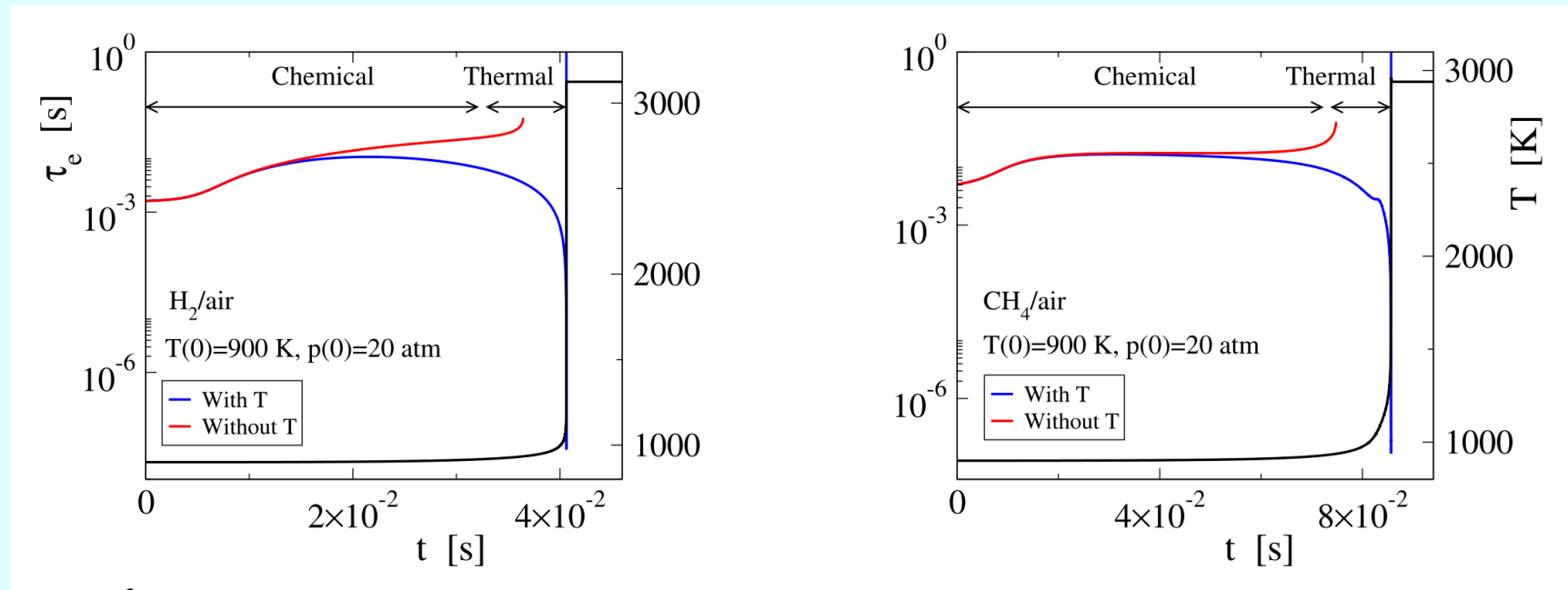
Optimal input in Machine Learning algorithms



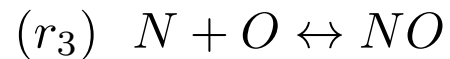
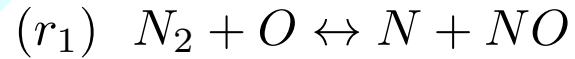
$\tau_e(0)$ input



Optimal input in Machine Learning algorithms

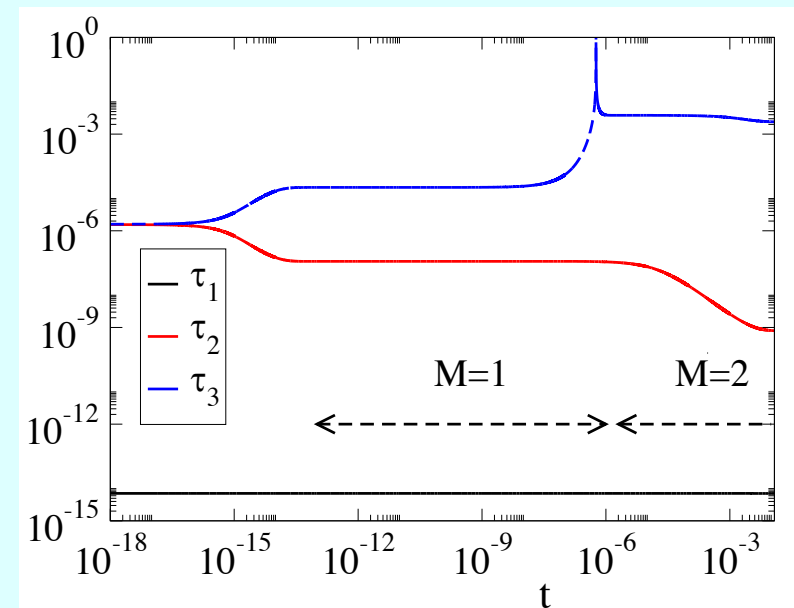
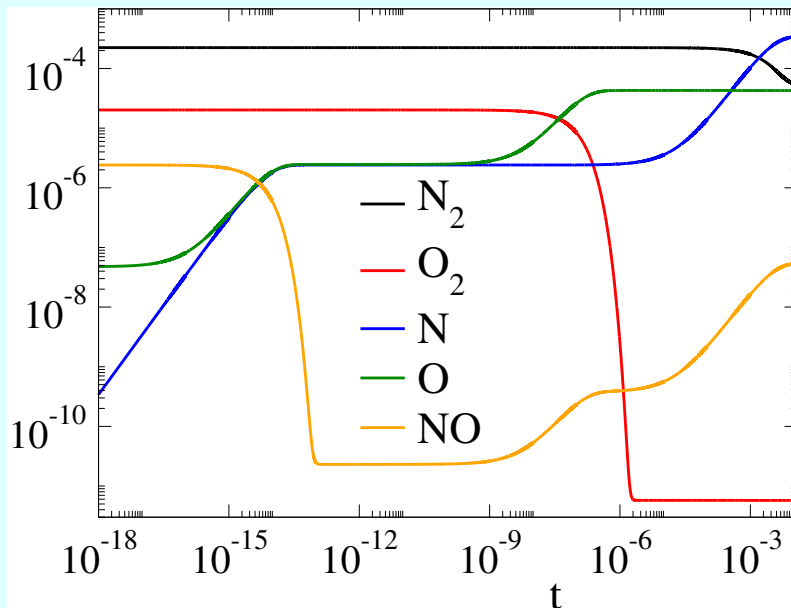


Sensitivity analysis: extended Zeldovich mechanism



Initial conditions:

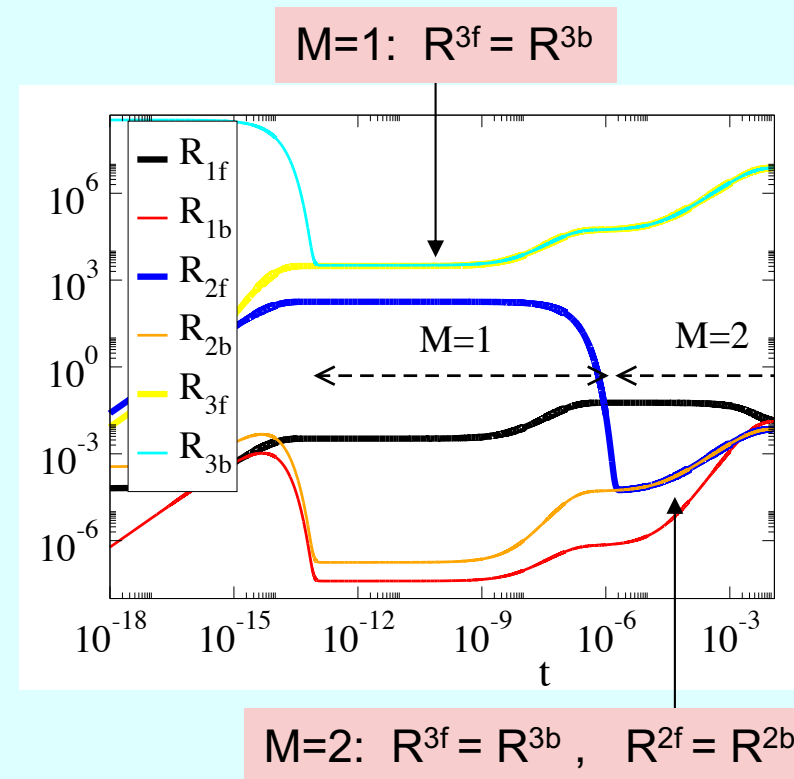
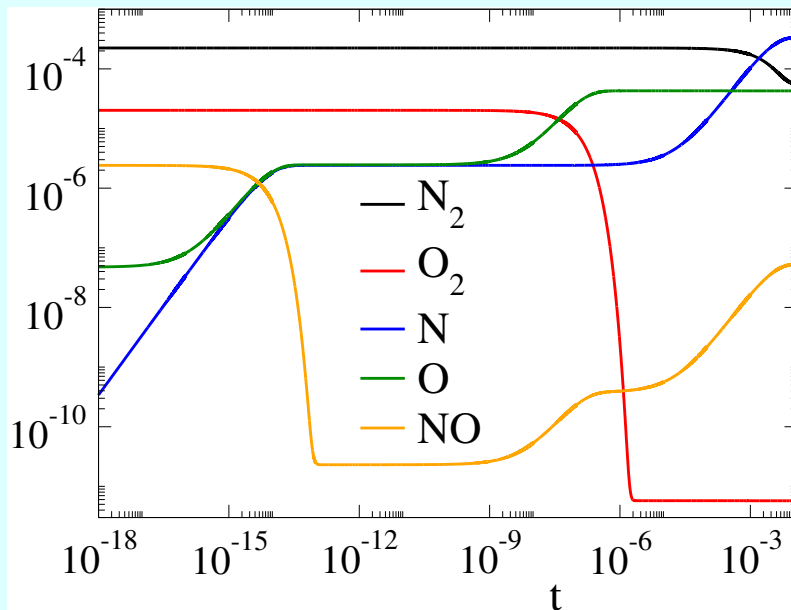
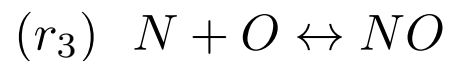
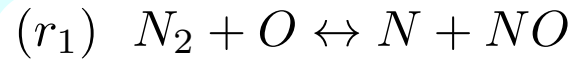
equilibrium of homogeneous, adiabatic constant volume autoignition of a NH_3 /air mixture; $\phi = 0.6$



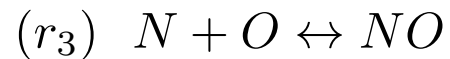
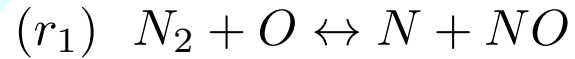
Tingas et al, submitted



Developing equilibria



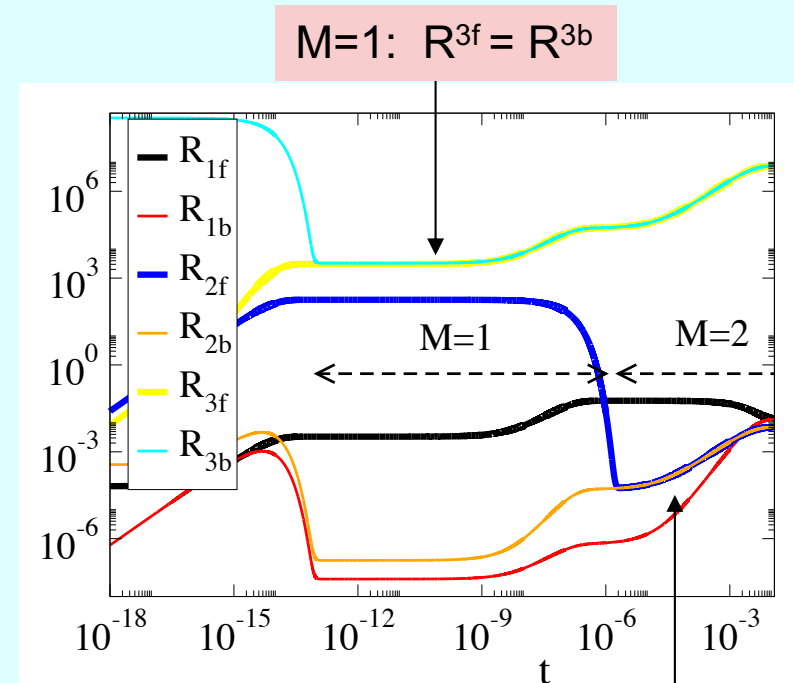
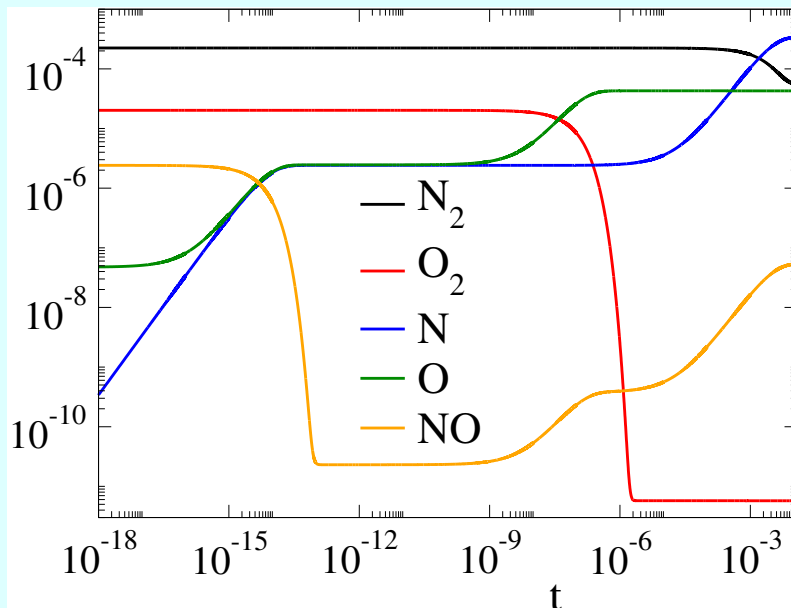
CSP diagnostics



M=1: Fast reaction/species: 3b / NO, $R^{3f} - R^{3b} = 0$

M=2: Fast reaction/species: 3b / NO, $R^{3f} - R^{3b} = 0$

Fast reaction/species: 2f / O₂, $R^{2f} - R^{2b} = 0$

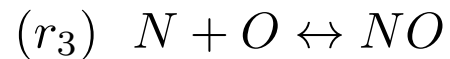
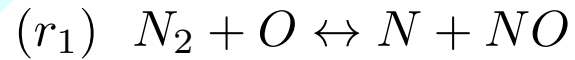


M=1: $R^{3f} = R^{3b}$

M=2: $R^{3f} = R^{3b}$, $R^{2f} = R^{2b}$



Phase space

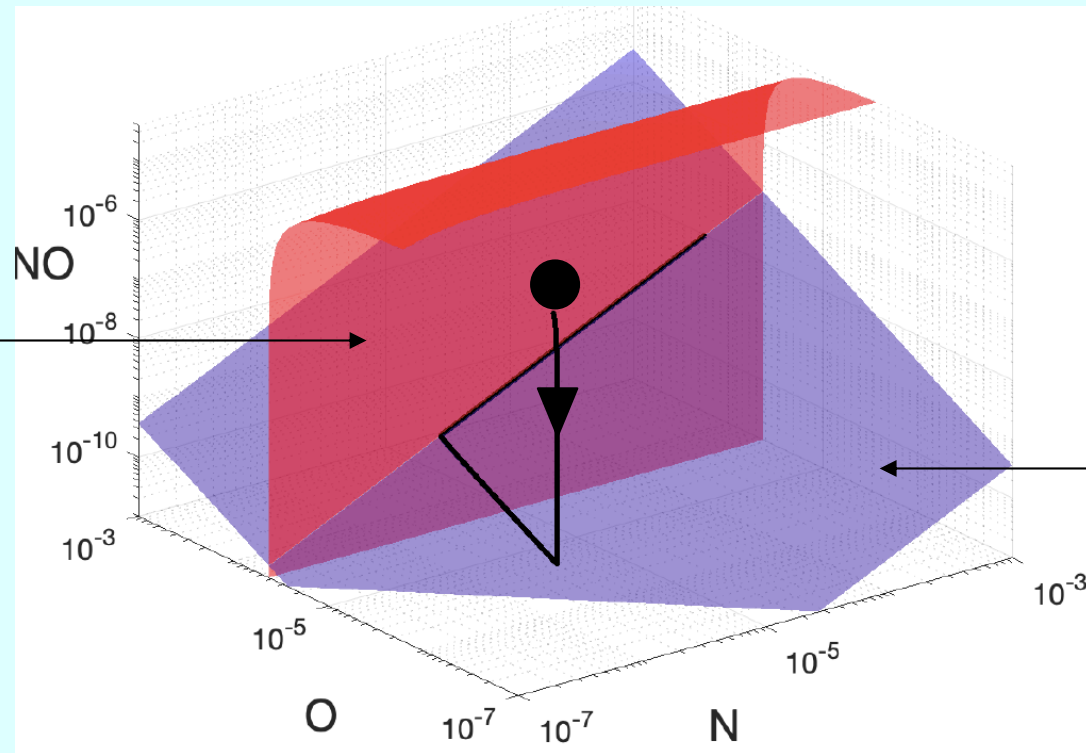


M=1: Fast reaction/species: 3b / NO, $R^{3f} - R^{3b} = 0$

M=2: Fast reaction/species: 3b / NO, $R^{3f} - R^{3b} = 0$

Fast reaction/species: 2f / O₂, $R^{2f} - R^{2b} = 0$

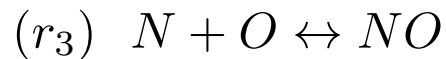
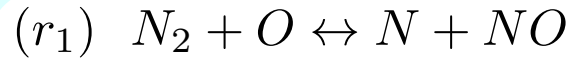
QSSA O₂



QSSA NO



Relations among the sensitivity coefficients



M=1: Fast reaction/species: 3b / NO, $R^{3f} - R^{3b} = 0$

$$k_{3f}[N][O] = k_{3b}[NO]$$

$$k_{3f} \frac{\partial}{\partial k_{3f}}$$

$$k_{3b} \frac{\partial}{\partial k_{3b}}$$

$$\cancel{k_{3f}[N][O]} + k_{3f} \frac{\partial[N]}{\partial \ln k_{3f}} [O] + k_{3f}[N] \frac{\partial[O]}{\partial \ln k_{3f}} = k_{3b} \frac{\partial[NO]}{\partial \ln k_{3f}}$$

$$k_{3f} \frac{\partial[N]}{\partial \ln k_{3b}} [O] + k_{3f}[N] \frac{\partial[O]}{\partial \ln k_{3b}} = k_{3b} \cancel{[NO]} + k_{3b} \frac{\partial[NO]}{\partial \ln k_{3b}}$$

$$k_{3f} \left(\frac{\partial[N]}{\partial \ln k_{3f}} + \frac{\partial[N]}{\partial \ln k_{3b}} \right) [O] + k_{3f}[N] \left(\frac{\partial[O]}{\partial \ln k_{3f}} + \frac{\partial[O]}{\partial \ln k_{3b}} \right) = k_{3b} \left(\frac{\partial[NO]}{\partial \ln k_{3f}} + \frac{\partial[NO]}{\partial \ln k_{3b}} \right)$$



Relations among the sensitivity coefficients

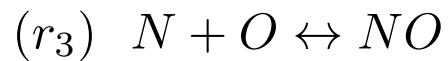
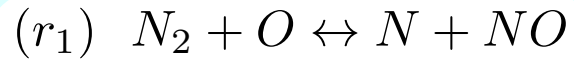
$$k_{3f} \left(\frac{\partial[N]}{\partial \ln k_{3f}} + \frac{\partial[N]}{\partial \ln k_{3b}} \right) [O] + k_{3f}[N] \left(\frac{\partial[O]}{\partial \ln k_{3f}} + \frac{\partial[O]}{\partial \ln k_{3b}} \right) = k_{3b} \left(\frac{\partial[NO]}{\partial \ln k_{3f}} + \frac{\partial[NO]}{\partial \ln k_{3b}} \right)$$

$$k_{3f} \left(\frac{\partial[N]}{\partial \ln k_{1f}} + \frac{\partial[N]}{\partial \ln k_{1b}} + \frac{\partial[N]}{\partial \ln k_{2f}} + \frac{\partial[N]}{\partial \ln k_{2b}} + \frac{\partial[N]}{\partial \ln k_{3f}} + \frac{\partial[N]}{\partial \ln k_{3b}} \right) [O] +$$

$$k_{3f}[N] \left(\frac{\partial[O]}{\partial \ln k_{1f}} + \frac{\partial[O]}{\partial \ln k_{1b}} + \frac{\partial[O]}{\partial \ln k_{2f}} + \frac{\partial[O]}{\partial \ln k_{2b}} + \frac{\partial[O]}{\partial \ln k_{3f}} + \frac{\partial[O]}{\partial \ln k_{3b}} \right) =$$

$$k_{3b} \left(\frac{\partial[NO]}{\partial \ln k_{1f}} + \frac{\partial[NO]}{\partial \ln k_{1b}} + \frac{\partial[NO]}{\partial \ln k_{2f}} + \frac{\partial[NO]}{\partial \ln k_{2b}} + \frac{\partial[NO]}{\partial \ln k_{3f}} + \frac{\partial[NO]}{\partial \ln k_{3b}} \right)$$

Relations among the sensitivity coefficients



$$k_{3f}[N][O] = k_{3b}[NO]$$

$$(k_{3f}[O])\delta[N] + (k_{3f}[N])\delta[O] - (k_{3b})\delta[NO] = 0$$

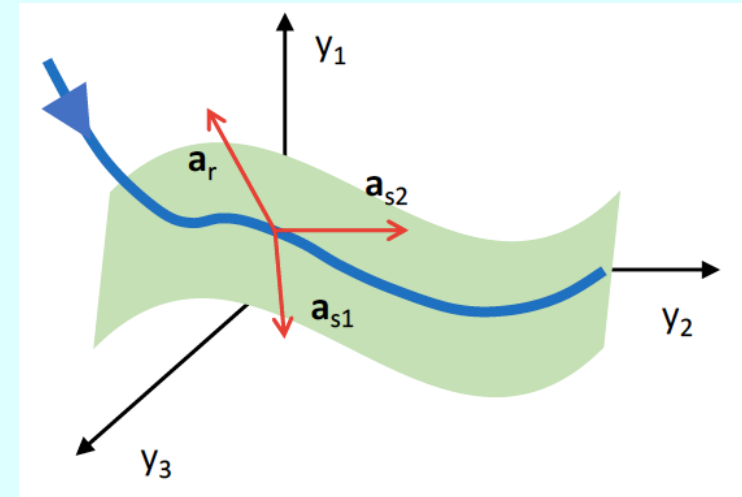
$$\delta = \left(\frac{\partial}{\partial \ln k_{1f}} + \frac{\partial}{\partial \ln k_{1b}} + \frac{\partial}{\partial \ln k_{2f}} + \frac{\partial}{\partial \ln k_{2b}} + \frac{\partial}{\partial \ln k_{3f}} + \frac{\partial}{\partial \ln k_{3b}} \right)$$

$$\underbrace{[0, 0, k_{3f}[O], k_{3f}[N], -k_{3b}]}_{\text{row vector}} \begin{bmatrix} \delta[N_2] \\ \delta[O_2] \\ \delta[N] \\ \delta[O] \\ \delta[NO] \end{bmatrix} = 0$$



Relations among the sensitivity coefficients

$$\mathbf{y} = \begin{bmatrix} sp_1 \\ sp_2 \\ sp_3 \\ \vdots \end{bmatrix} \quad \frac{d\mathbf{y}}{dt} = \mathbf{g}$$



$$\mathbf{G} = \begin{bmatrix} \tilde{\partial}[sp_1] \\ \tilde{\partial}[sp_2] \\ \tilde{\partial}[sp_3] \\ \vdots \end{bmatrix} \quad \frac{d\mathbf{G}}{dt} = \mathbf{g} + \mathbf{JG}$$

$$\tilde{\partial} = \left(\frac{\partial}{\partial \ln k_{1f}} + \frac{\partial}{\partial \ln k_{1b}} + \frac{\partial}{\partial \ln k_{2f}} + \dots \right)$$

$$\mathbf{J} = \text{grad}(\mathbf{g})$$

$$\mathbf{Q} = \mathbf{a}_{s1} \mathbf{b}^{s1} + \mathbf{a}_{s2} \mathbf{b}^{s2}$$

$$\mathbf{b}^i \cdot \mathbf{a}_j = \delta_j^i$$

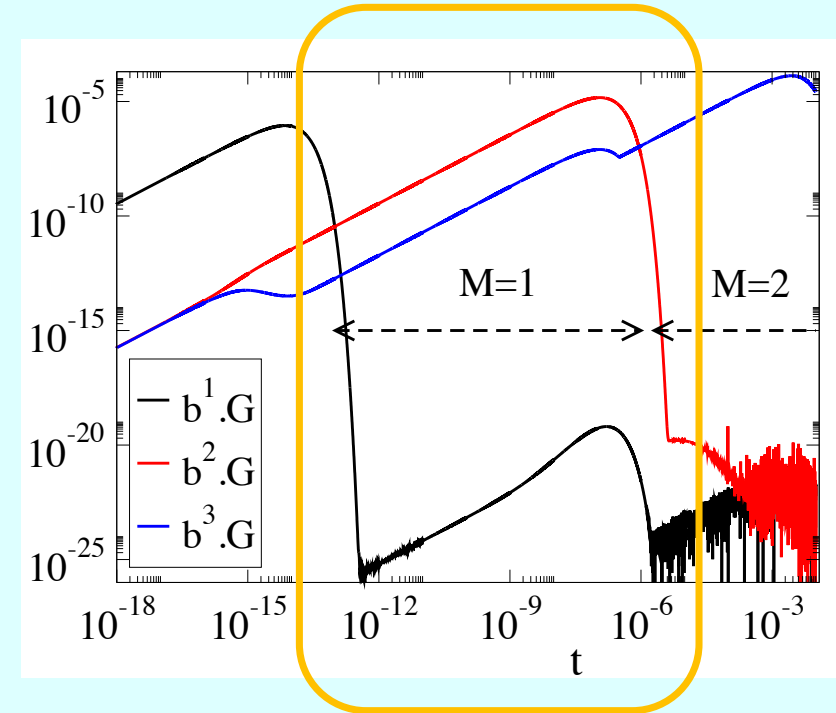
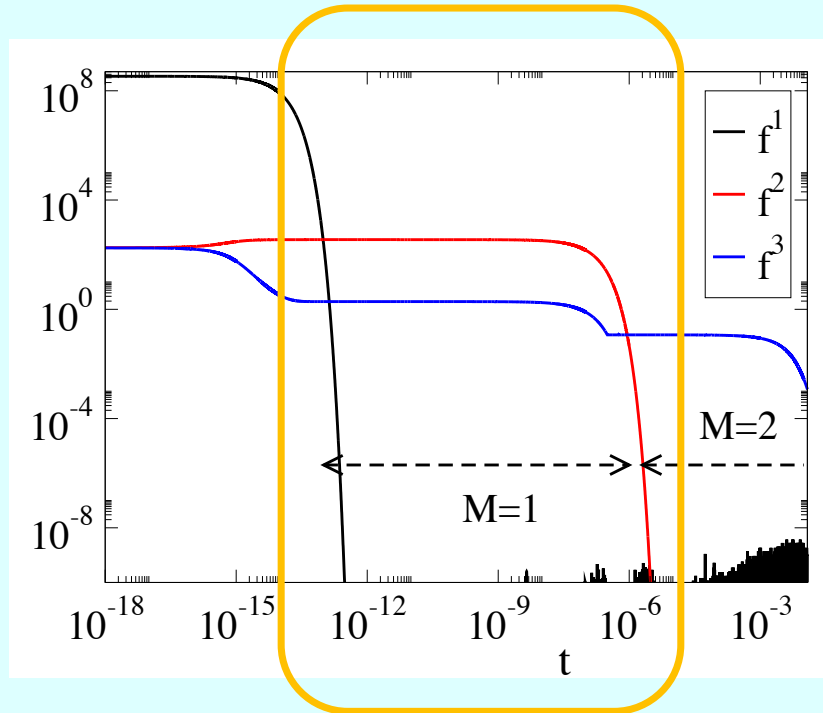
$$\frac{d\mathbf{y}}{dt} = \mathbf{Qg}$$

$$\mathbf{b}^r \cdot \mathbf{g} = 0$$

$$\frac{d\mathbf{G}}{dt} = \mathbf{Q}(\mathbf{g} + \mathbf{JG})$$

$$\mathbf{b}^r \cdot \mathbf{G} = 0$$

Relations among the sensitivity coefficients



$M = 1$

$$f^1 = b^1 \cdot g = b^r \cdot g \approx 0$$

$$f^2 = b^2 \cdot g = b^{s1} \cdot g$$

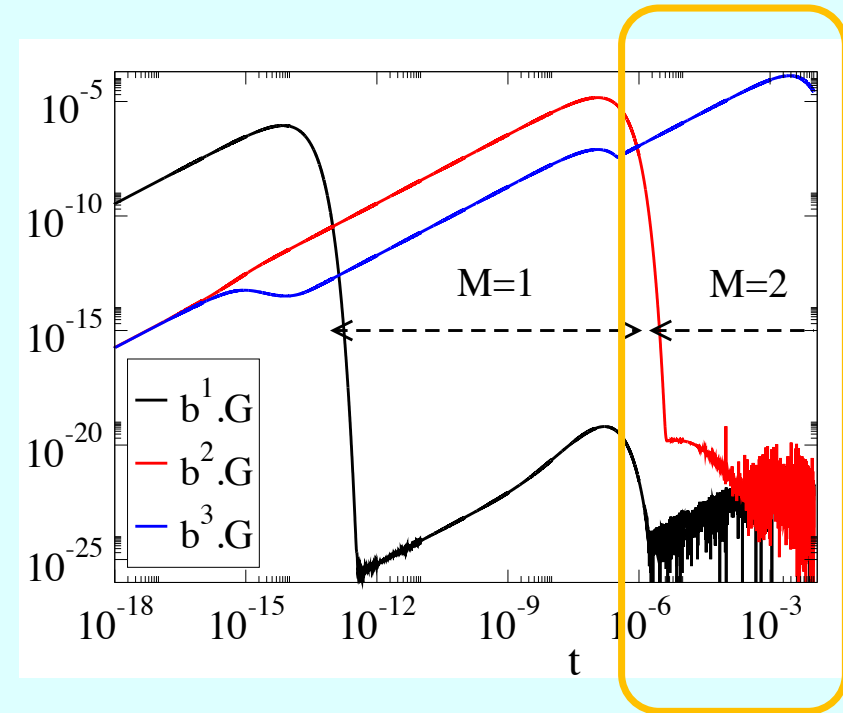
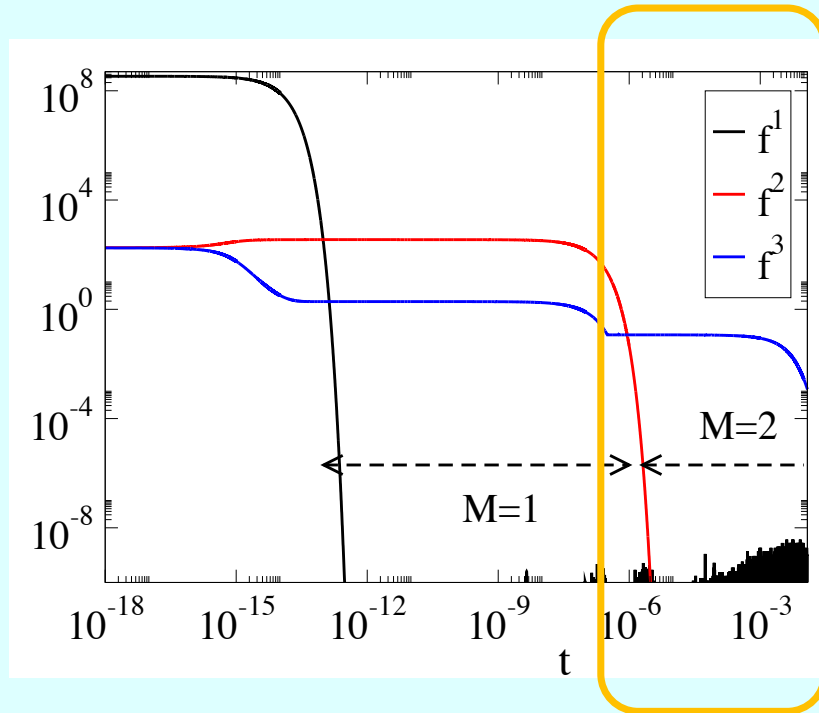
$$f^3 = b^3 \cdot g = b^{s2} \cdot g$$

$$b^1 \cdot G = b^r \cdot G \approx 0$$

$$b^2 \cdot G = b^{s1} \cdot G$$

$$b^3 \cdot G = b^{s2} \cdot G$$

Relations among the sensitivity coefficients



$M = 2$

$$f^1 = b^1 \cdot g = b^{r1} \cdot g \approx 0$$

$$f^2 = b^2 \cdot g = b^{r2} \cdot g \approx 0$$

$$f^3 = b^3 \cdot g = b^s \cdot g$$

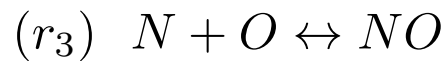
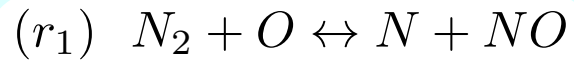
$$b^1 \cdot G = b^{r1} \cdot G \approx 0$$

$$b^2 \cdot G = b^{r2} \cdot G \approx 0$$

$$b^3 \cdot G = b^s \cdot G$$



M=1: Physical understanding



CSP Fast reaction: 3b

Fast species: NO

Equilibrium: $QSSA_{NO} \quad R^{3f} - R^{3b} = 0$

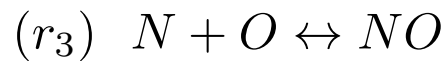
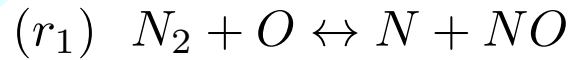
LSA Fast variable: $\delta[NO]$

Equilibrium: $\frac{\partial[NO]}{\partial \ln k_{3f}} + \frac{\partial[NO]}{\partial \ln k_{3b}} = 0$

Fast species
Fast reaction



M=2: Physical understanding



CSP Fast reaction: 2f

Fast species: O_2 ,

Equilibrium: QSSA O_2 , $R^{2f} - R^{2b} = 0$

LSA Fast variable: $\delta[O_2]$

Equilibrium:
$$\frac{\partial[O_2]}{\partial \ln k_{2f}} + \frac{\partial[O_2]}{\partial \ln k_{2b}} = 0$$

Fast species
Fast reaction





Appendix A

Lindemann Mechanism



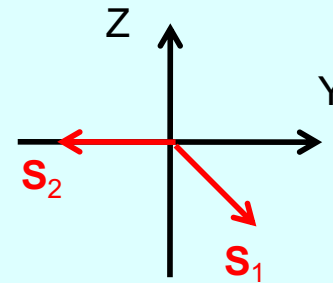
The traditional method (asymptotics): Lindemann mechanism



$$\frac{d}{dt} \begin{bmatrix} Y \\ Z \end{bmatrix} = \begin{bmatrix} +1 \\ -1 \end{bmatrix} k_{1f} Z^2 + \begin{bmatrix} -1 \\ +1 \end{bmatrix} k_{1b} YZ + \begin{bmatrix} -1 \\ 0 \end{bmatrix} k_2 Y$$

$[B] = Y$
 $[A] = Z$

$$\frac{dy}{dt} = S_1 (R^{1f} - R^{1b}) + S_2 R^2$$



First steps in asymptotic analysis: 1) get the system in **non-dimensional form**
 2) define the **small parameter ε**



Three different non-dimensional systems

1st system

$$\frac{dy}{dt} = \frac{z^2}{\varepsilon} - \frac{yz}{\varepsilon} - y$$

$$\frac{dz}{dt} = -r \left(\frac{z^2}{\varepsilon} - \frac{yz}{\varepsilon} \right)$$

$$\varepsilon = \frac{k_2}{k_{1b}A_o} \ll 1$$

2nd system

$$\frac{dy}{dt} = z^2 - yz - y$$

$$\frac{dz}{dt} = -r(z^2 - yz)$$

$$\varepsilon = \frac{k_2}{k_{1b}A_o} = O(1)$$

3rd system

$$\frac{dy}{dt} = \frac{z^2}{\varepsilon} - yz - \frac{y}{\varepsilon}$$

$$\frac{dz}{dt} = -r(z^2 - \varepsilon yz)$$

$$\varepsilon = \frac{k_{1b}A_o}{k_2} \ll 1$$

$$r = \frac{k_{1f}}{k_{1b}}$$

$$[A] = A_o z$$



Three non-dimensional systems

1st system

$$\frac{dy}{dt} = \frac{z^2}{\varepsilon} - \frac{yz}{\varepsilon} - y$$

$$\frac{dz}{dt} = -r \left(\frac{z^2}{\varepsilon} - \frac{yz}{\varepsilon} \right)$$

PEA-1

$$r=O(1), \quad \varepsilon \ll 1$$

2nd system

$$\frac{dy}{dt} = z^2 - yz - y$$

$$\frac{dz}{dt} = -r(z^2 - yz)$$

QSSA-z

$$r \gg 1, \quad \varepsilon = O(1)$$

3rd system

$$\frac{dy}{dt} = \frac{z^2}{\varepsilon} - yz - \frac{y}{\varepsilon}$$

$$\frac{dz}{dt} = -r(z^2 - \varepsilon yz)$$

QSSA-y

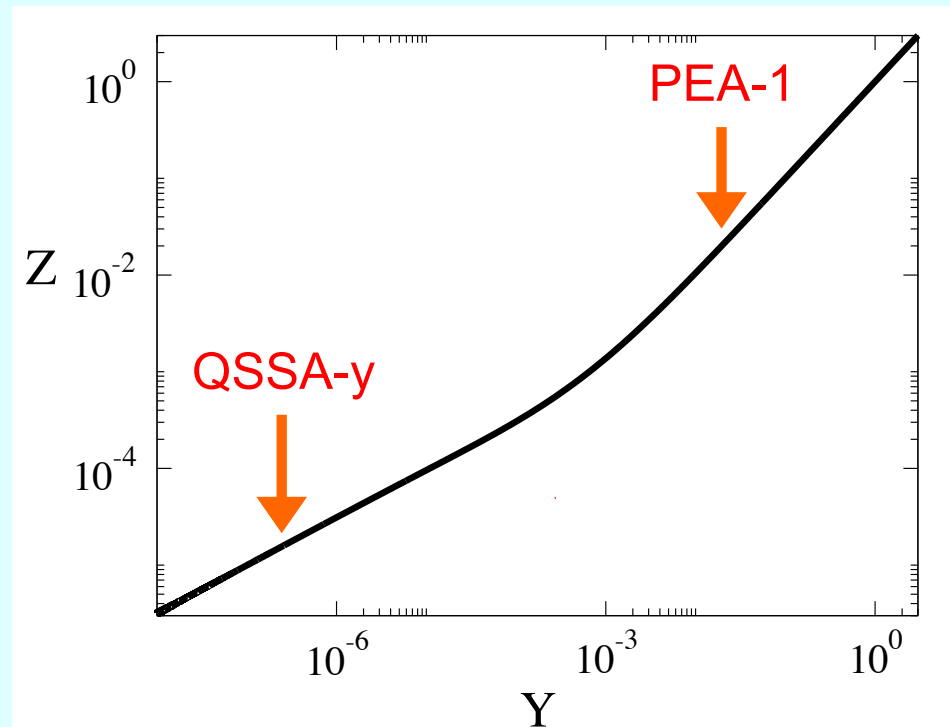
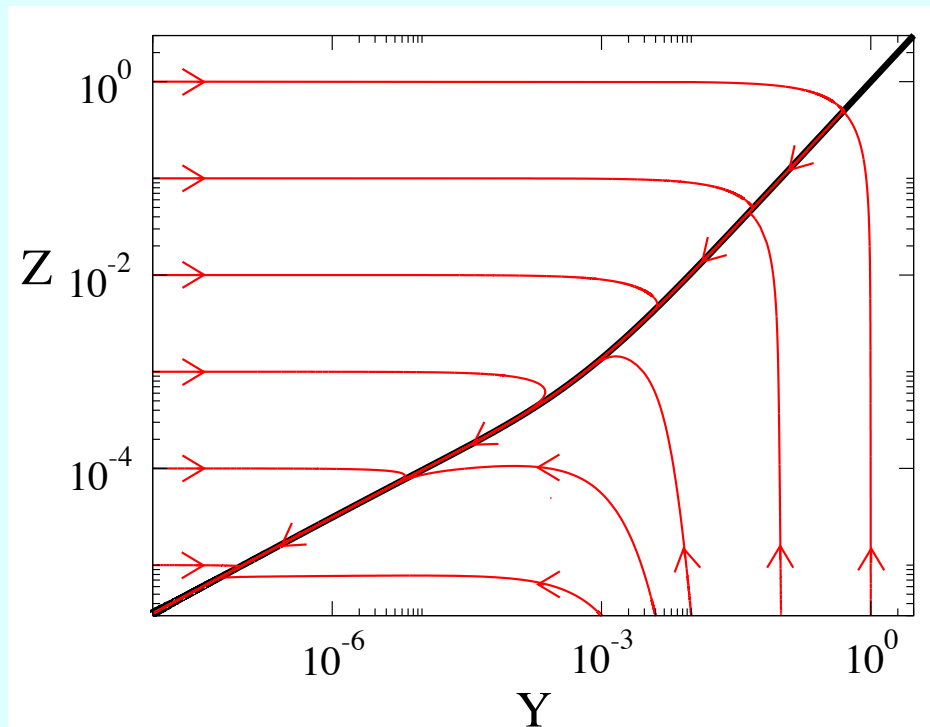
$$r=O(1), \quad \varepsilon \ll 1$$

⋮



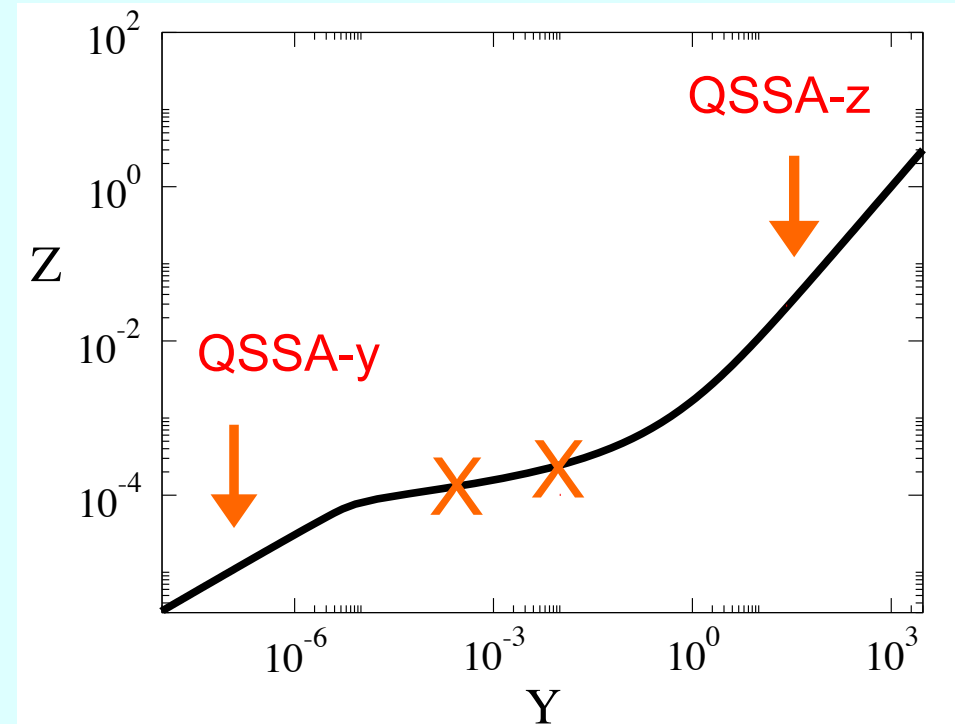
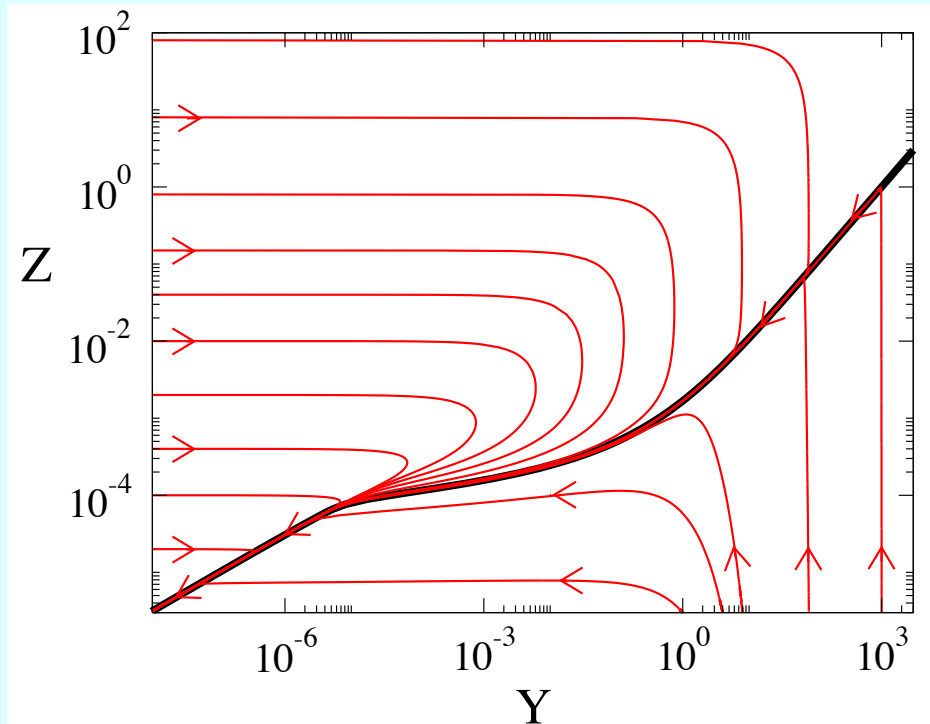
Trajectories: $k_{1f}=10^3$ $k_{1b}=10^3$ $k_2=1$

Leading order asymptotics



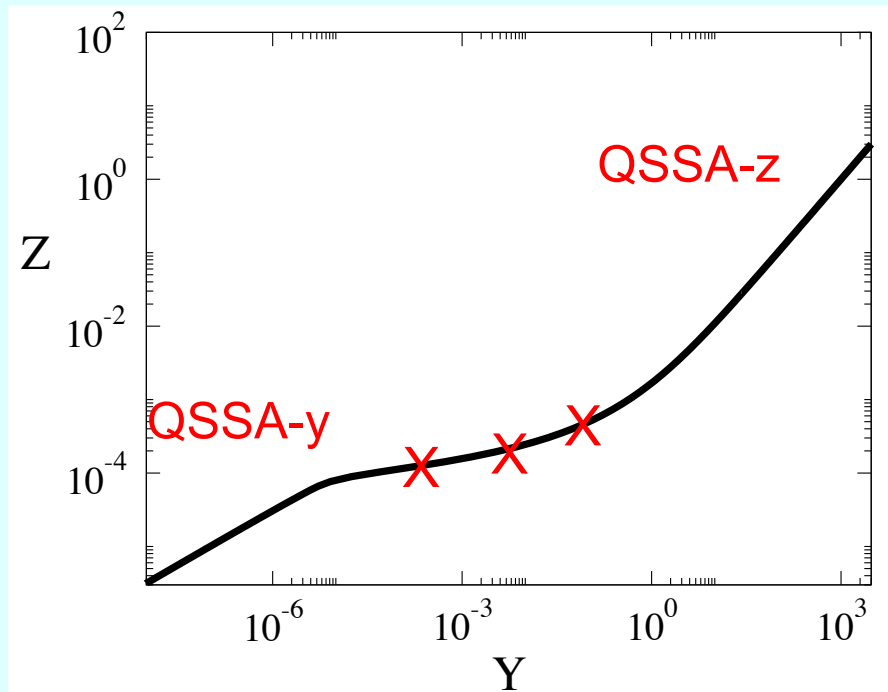
Trajectories: $k_{1f}=10^3$ $k_{1b}=1$ $k_2=1$

Leading order asymptotics

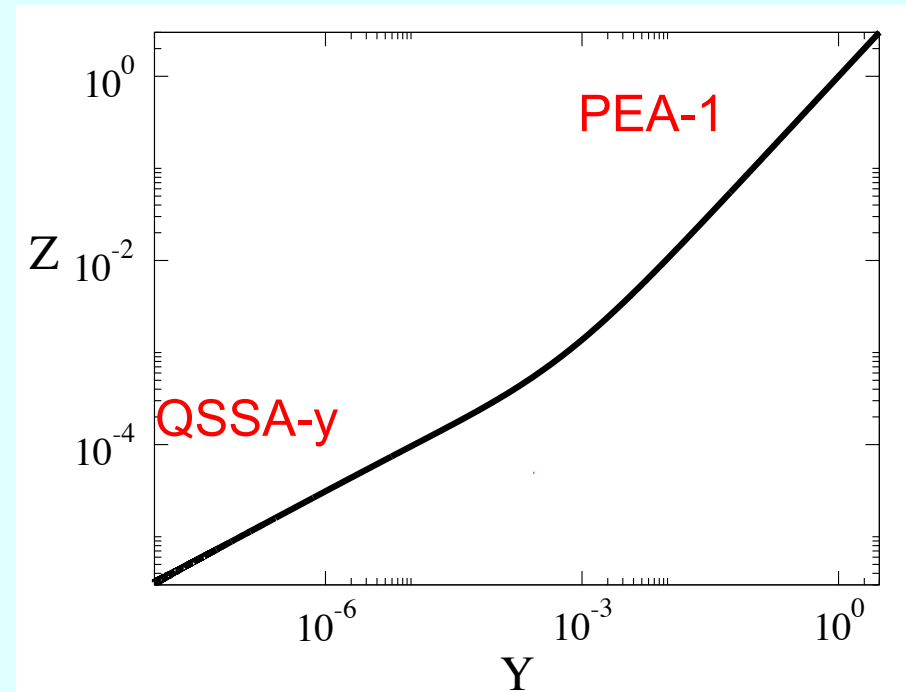


Long term (global) behavior: 3 different approximations

$$k_{1f}=10^3 \quad k_{1b}=1 \quad k_2=1$$

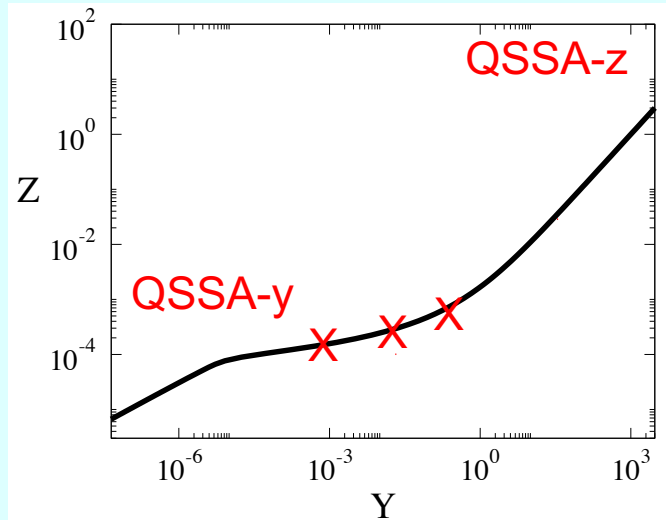


$$k_{1f}=10^3 \quad k_{1b}=10^3 \quad k_2=1$$

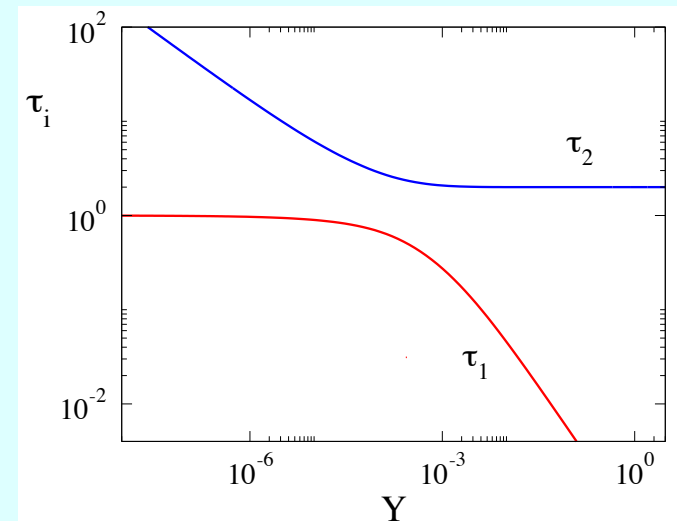
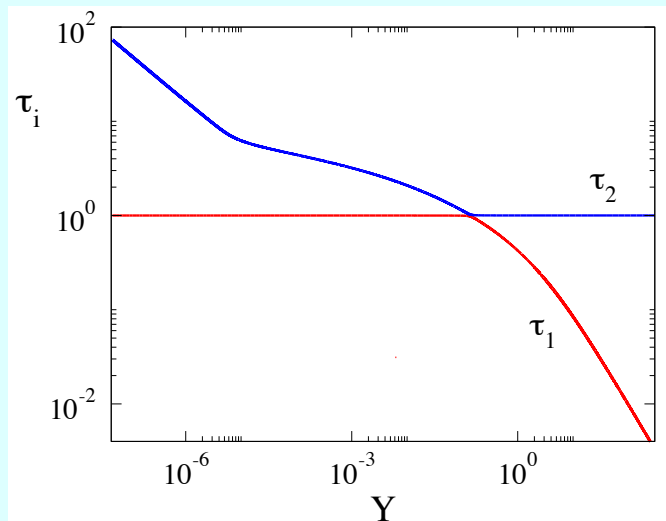
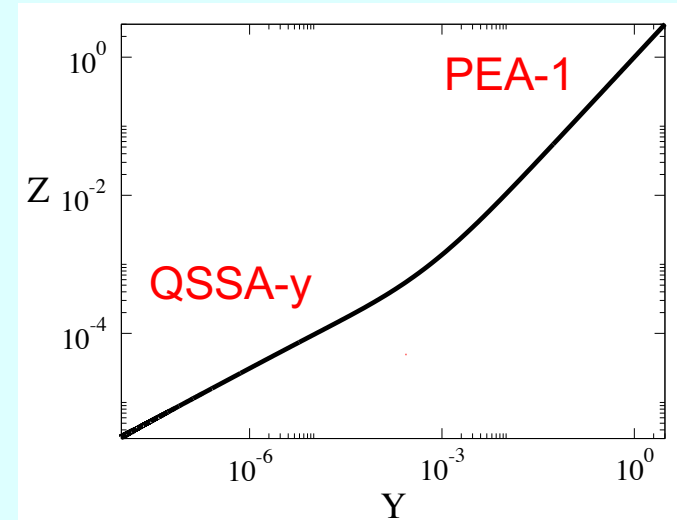


Long term (global) behavior: 3 different approximations

$$k_{1f}=10^3 \quad k_{1b}=1 \quad k_2=1$$

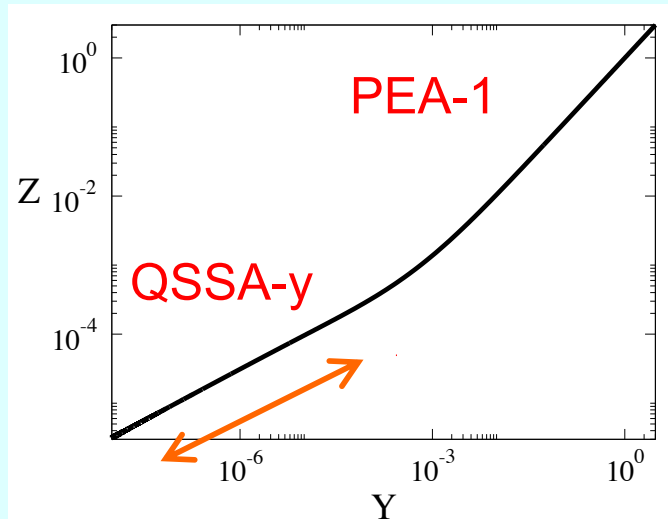


$$k_{1f}=10^3 \quad k_{1b}=10^3 \quad k_2=1$$



Long term behavior: QSSA - y

$$k_{1f}=10^3 \quad k_{1b}=10^3 \quad k_2=1$$



Normal form

$$\varepsilon \frac{dy}{dt} = z^2 - yz - \varepsilon y$$

$$y = y_0 + \varepsilon y_1 + \dots$$

$$\frac{dz}{dt} = -z^2 + \varepsilon yz$$

$$z = z_0 + \varepsilon z_1 + \dots$$

$$y_0 = z_0^2$$

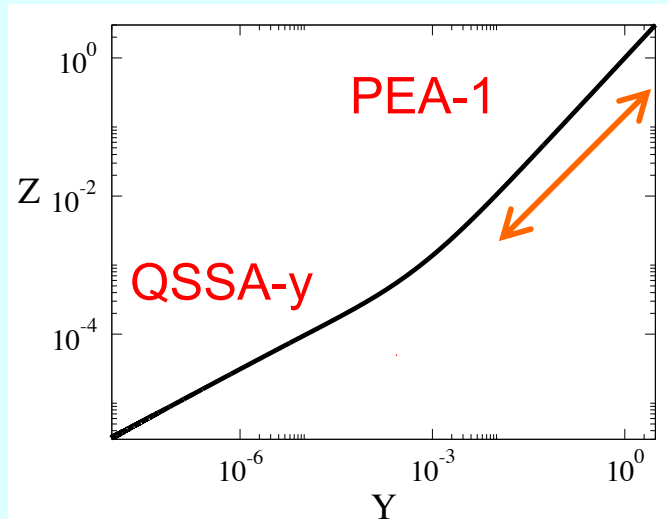
$$\frac{dz_0}{dt} = -z_0^2$$

Fine !



Long term behavior: PEA-1

$$k_{1f}=10^3 \quad k_{1b}=10^3 \quad k_2=1$$



Not in normal form

$$\varepsilon \frac{dy}{dt} = z^2 - yz - \varepsilon y$$

$$y = y_0 + \varepsilon y_1 + \dots$$

$$\varepsilon \frac{dz}{dt} = -z^2 + yz$$

$$z = z_0 + \varepsilon z_1 + \dots$$

$$y_0 = z_0$$

$$y_0 = z_0$$

Problem !



Getting the non-dimensional system in normal form

$$\varepsilon \frac{dy}{dt} = z^2 - yz - \varepsilon y$$

$$\varepsilon \frac{dz}{dt} = -z^2 + yz$$

$$w = y + z$$

$$\varepsilon \frac{dy}{dt} = (w - y)(w - 2y) - \varepsilon y$$

$$\frac{dw}{dt} = -y$$

$$y = y_0 + \varepsilon y_1 + \dots$$

$$w = w_0 + \varepsilon w_1 + \dots$$

$$w_0 = 2y_0$$

$$\frac{dw_0}{dt} = -y_0$$

$$z = w - y$$

$$y_0 = z_0$$

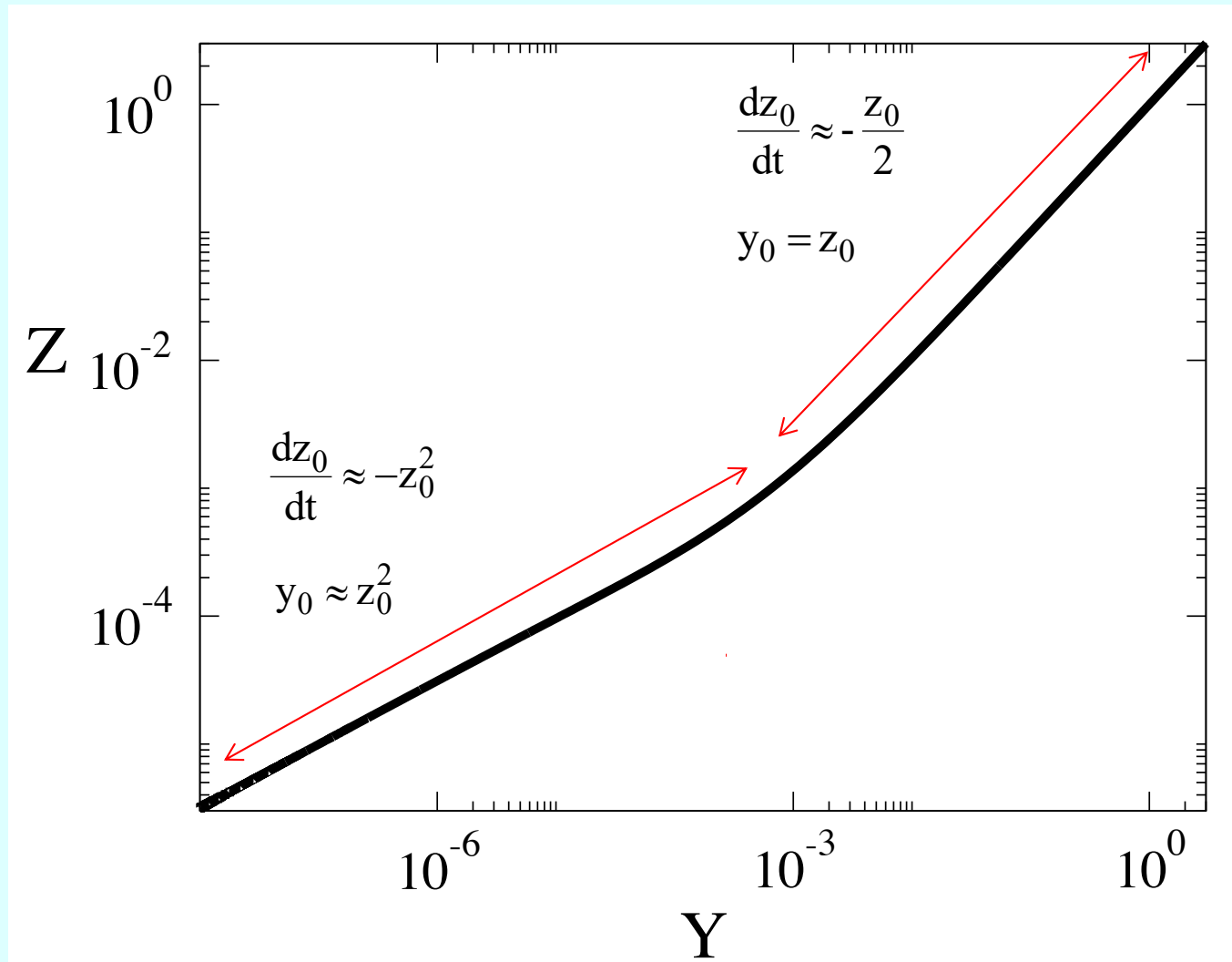
$$\frac{dz_0}{dt} = -\frac{y_0}{2}$$

Fine !



Long term (global) behavior: 3 different models

$$k_{1f}=10^3 \quad k_{1b}=10^3 \quad k_2=1$$



Long term (global) behavior: 3 different models

$$k_{1f}=10^3 \quad k_{1b}=10^3 \quad k_2=1$$

